TiO₂ and Ag Nanoparticles in a River Environment

Transformation in the organism and in the environment:

What do we measure and how do we develop testing strategies to measure impacts of transformed particles in the environment

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Ann Miracle, Ph.D.



Understand the transformation of nanomaterials under different environmental conditions

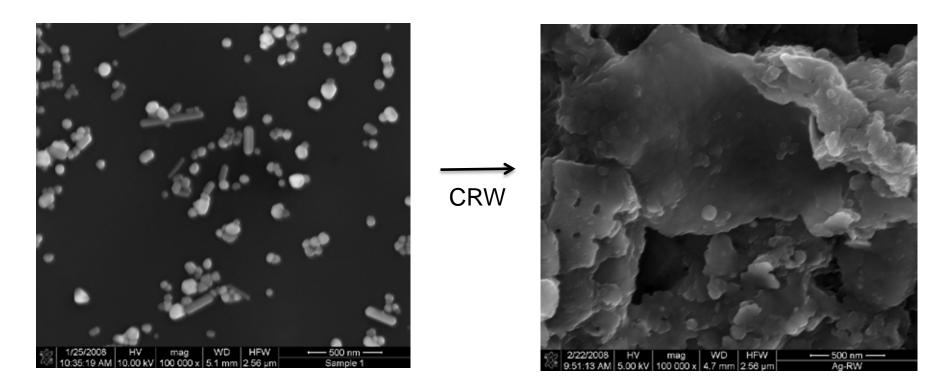
- Titania and Silver nanoparticles in a simulated river/sediment system
 - Columbia River water (TSS= 7 mg/L; pH=7.65; hardness=77 mg/L as CaCO₃)
 - Sand sediments

Titania and Silver citrate in static cells and flow through river mesocosms

- Microbial community changes (static only)
- Uptake by clams and amphipods
- Deposition on sediments
- Aggregation in flowing water



Silver Citrate Materials



30 - 200 nm for spheres

80 – 400 nm x 30 – 50 nm for rods



Microbial Community Silver Exposures

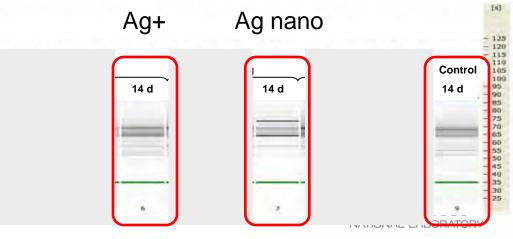
Static Exposure Study üHomogenized sediment from surface water mesocosm üExposures (1, 4 and 14 d): Doses in CRW (detection limit 3 ng/L): 1 ug/g Ag nano 4 ug/g Ag+ •Shift in domina

Controls

5.0 4.0 Total Ag (µg/g) 125 120 Control 115 Ag lonic 110 105 Ag Nano 2.0 1.0 60 45 0.0 30. 8 10 12 14 2 25 Exposure Time (Days)

•Shift in dominant microbial species at 14 days

•Ag nano had greater community shift than Ag+



Silver Mesocosm Exposure



24 hr exposure, 24 hr depuration

- Columbia River water (CRW)
- Clams
- Amphipods
- Microbial community in sand sediments

