

# Quantification of carbon nano materials in complex matrices

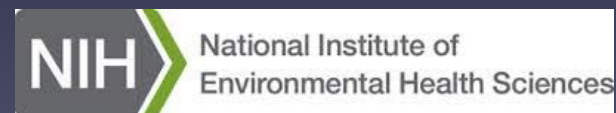
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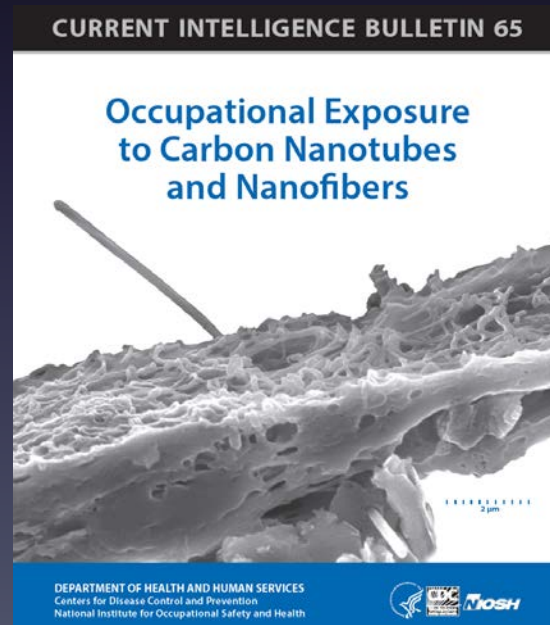


Nano-Go Funding: RES018801Z



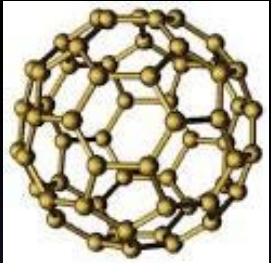
# Outline

- 1D to 3D Types of Carbon
- Range of analytical techniques & response ranges
- Standards & Analytics
- Extraction / Separation Strategies
- Paths forward for Exposure monitoring



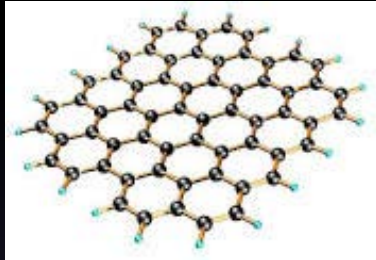
# Types of Carbon Nanomaterials

$C_{60, 70, \text{etc}}$

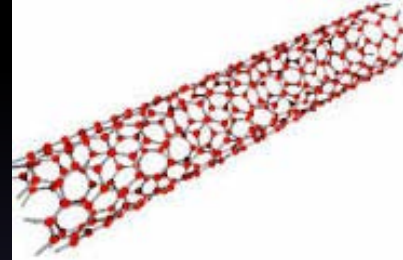


$n C_{60}$

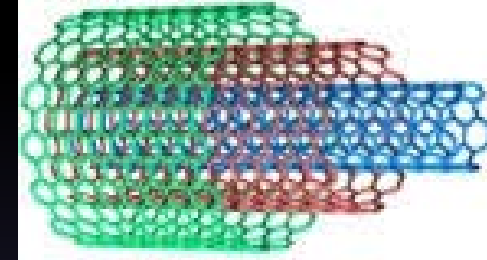
Few Layer Graphene



Single Wall CNT



Multi- Wall CNT



3 - D



2 - D



1 - D

720 Dalton

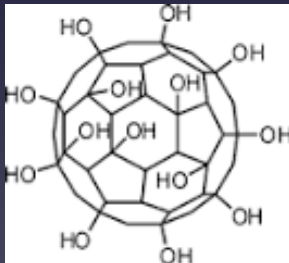


$10 \text{ \AA} (1\text{nm}) \times \mu\text{m}^2$

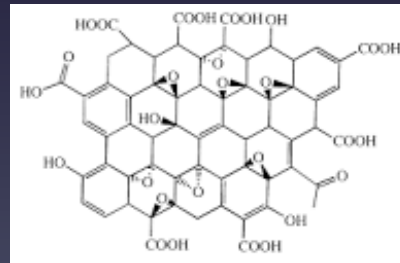


1-10 nm by  $\mu\text{m}$ -mm

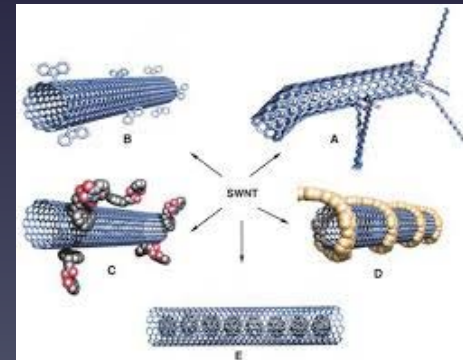
Wide Ranges of surface Functionalities can occur



Fullerol  
(Hydroxy-fullerene)



Graphene-Oxide



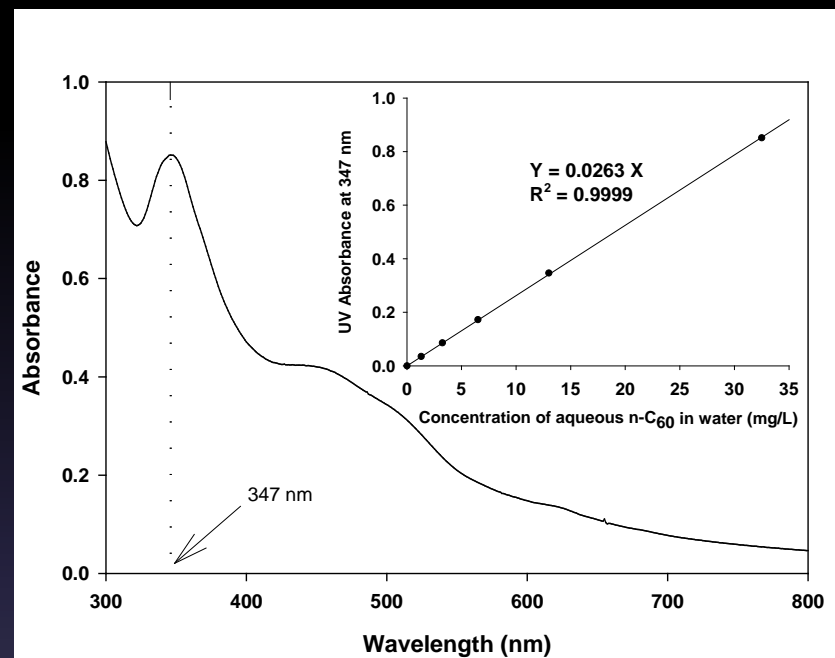
# Quantification Techniques

Technique	C60 & nC60	FLG & GO	SW- & MW-CNT
Light scattering		✓	✓
Absorbance			
- UV	✓		
- NIR fluorescence			✓
- Gel electrophoresis			✓
HPLC-UV (FFF-UV)	✓		
LC-Mass Spec	✓		
Thermal			
- Combustion / CO/CH4	✓	✓	✓
- Microwave induced heat			✓
- TGA mass loss		✓	✓
<sup>14</sup> C			
Scintillation	✓	✓	✓
Thermal-Mass spec	✓		✓
Raman spec.		✓	✓
Photo-acoustic/thermal			✓
Single particle ICP-MS of catalyst			✓

# Examples – Fullerenes (C<sub>60</sub>)

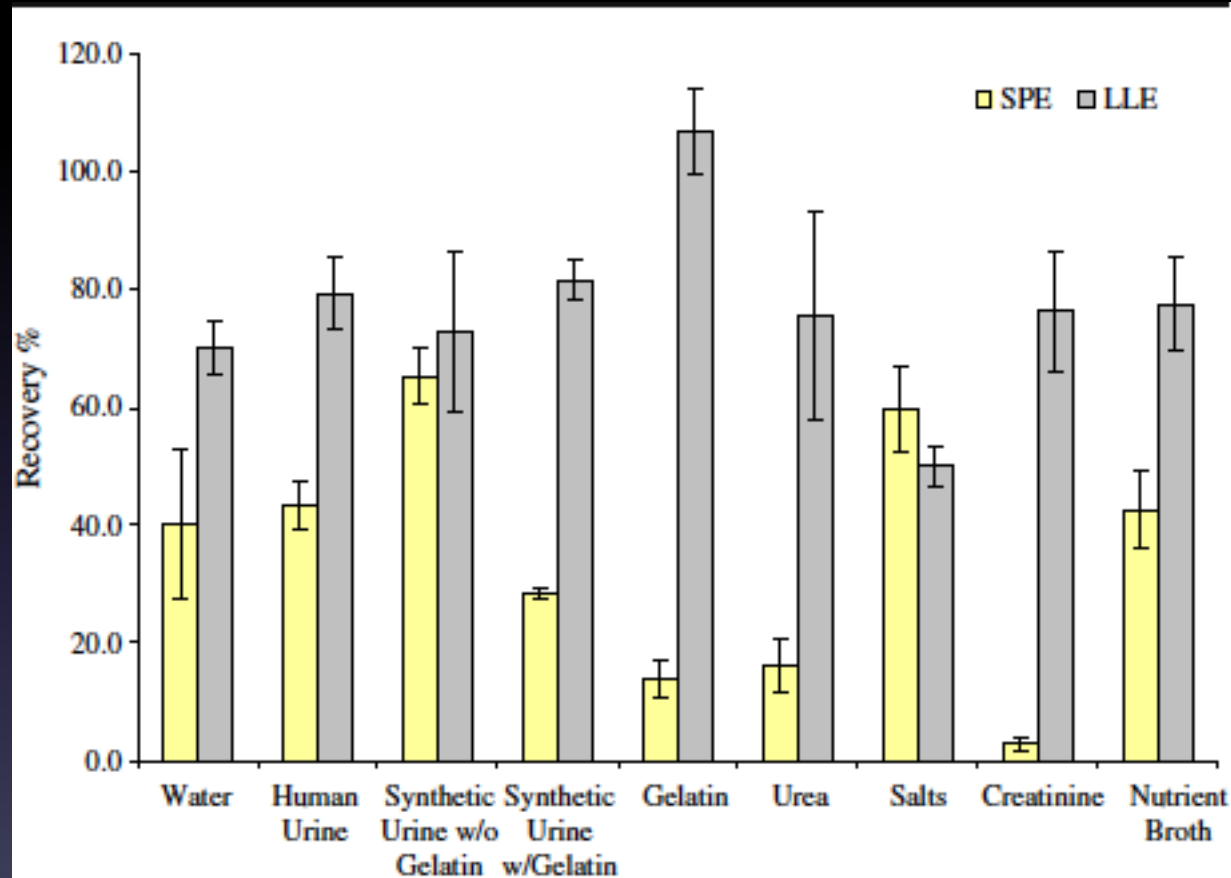


- Light scattering ( $\lambda_{347 \text{ nm}}$ )
- Liquid phase combustion (TOC)
- Thermal optical transmittance
- HPLC plus  $\lambda_{347 \text{ nm}}$
- LC/MS using 720 m/z



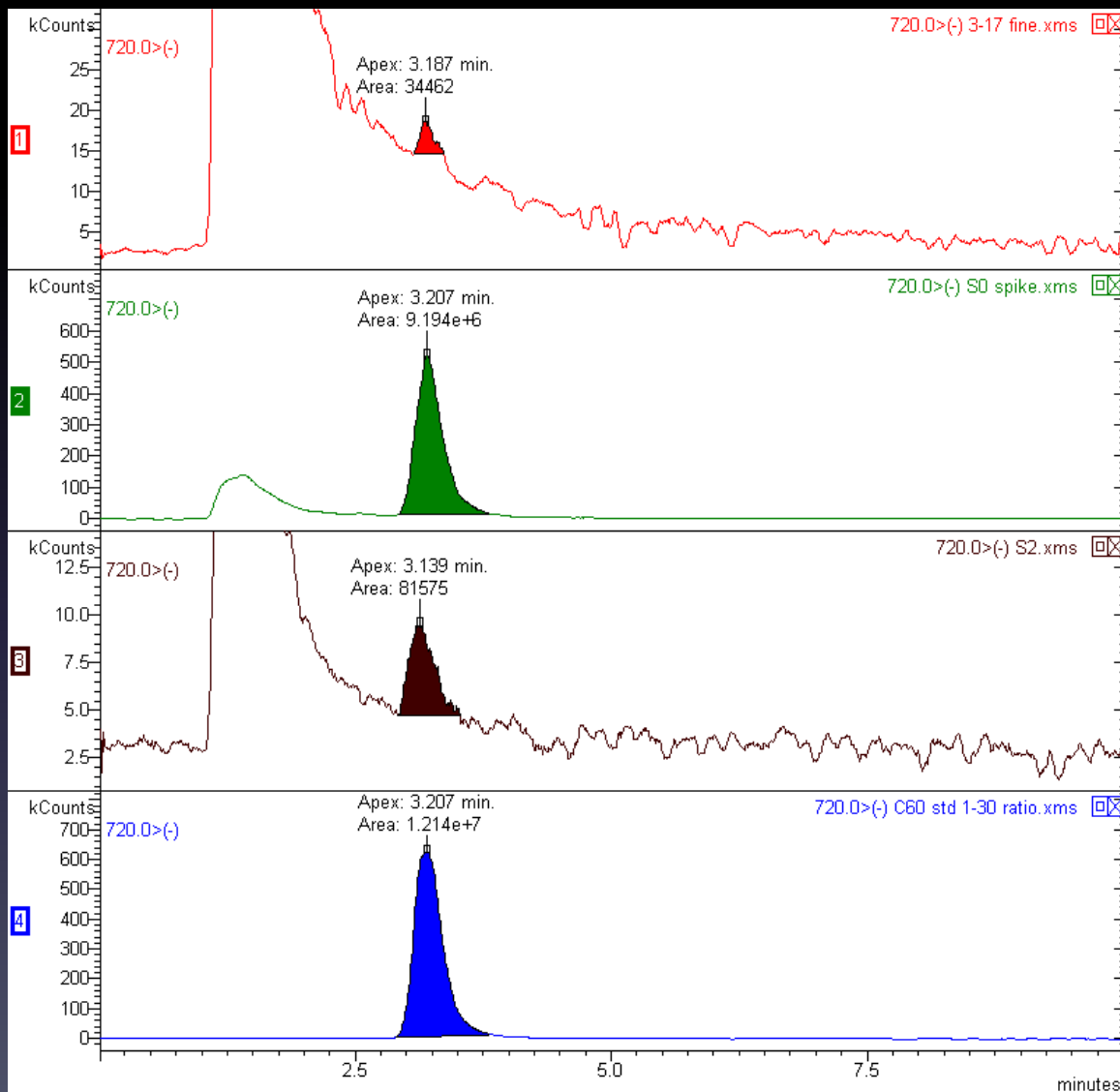
*Improving specificity and lower detection limits*

# C<sub>60</sub> analysis by HPLC-UV or –MS Requires Solvent Extraction



**Fig. 3** Recovery of C<sub>60</sub> from synthetic and human urine matrices using LLE and SPE. C<sub>60</sub> was spiked to a final concentration of 180 µg/L and allowed to equilibrate in the media overnight. *Error bars* indicate the variability in quantification between three extractions of each sample matrix

# Application of LC/MS



C<sub>60</sub> detection in

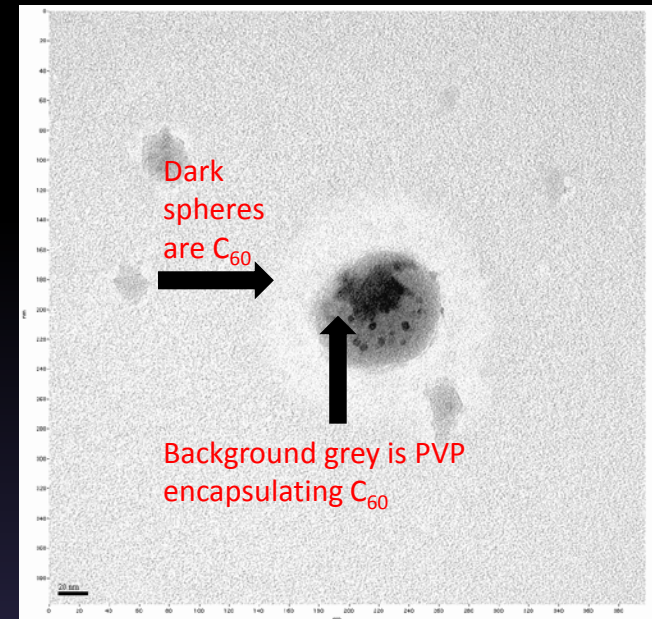
a) a parking garage air sample of PM<sub>2.5</sub>,

b) a filtered air sample spiked with C<sub>60</sub>,

c) a parking garage air sample of size > PM<sub>2.5</sub>,

d) C<sub>60</sub> standard; Detection limit ~500 ng/L

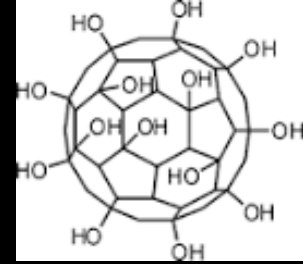
# Fullerenes From Cosmetics



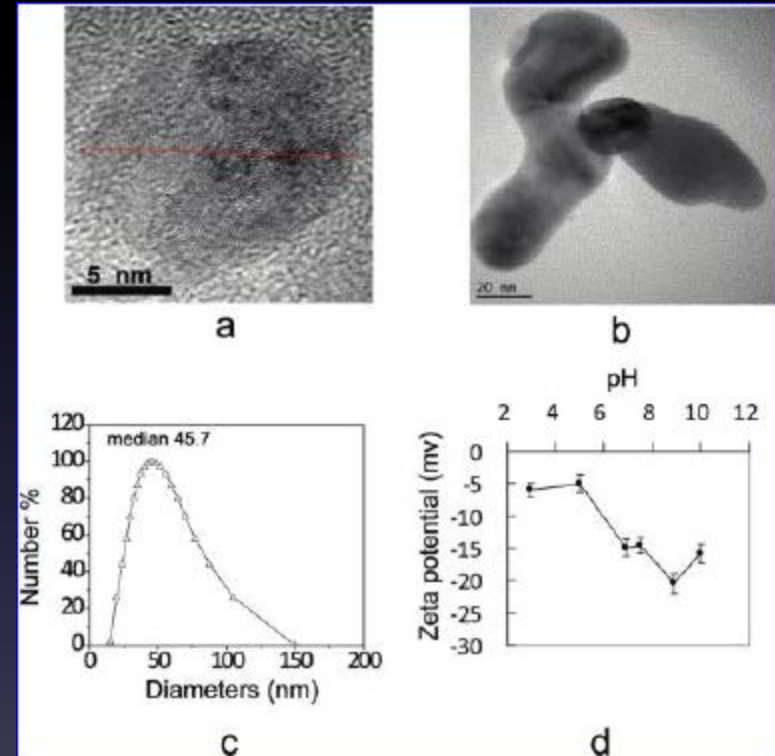
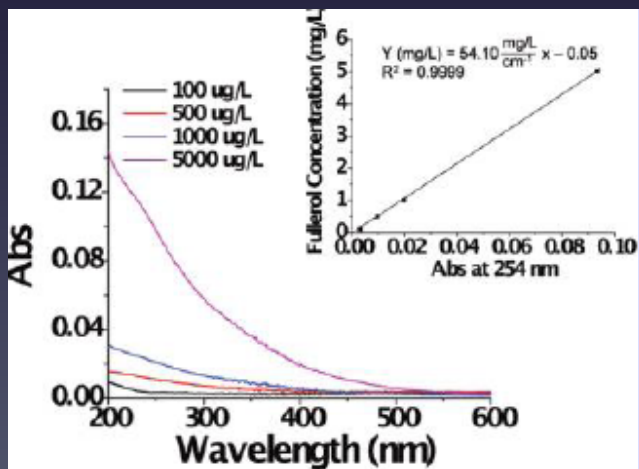
- A common cosmetic formulation disperses fullerenes using polyvinylpyrrolidone (C<sub>60</sub>-PVP)- see TEM
- LC/SM was used to separate and specifically detect fullerenes (C<sub>60</sub> and C<sub>70</sub>) from interfering substances typically present in cosmetics (e.g., castor oil).
- C<sub>60</sub> was detected in 4/5 commercial cosmetics ranging from 0.04 to 1.1 µg/g, and C<sub>70</sub> was qualitatively detected in 2/5 samples.
- A single-use quantity of cosmetic (0.5 g) may contain up to 0.6 µg of C<sub>60</sub> and demonstrates a pathway for human exposure to engineered fullerenes.



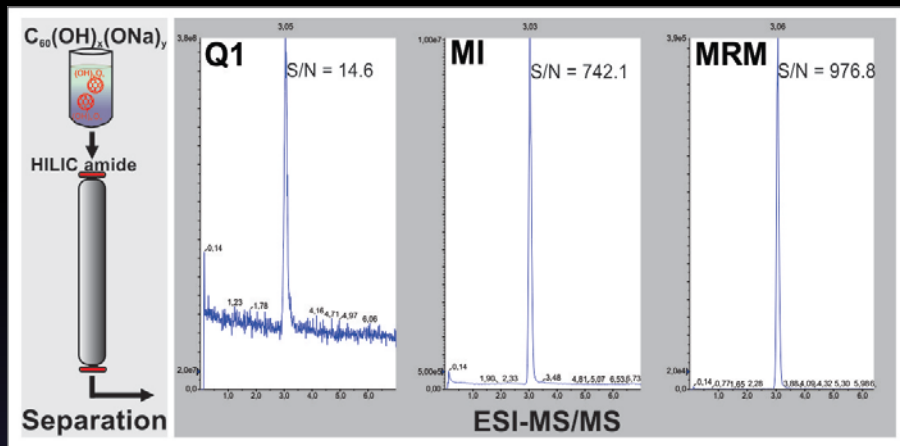
# Fullerols ( $C_{60}(O)$ , etc)



- Light scattering ( $\lambda_{347\text{ nm}}$ )
- Liquid phase combustion (TOC)
- Thermal optical transmittance
- LC/MS using 720 m/z