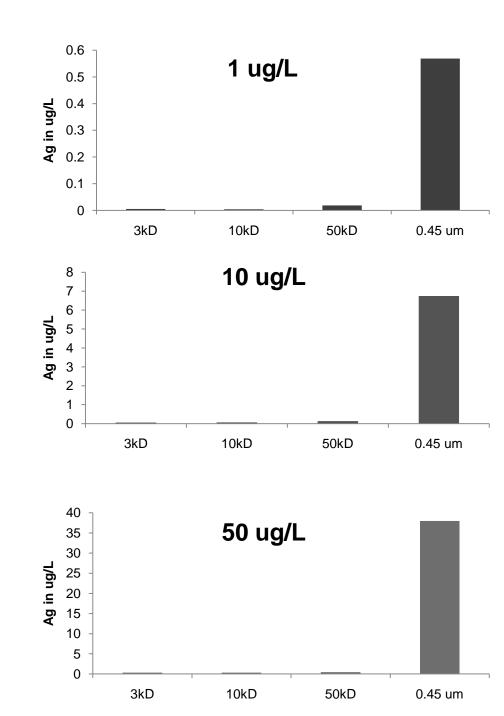
Control,1 µg/L, 10 µg/L, 50 µg/L



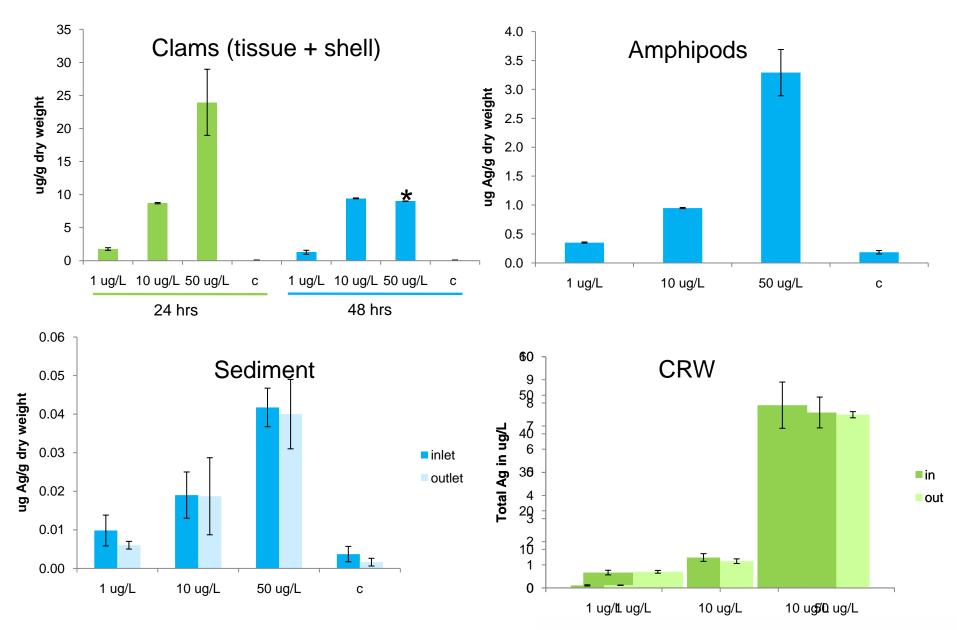


Ag particle size in CRW

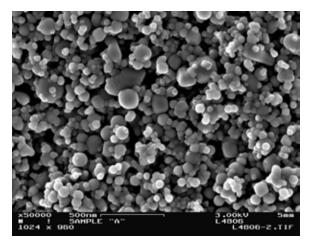
- Low concentrations of dosed Ag nanoparticles fractionated to larger particle sizes
- Degree of fractionation occurs over 24 hours
- Prior studies show dissolved fractions at doses > 100 ug/L



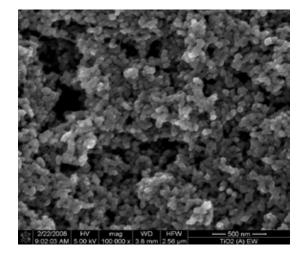
Accumulation of Silver

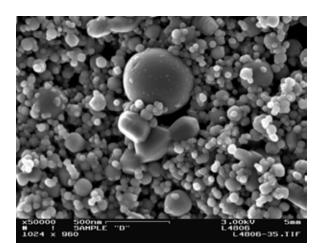


Titanium Oxide Materials



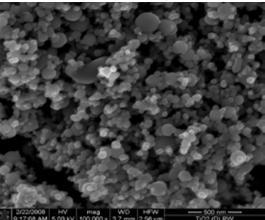
5-30 nm anatase





CRW

CRW





<75 nm rutile/anatase

Titania Mesocosm Exposures



-5 mg/L over 12 hour flow-through-36 hr flow-through depuration

Titania exposures



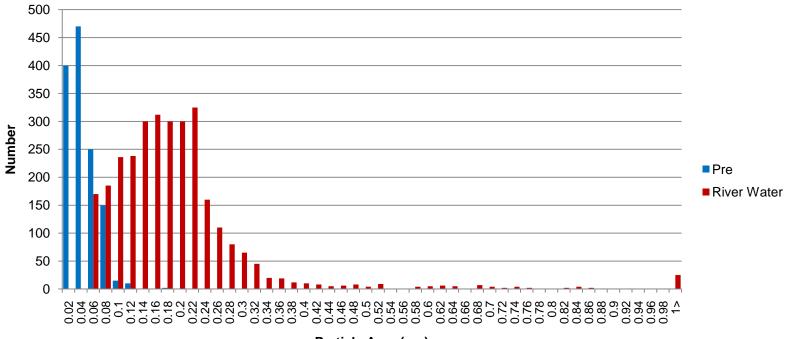


Variable	TiO ₂ (mg/g dry weight)		
	or		
	% total dose		
	<u>(5 mg/L)</u>		
	Flow - Through Static*		
	A	A/R	А
amphipods	47.9	64.8	2.1
clams	0.55	1.04	0.03
sediment	66%	13%	34%

Clam : Amphipod uptake ratio ~1:70



TiO₂ Size Distribution from SEM



Particle Area (um)

- Mean equivalent diameter*
 - Distilled Water 30 nm
 - CRW 200 nm



Two Materials – One Exposure Scenario Abiotic and Ecosystem-Wide Effects

NP size affected by environmental characteristics

- Specific properties of NP material may affect bioaccumulation and downstream ecosystem impacts
 - Silver uptake higher in clams; stays in water column
 - Titania uptake higher in amphipods; settling out greater

Acute toxicity not observed in Columbia River water



Research Gaps Remain

- NP toxicity/effect may be different in a complex environmental setting compared with single variable/static lab exposures
- Chronic (long-term) studies under complex environmental conditions need to be matched with ability to measure and characterize NPs in complex environmental samples
 - absorption, distribution, metabolism, excretion
 - recycled NPs
 - route(s) of exposure absorption, dietary



Case Study

- Seeing changes that reflect ecosystem scale disturbance
 - Birds, fish dead
 - Deformed frogs
 - Selective flora die-offs
- Relevance of materials in complex matrix
 - New paradigm vs. a standard tier-testing approach?
 - Choice of organisms for toxicity endpoints
 - Transformation of materials in complex media

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