# Fullerol – Comparison of Detection Methods

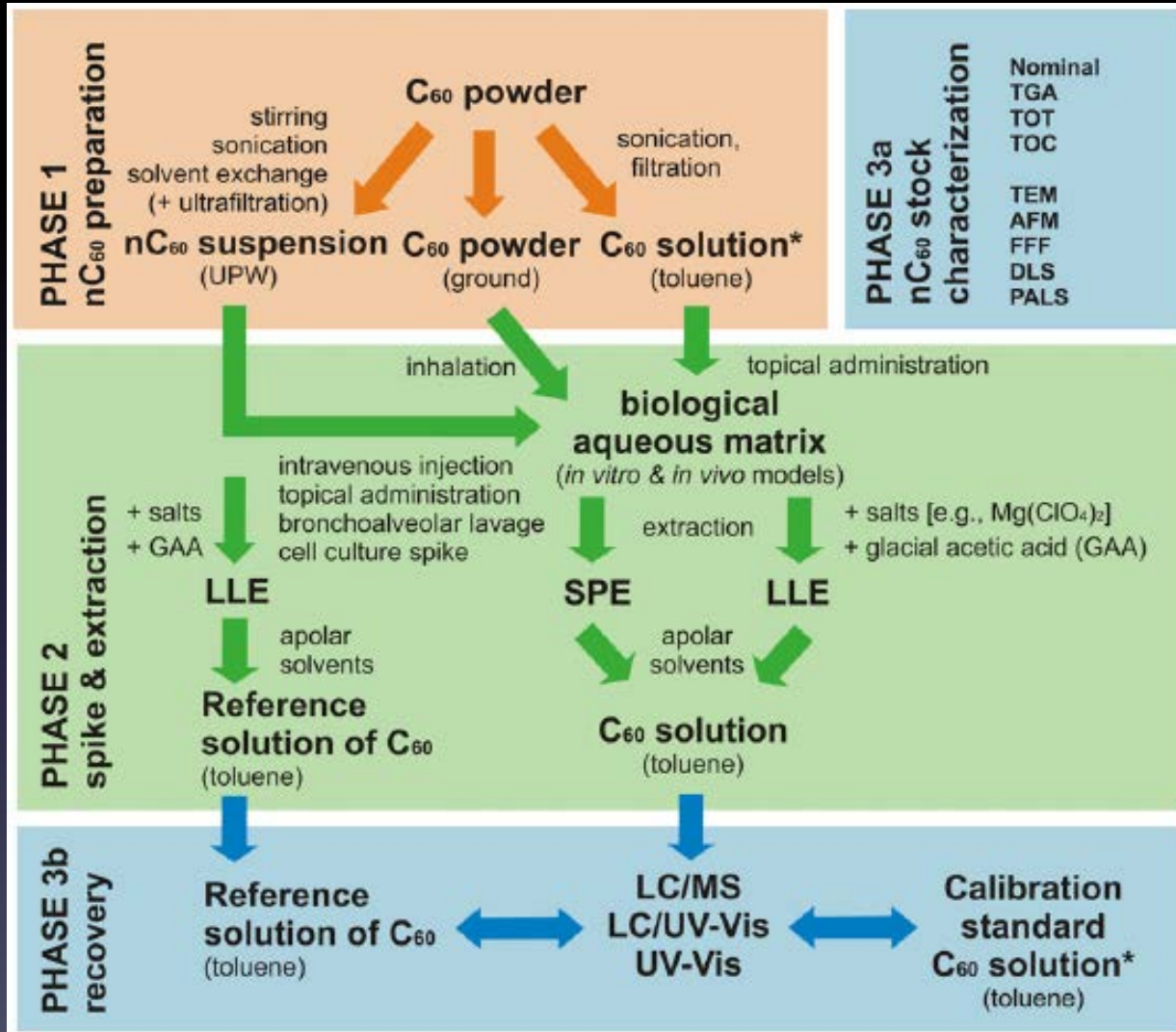


single quad scan, Q1  
 multiple ion monitoring, MI  
 MS with multiple reaction monitoring, MRM

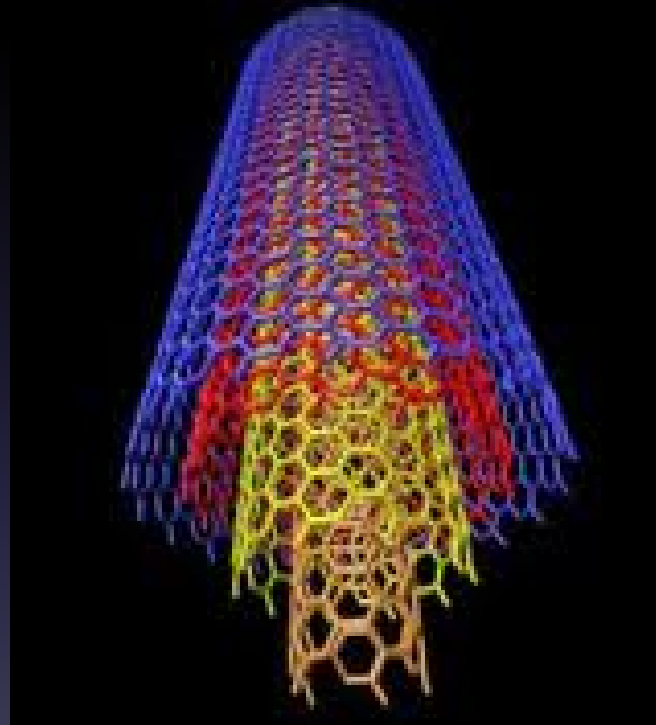
*No standards available*

	R <sup>2</sup> <sup>a</sup>	MDL <sup>b</sup> [pg/mL]	RSD <sup>c</sup>	SRFA <sup>d</sup>
UV/Vis	0.999	42 780	n.d.	n.d.
Q1 scan	0.9996	125	2.9 %	29.4 %
MI scan	0.9999	1.5	0.7%	8.6 %
MRM scan	0.9999	0.19	0.5 %	2.5 %

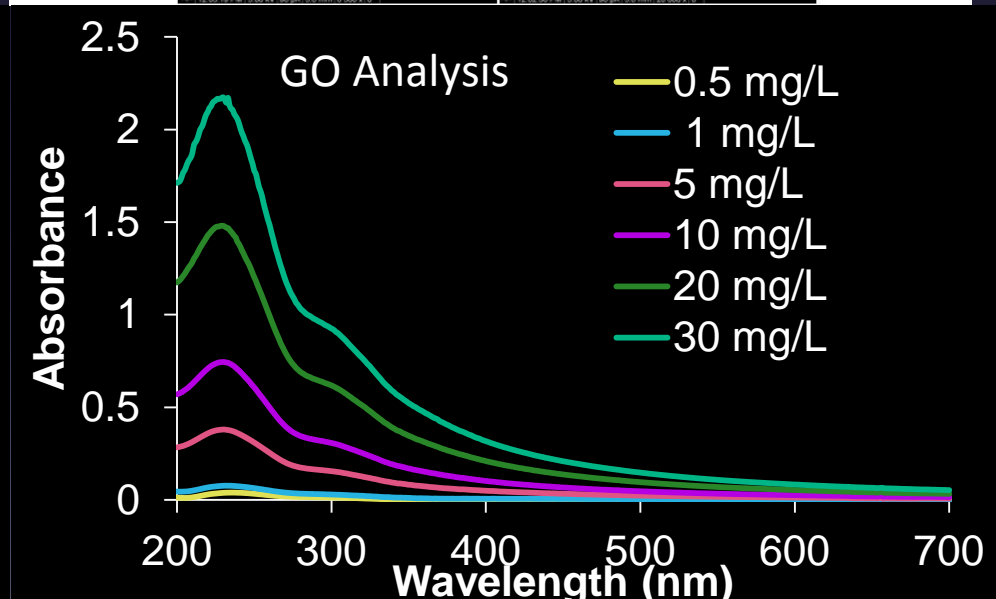
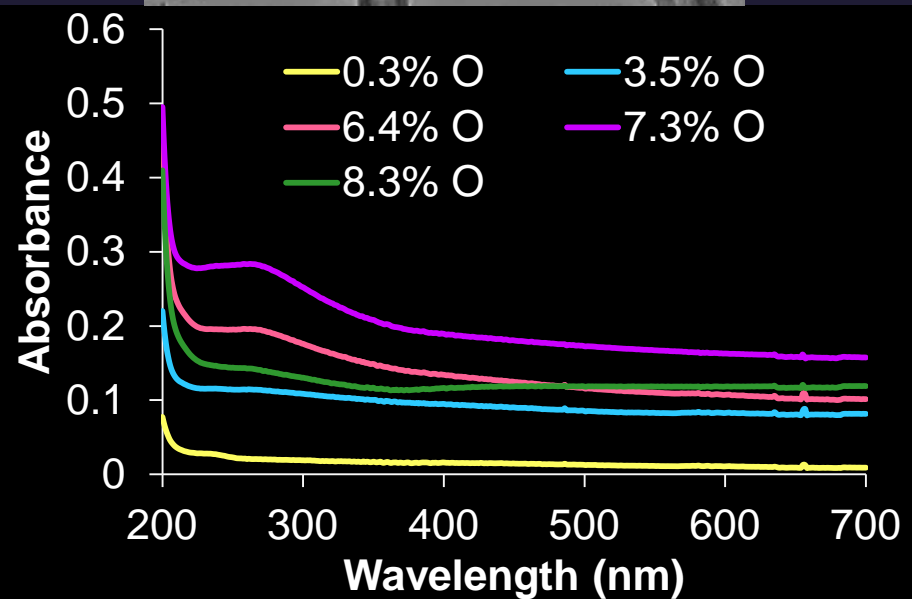
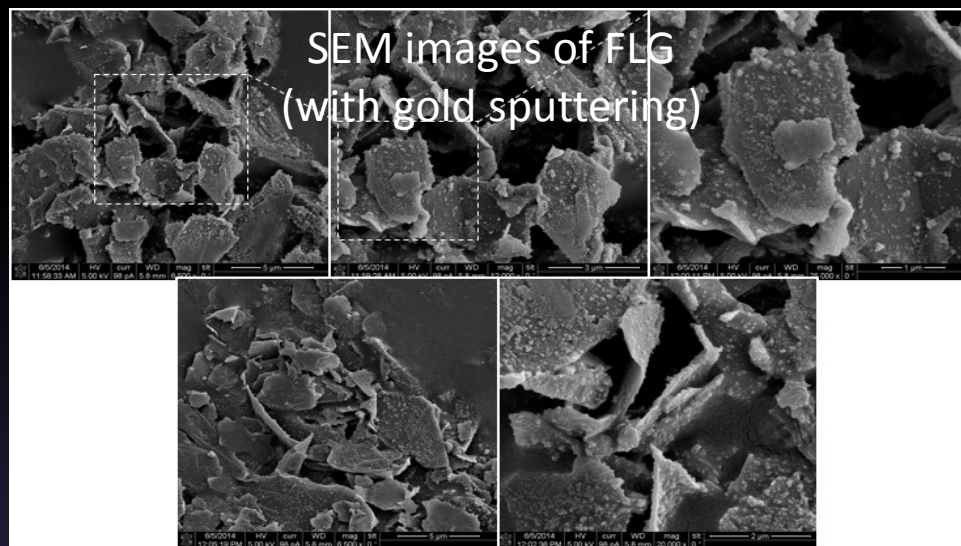
# Fullerene Summary



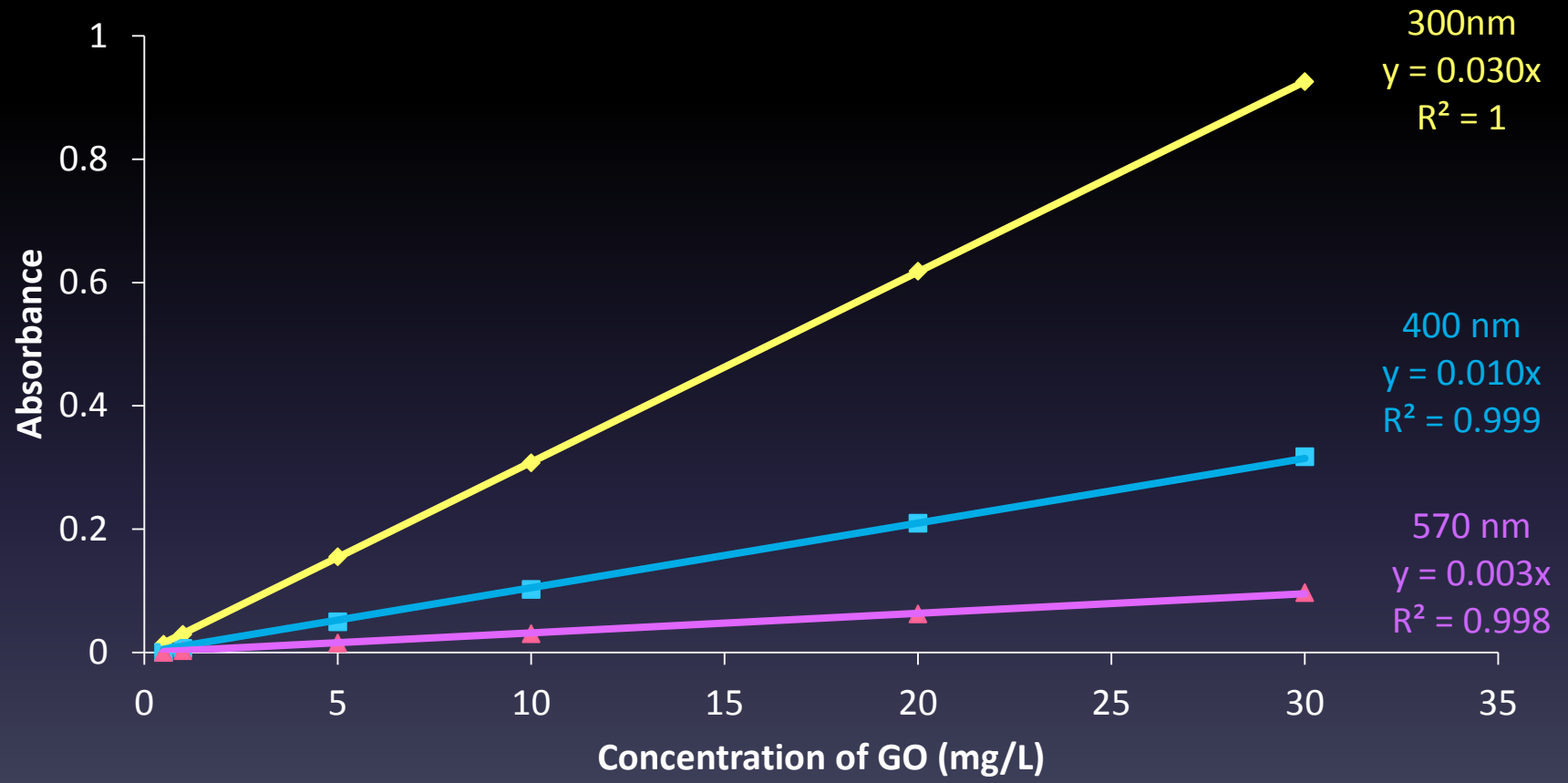
# Quantification – CNTs



# CNT & Graphene UV/VIS Absorbance



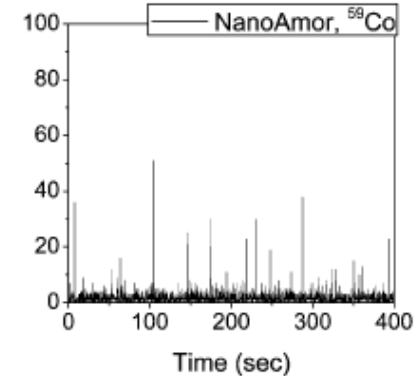
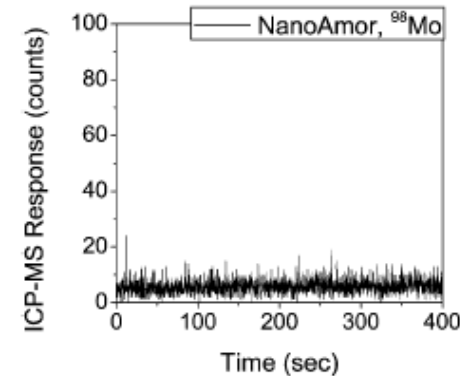
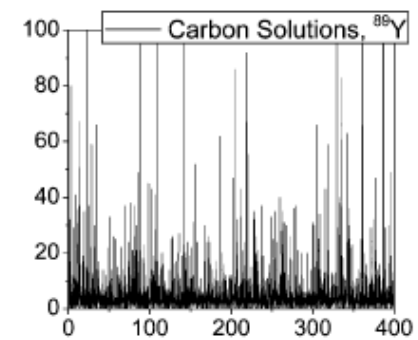
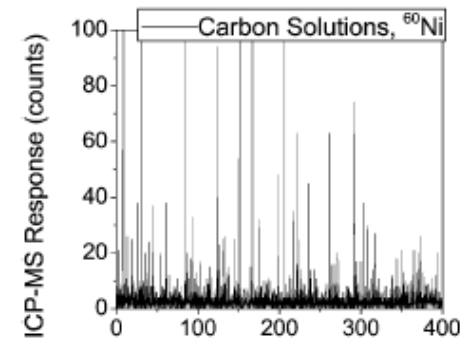
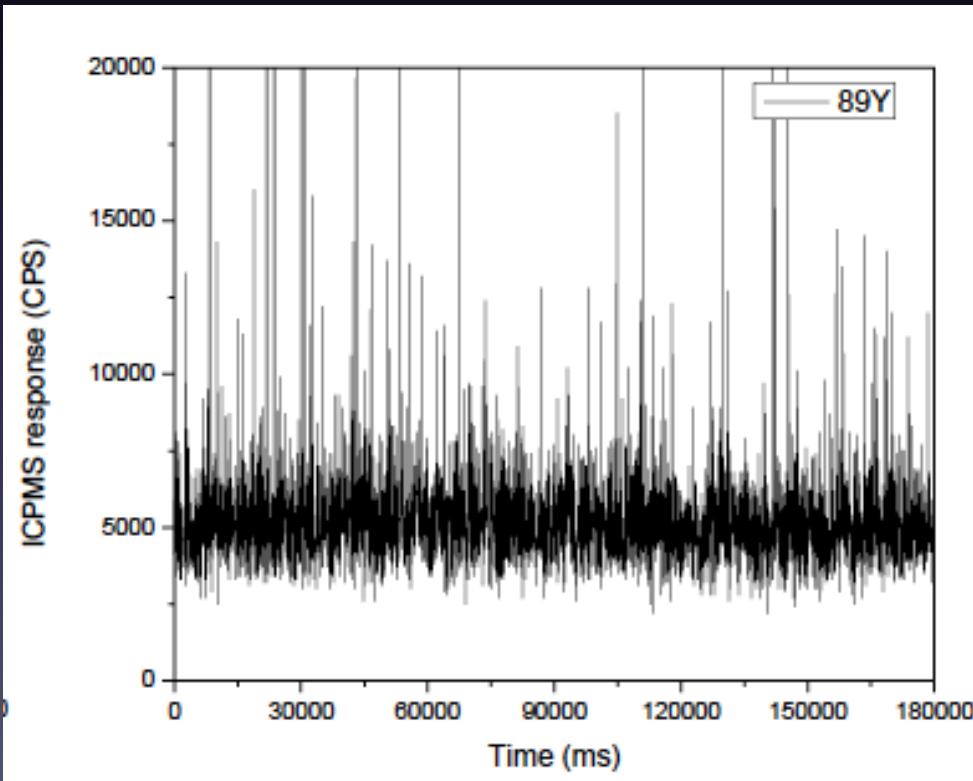
# Standard curve for GO at different wavelengths



# Indirect Measurement

Reed et al., 2013

- Single Particle ICP-MS (spICP-MS)
  - Track residual catalyst rather than carbon
  - Ranville et al. showed for SWCNT
  - Our group has evidence for MWCNT



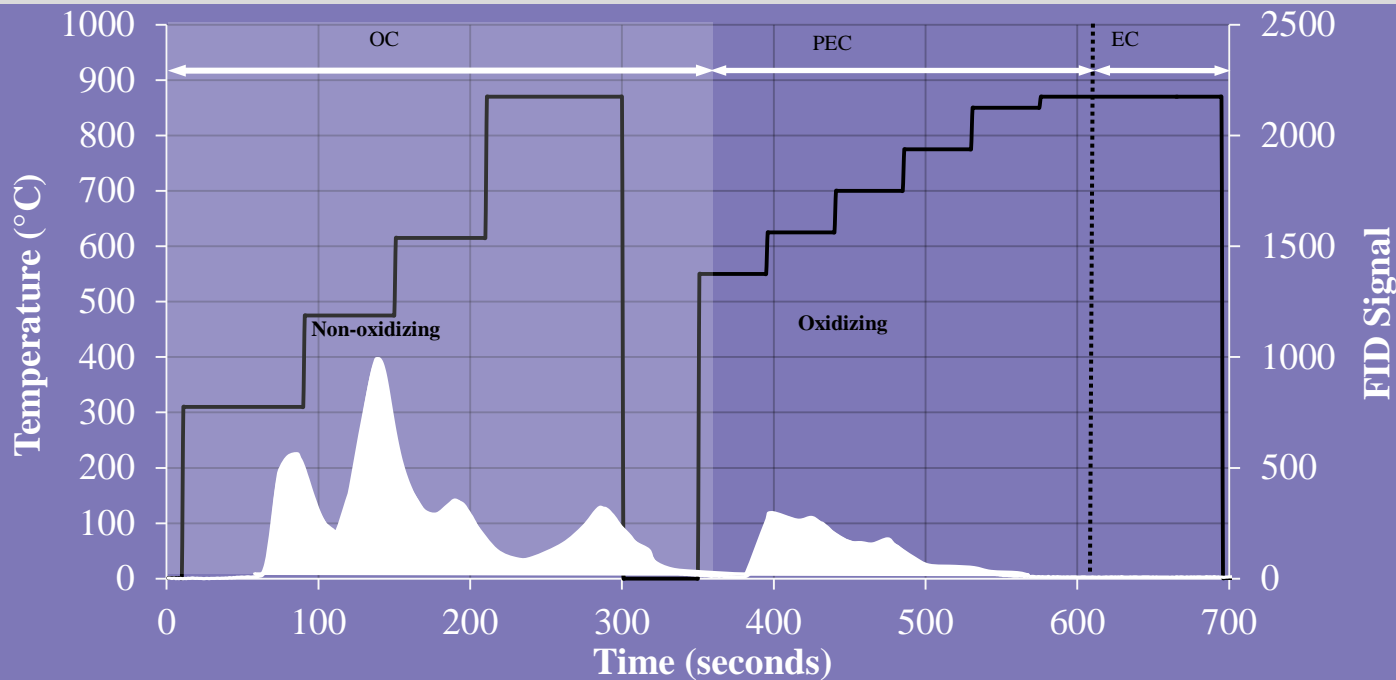
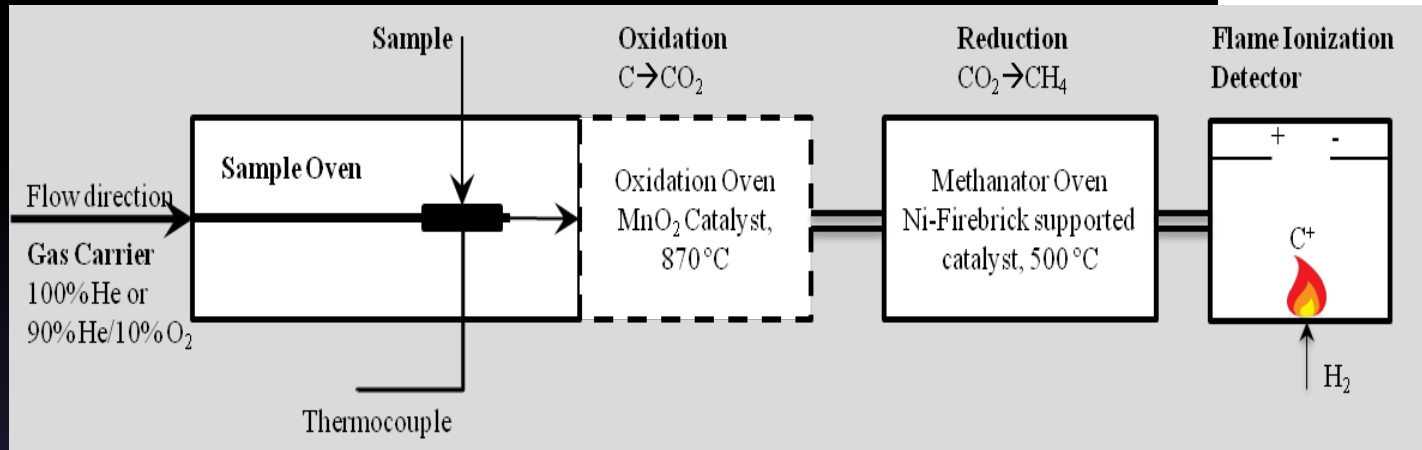
**Fig. 1** Real-time ICP-MS response data, for determination of the best analyte metal for CNTs used in this study. The CNTs were at  $1 \mu\text{g L}^{-1}$  to ensure enough CNTs would be in solution for analyte comparison. Only data for one isotope each of Ni (<sup>60</sup>Ni) and Mo (<sup>98</sup>Mo) are shown here. Other isotopes were not as usable due

Pulses  $\approx ? \approx$  CNTs

# Thermal Optical Transmittance (Temperature Programmed Oxidation)



Sunset  
Laboratories Lab  
OC-EC Aerosol  
Analyzer

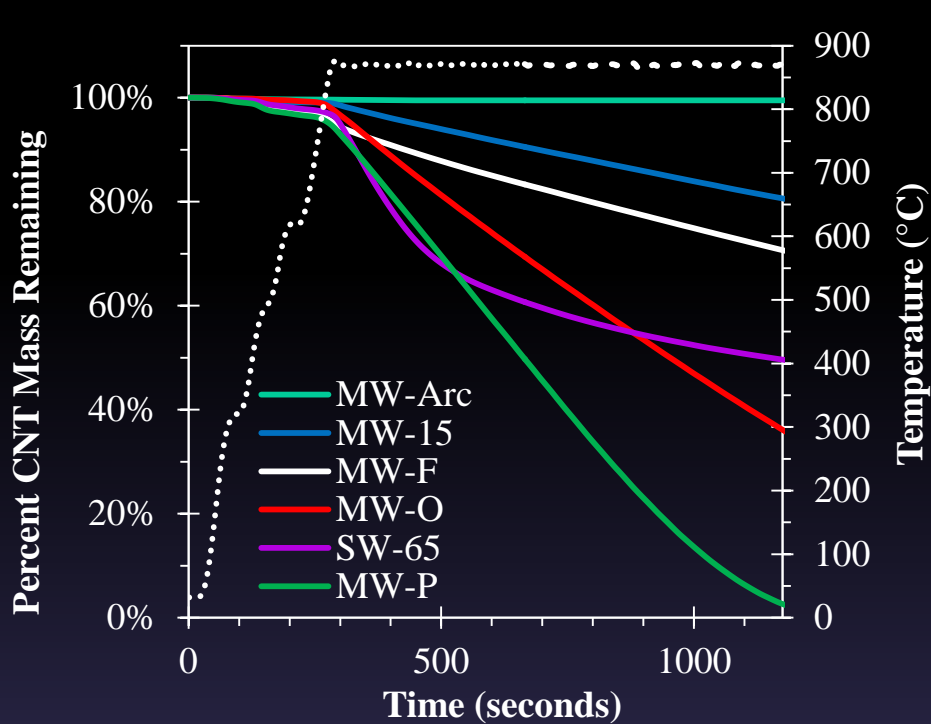




# Comparison of 15 CNTs

CNT ID	CNT Type	State	Purity <sup>a</sup>	Metal Content	Outer Diameter (nm)	Inner Diameter (nm)	Length (μm)
MW-O	MWCNT	Raw	>95%	<5%	20-30	5-10	10-30
MW-P	MWCNT	Purified	>98%	<2%	20-30	5-10	10-30
MW-F	MWCNT	Functionalized	>99.9%	<0.01%	20-30	5-10	10-30
MW-100	MWCNT	Raw	>95%	<5%	60-100	5-10	0.5-500
MW-30	MWCNT	Raw	>95%	<5%	10-30	5-10	0.5-500
MW-20	MWCNT	Raw	>95%	<5%	10-20	5-10	0.5-200
MW-15	MWCNT	Raw	>95%	<5%	7-15	3-6	0.5-200
MW-Arc	MWCNT <sup>c</sup>	Raw	<50%	0%	5-10 <sup>b</sup>	-	-
MW-15G	MWCNT <sup>d</sup>	Annealed	>97%	<1%	7-15	3-6	0.5-200
MW-Mitsui	MWCNT	-	>98%	<1%	20-70	-	-
MW-OH	MWCNT	Functionalized	>95%	<1.5%	8-15	3-5	10-50
MW-COOH	MWCNT	Functionalized	>95%	<1.5%	8-15	3-5	10-50
SW	SWCNT	Raw	<50%	<10%	1.1	-	0.5-100
SW-65	SWCNT	Purified	<75%	<10%	0.8	-	0.45-2

# Thermal Properties of CNTs

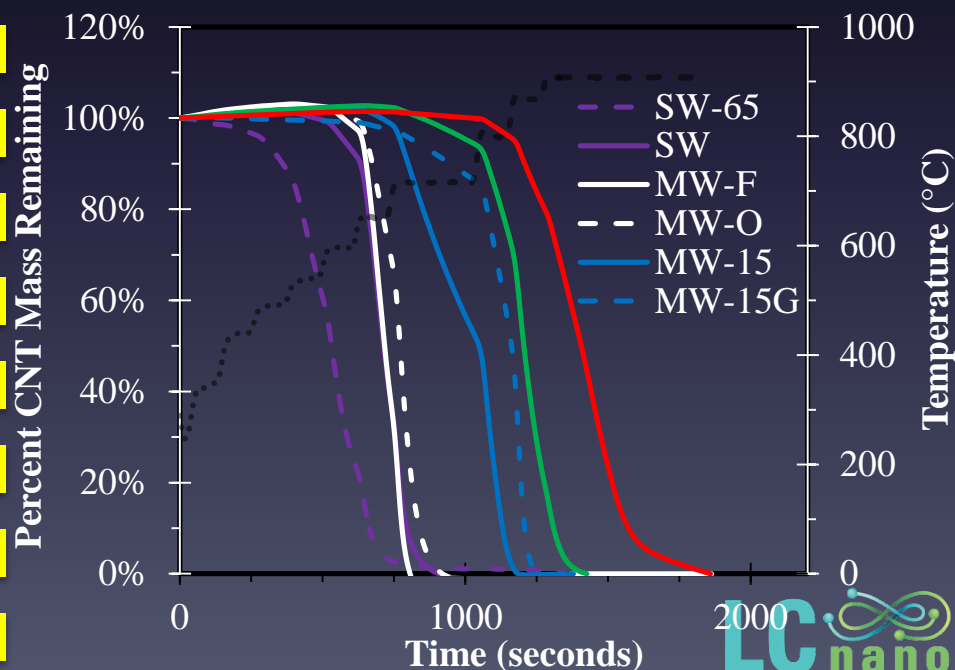


Oxidizing conditions  
(90% He/ 10% O<sub>2</sub>)

**Conclusion: Not all CNTs  
"burn" at the same  
temperature**

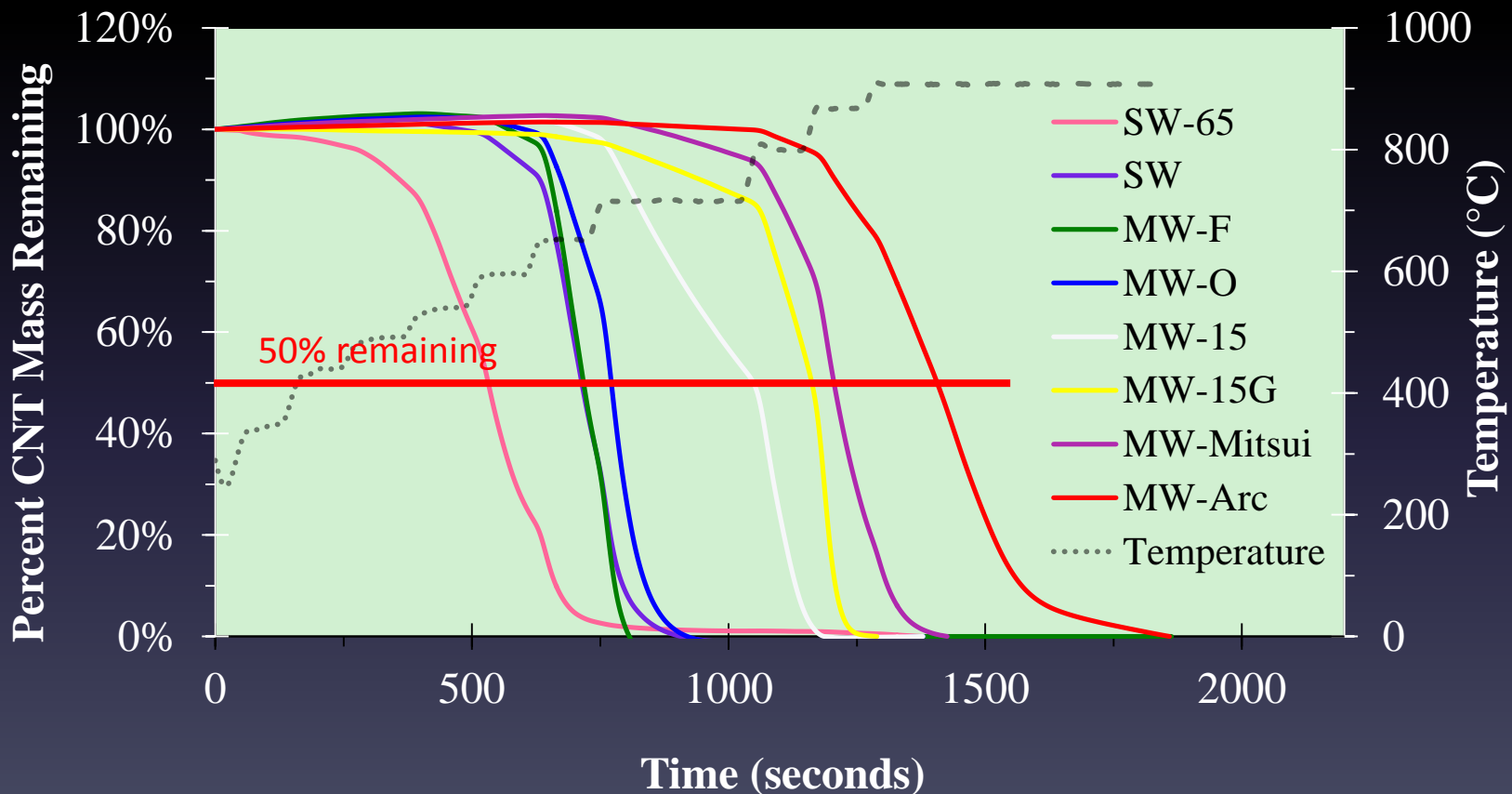
Inert conditions (100% He)

**Conclusion: Surface  
oxygen groups allow some  
CNTs to burn in inert gas  
environment**

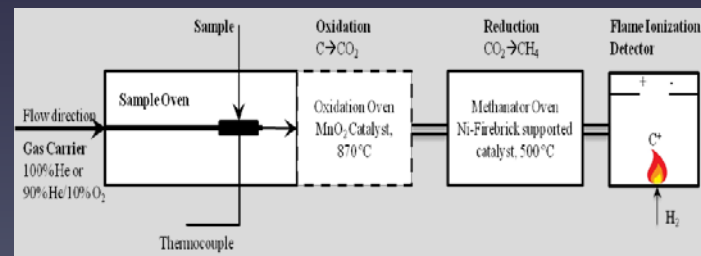
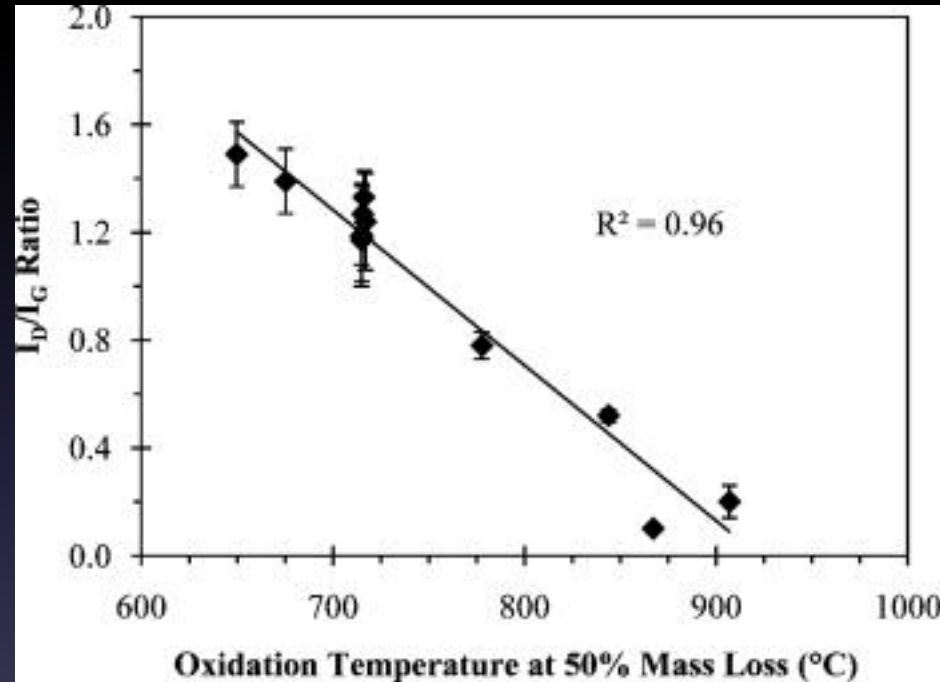
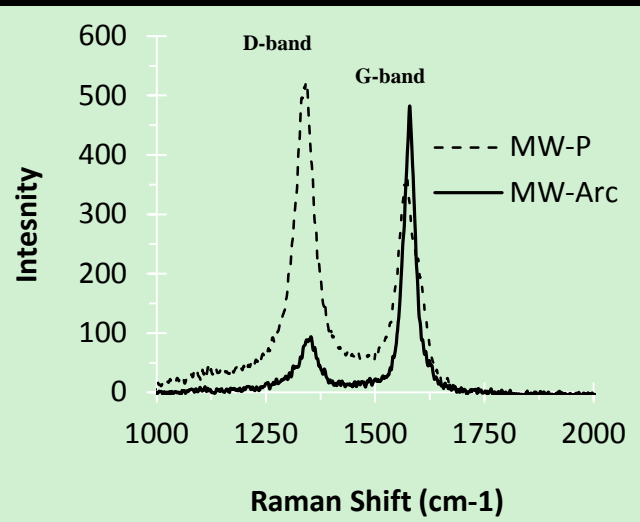


# Temperature Comparison for 50% CNT combustion

PTA method is a refinement to NIOSH Method 5040

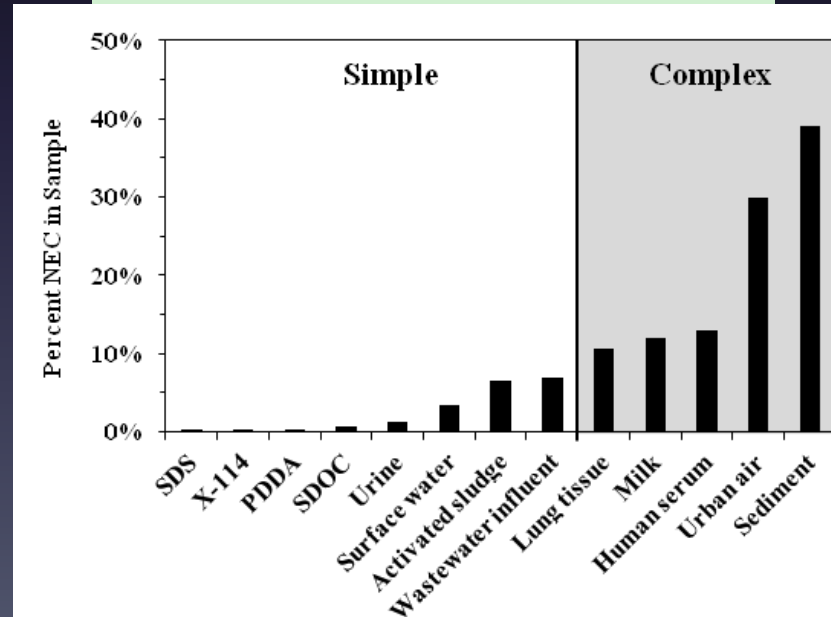
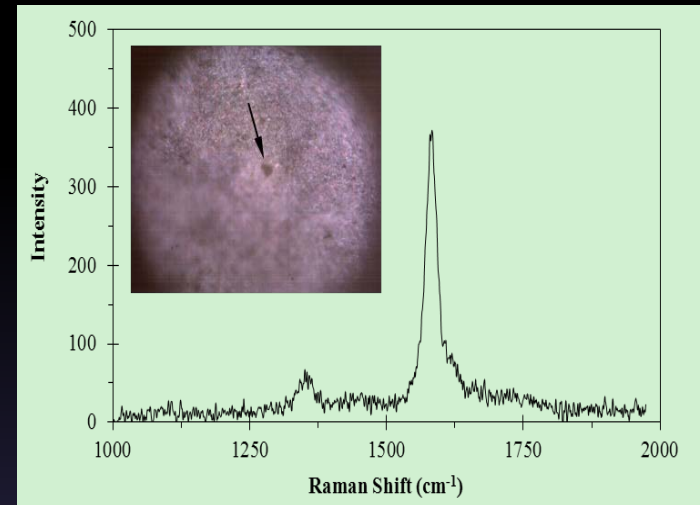


# Thermal Properties of CNTs related to Structural Defects



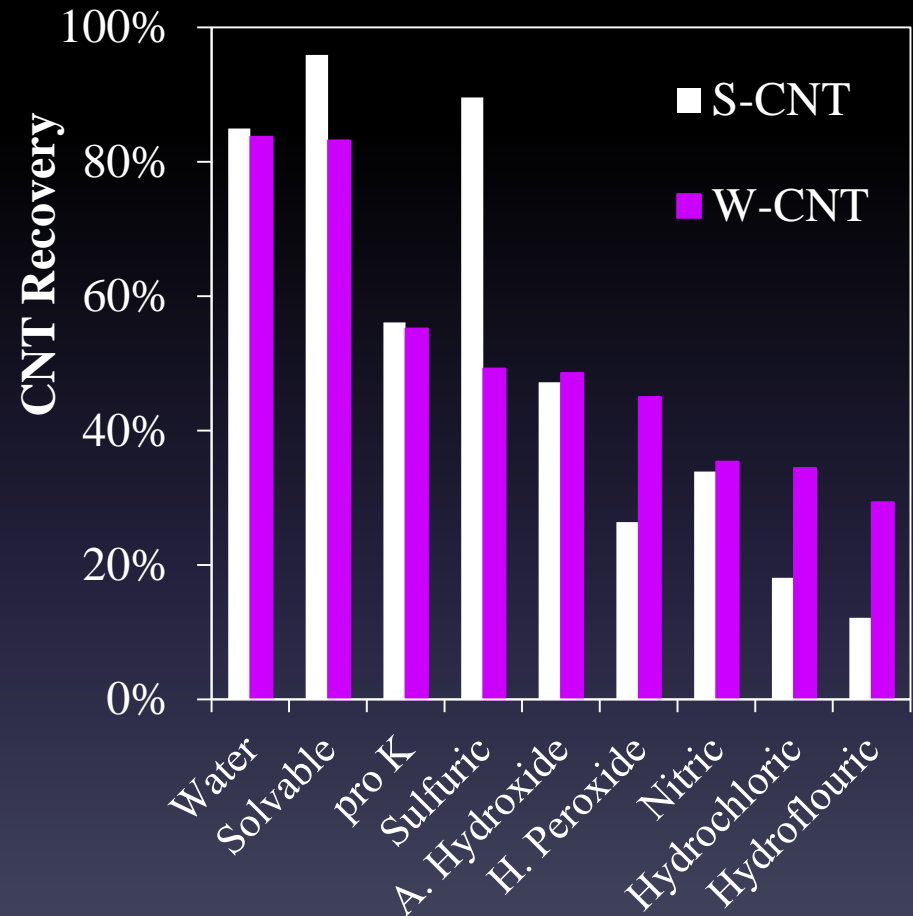
# CNTs & Other NM Detection in biomass

- Raman and imaging can detect CNTs, but not quantify them well
- Extraction protocol must:
  - Minimize oxidation of CNT
  - Remove interfering background organic carbon (from rat lung tissue)
  - Separate solid-phase CNT from liquid-phase dissolved tissue

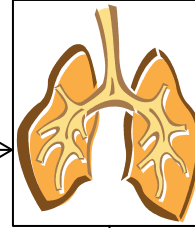
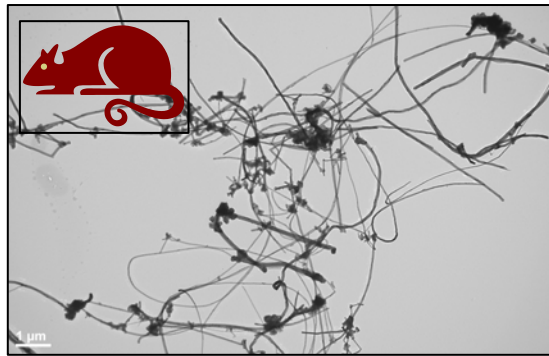


# Selective digestion can remove organic matter & facilitate CNT quantification

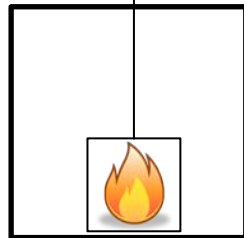
- Can solvents remove organic matter?
  - Yes
  - Oxidants ( $\text{H}_2\text{O}_2$ )
  - Acids ( $\text{HNO}_3$ ,  $\text{H}_2\text{SO}_4$ )
  - Alkali ( $\text{NaOH}$ ,  $\text{KOH}$ ,  $\text{NH}_4\text{OH}$ , Solvable)
  - Enzymes (TMAH, ProtK)
- But, do solvents affect CNT detection?
  - Mostly, yes
  - Surface oxygenation results in combustion at lower temperatures
  - Enzymes and customized alkali fair best



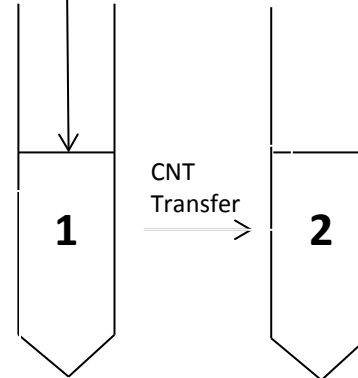
# Application of Extraction Method to Rat Lungs



Result: CNT Dose



Alkali  
Treatment  
(Solvable)



Enzymatic  
Treatment  
(Proteinase K)

Analyze CNTs



Collect CNTs

Wash/Centrifuge



**2.9 ± 0.19 μg CNTs, whole lung – 93% recovery**

# Hazard & Exposure Analyses

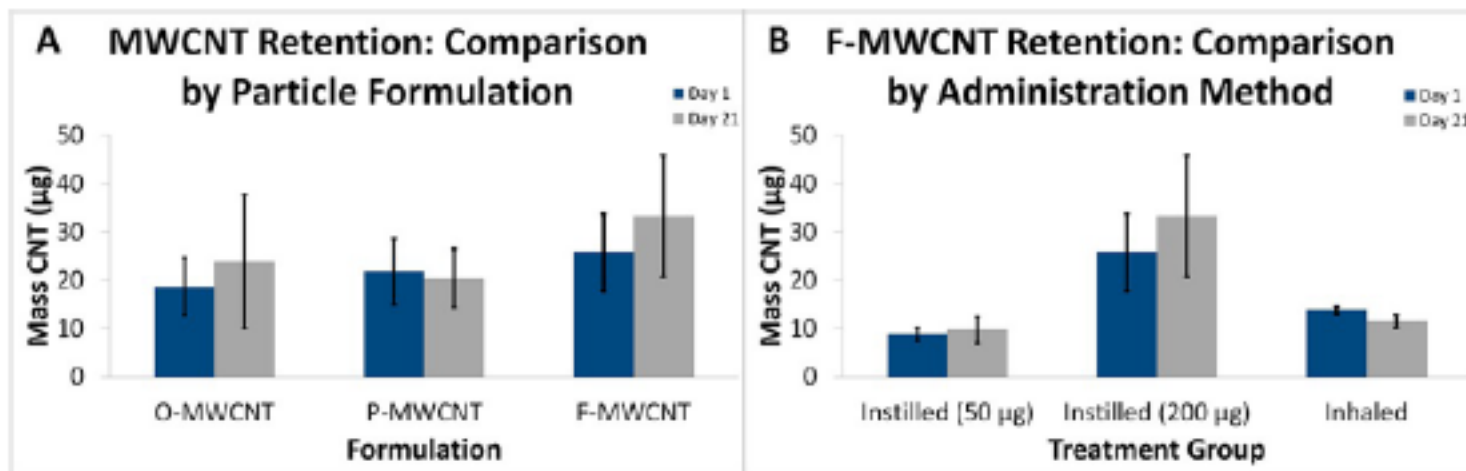
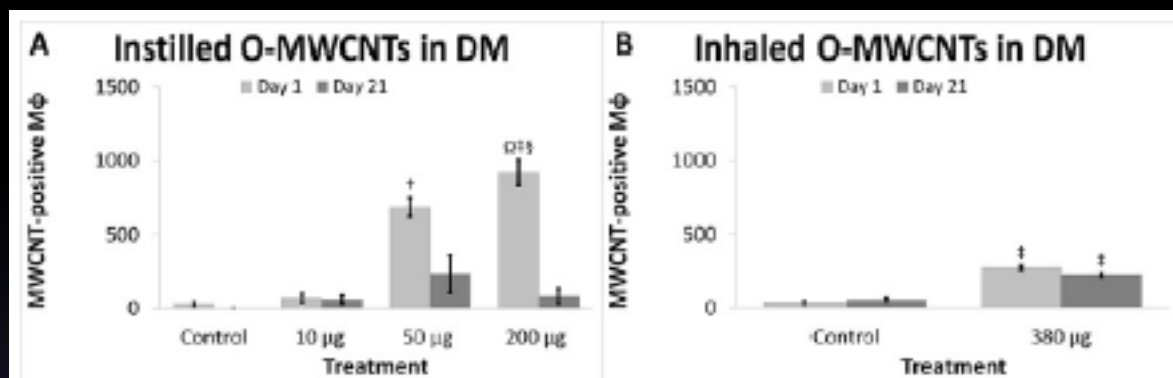
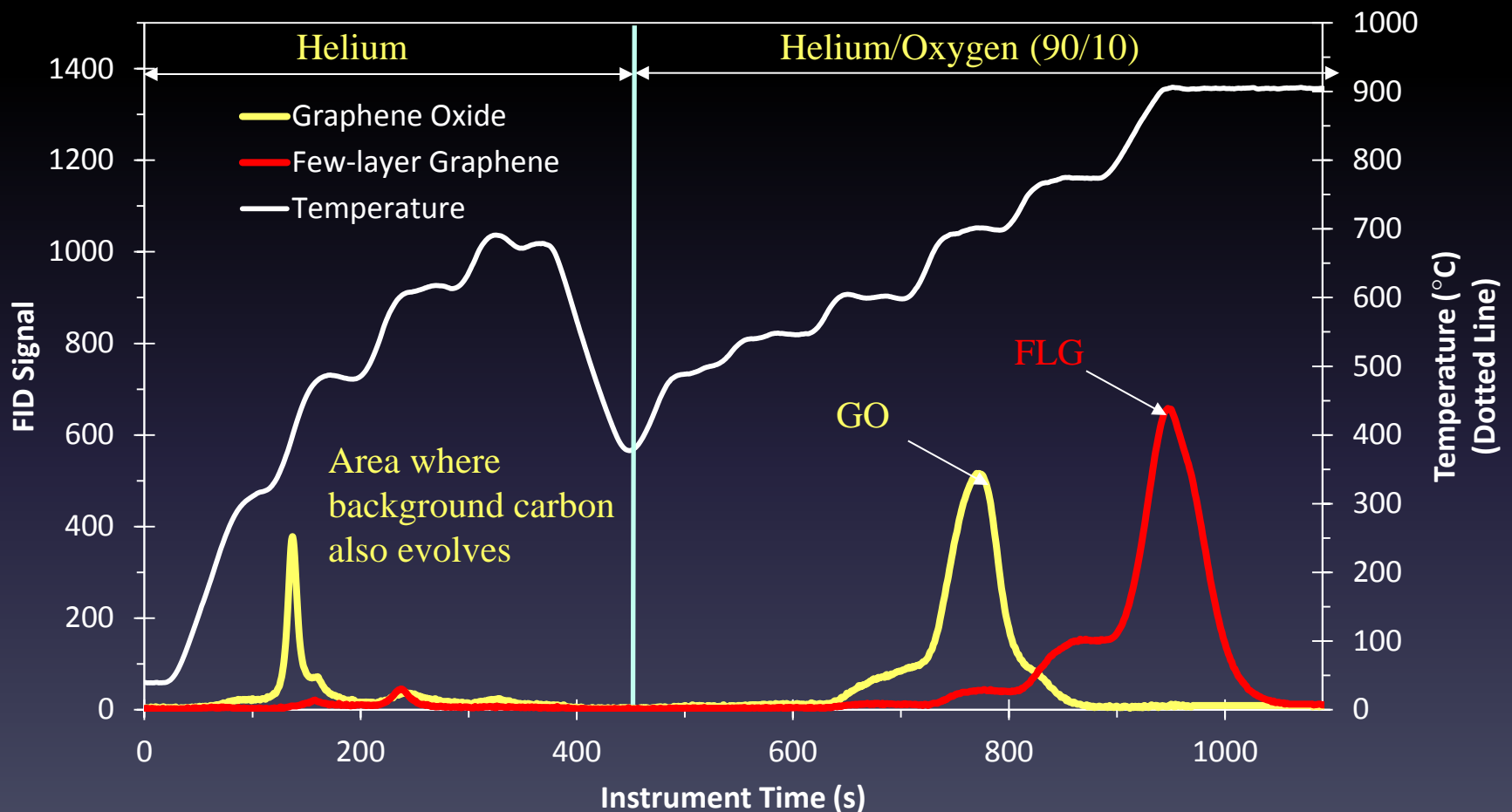


Figure 12. Programmed thermal analysis quantification of MWCNTs in lung tissue. (A) Day 1 and 21 time-course data from animals instilled with O-, P-, or F-MWCNTs. (B) Time-course data from animals exposed to F-MWCNTs by intratracheal instillation and inhalation.



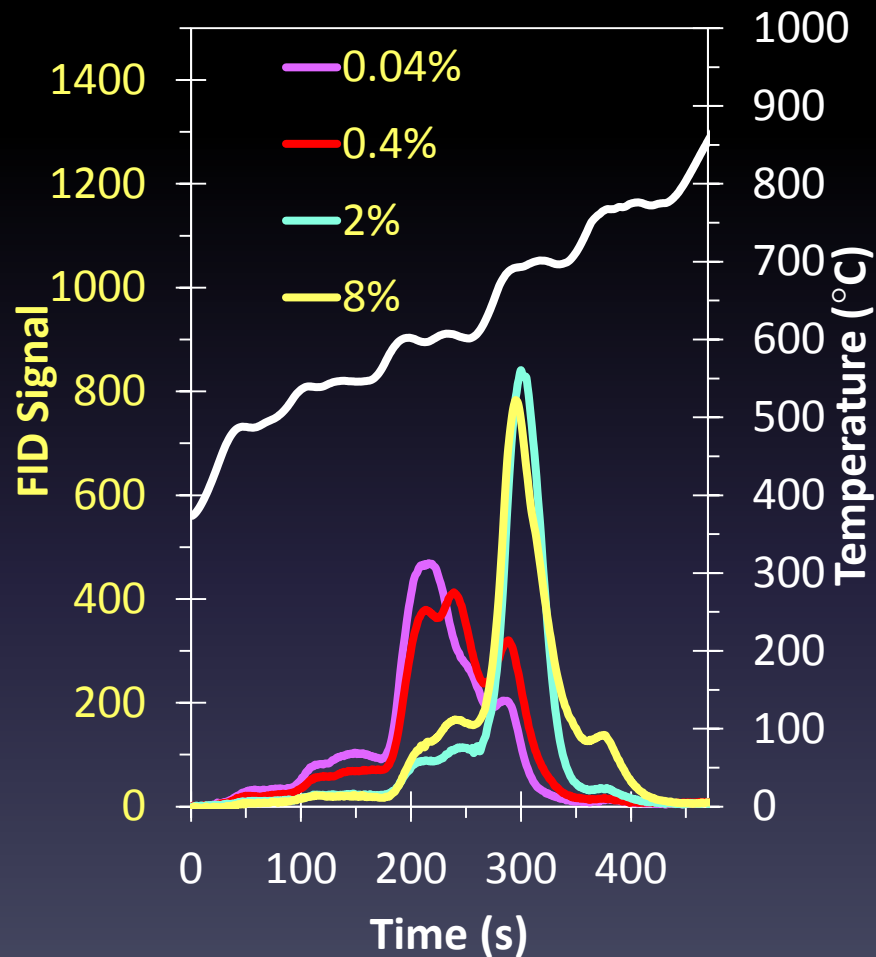
# Similar Approach for Graphene (GO & FLG)

## Step 1- Programmed Thermal Analysis (PTA)



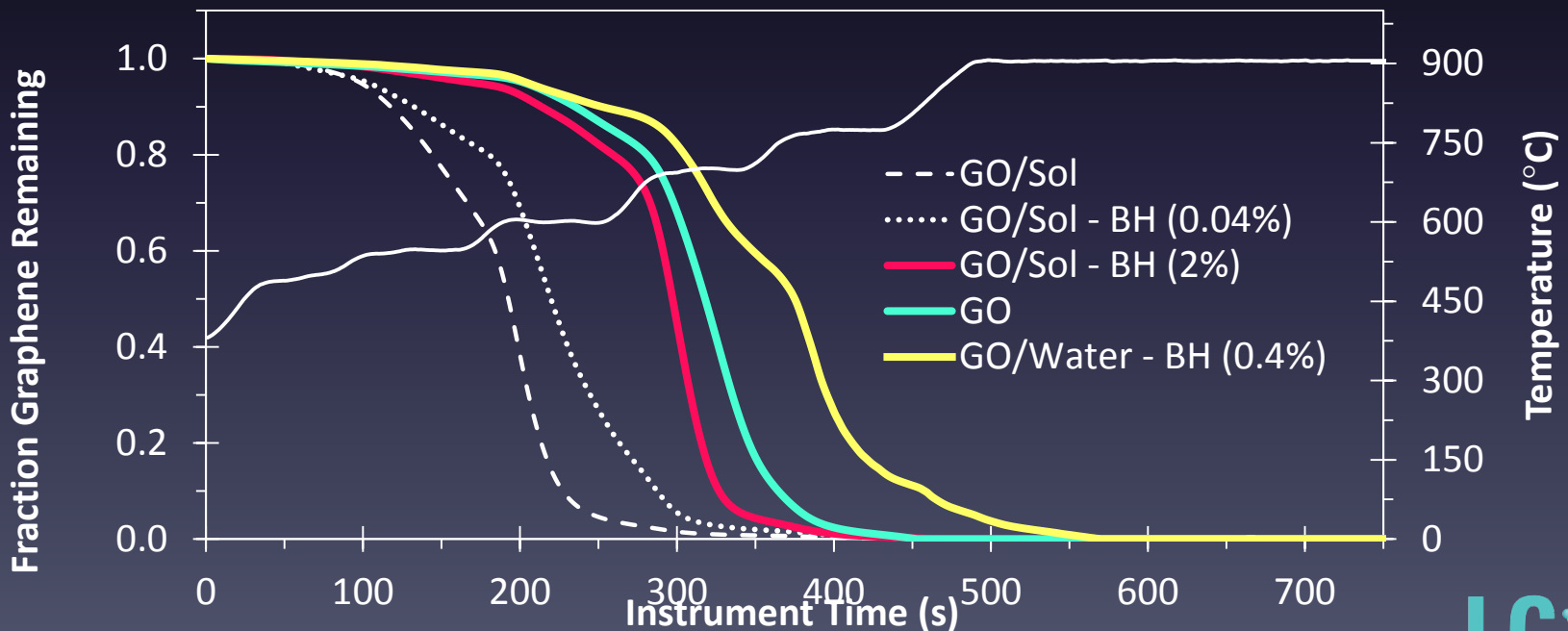
# Step 2: Improve separation of GO signal from background organics

- Add reductant ( $\text{NaBH}_4$ )
- Reduced graphene oxide (RGO) analysis by XPS yields decreases number of C-O & C=O bonds by > 5 fold
- PTA thermogram improves

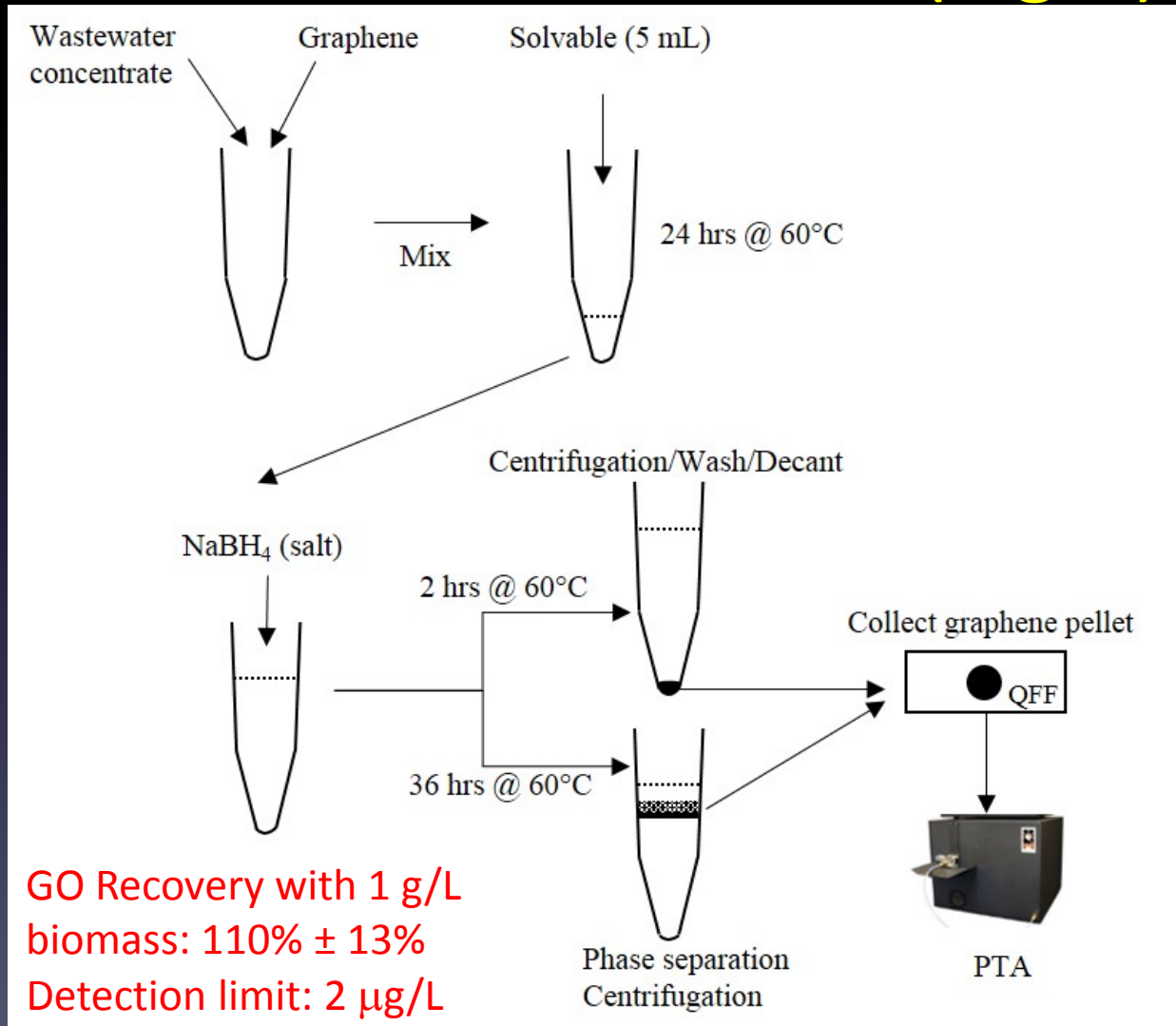


# Step 3: Adding Solvable™ to degrade organics

- Solvable is an alkaline digestate that degrades organics; surfactant helps separation
- Solvable + NaBH<sub>4</sub> produces good pellet for separation & analysis



# Final Digestion Method to Handle Separation of FLG, GO (or CNT) from High Biomass Concentrations (1 g/L)



# Carbon NM Monitoring in Air Samples?



- Goal: To quantify the presence of CNTs in the presence of background air particulates
- Samples analyzed for organic carbon and CNT by PTA – which is a refinement to NIOSH Method 5040

## Recovery of CNTs on air filter samples

(Conclusion: Excellent CNT recoveries indicates viability to monitor CNTs in workplace air)

Indoor air (MWCNT spiked onto filter)

Spiked CNT / ug	TOT data / ug
1	1.00±0.15
5	4.35±0.32
10	9.59±0.58

Outdoor air

Spiked CNT / ug	TOT data / ug
1	0.80±0.17
5	4.48±0.36
10	10.06±0.63

# Conclusions

- C<sub>60</sub> derivatives
  - Extraction in solvent (toluene) gives lowest detection limits using LC-MS
  - Solvent extraction from tissue and commercial products is possible
  - Extraction from urine and fluids can use solid phase extraction
- Graphene (FLG/GO) & CNTs (SW / MW)
  - Thermal methods can be non-selective unless related to known CNT or Raman analysis
  - spICP-MS emerging as potential indirect approach to quantify metal catalyst rather than carbon itself
  - Extraction from tissues aim to minimize oxygen incorporation => Alkaline conditions are better + Enzymatic digestion
- Personal monitoring devices can collect Carbon Materials and filters can readily be extracted for analysis