### **Characterization of Mechanical and UV-Induced Nanoparticle Release from Commercial Products**

Lipiin Sung<sup>1</sup>, Keana Scott<sup>2</sup> & Treye Thomas<sup>3</sup> 1Engineering Laboratory, NIST, Gaithersburg, MD 2Material Measurement Laboratory, NIST, Gaithersburg, MD 3Office of Hazard Identification and Reduction, CPSC, Rockville, MD

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# **Nano Release at NIST**

- NIST-CPSC Projects
	- MWCNT, metal oxide & inorganic nanoparticle release from commercial products
	- Nanomaterial release from fire retardant products
- NIST Projects
	- MWCNT release from composite materials
	- MWCNT release visualization
	- Impact of weathering on nanoparticle release from composite materials

# **Mechanically induced MWCNT release from nanocomposites**

- Characterization of intact nanocomposite materials
	- Raman, SEM & TEM
	- Commercial materials often have carbon fibers as well as MWCNTs – additional analytical challenges
- Mechanical release cutting, sawing, abrasion
- Released particle collection and analysis
	- Passive collection, MOUDI, electrostatic precipitator, filtering
	- Real-time particle analysis CPC, SMPS
	- Release particle analysis Raman, SEM/STEM, LM

# **Passive sample collection from sawing and cutting**

- Mostly  $\mu$ m- to mm-sized particles consisting fiber bundles, resin pieces, paint chips, etc.
- Might contain bare or small clusters of nanoparticles.





# **Aerosol sampling challenges**





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# **Nanoparticles from cutting debris**

- *What do we mean by released MWCNT?*
	- Partially embedded
	- Attached
	- Loose



• *Are rod shaped particles MWCNTs?*



• *What about other nano-sized particles?*

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# **STEM in SEM**



- STEM in SEM can provide MWCNT distribution and size information.
- Easier, faster and cheaper than dedicated TEM investigation.

Cannot visualize the wall structure in epoxy matrix (but it can do it with freestanding CNTs).

# **Size separated sampling helps but …**

- Relatively high resolution (30 nm x 30 nm pixel) imaging is needed to located individual CNT particles
- Manual survey is not realistic.





2500 images needed to cover the sampling area fully!



35+ GB of images

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# **Automated SEM imaging**



# **Challenges for nanorelease characterization**

- Better process control for particle sampling
	- Loss through diffusion?
	- Setup (tube length, inlet location, flow rate, collection substrate, etc.) dependent variations
	- Effective size separated sampling
- Automated and faster imaging and analysis process
	- Very small objects (nano) in a large field of view (statistics)
- Data management must be part of the solution

*Quantitative analysis of release may be difficult until experimental processes are fully characterized*

## NIST/EL - CPSC Nanoparticle Release Research

**Release Pathways of Nanoparticles (NP) During the Life Cycle of Nanocomposites: Mechanical, Matrix Degradation, Chemical Dissolution, Fire/Incineration, etc.** 



# Airborne release particles- *working with Indoor Air Quality Group/EL*

- Goal: • To develop test methods and measurement protocols for determining the quantities and properties of nanoparticles released from polymer nanocomposites
- To understand the mechanism that causes nanoparticles to leave the polymer matrix during exposures to the environments

 $\rightarrow$  Providing data needed for assessing and managing potential EHS risks of NP release during nanocomposites' life cycles. The set of the set o

## **Nanoparticle Release Process and Collection**

**Speed of the Contractor** 

**Load** 

**# of cycles of cycles** 

Type of wheels

Taber rotary abraser (ASTM D 4060-14, organic coatings)

Latex interior paint

containing 1.2 % nanoTiO<sub>2</sub>

Control latex in

naint

#### **Mechanical abrasion Matrix Degradation via UV**

**NIST SPHERE High Throughput, High Intensity UV Chamber** 









1. Characterize abraded surfaces (LSCM, SEM, EDX)

atex interior pair

containing 1.2 %

 $200$ 

- 2. Remove Particles from Abraded Surface (TEM grid pressed against the surface or using an Adhesive Tape)
- 3. Collect residues from abrasion wheels

#### 2 &  $3\rightarrow$  Inductively Coupled Plasma-Optical Emission Spectrometry (ICP-OES), SEM/EDX

### **Lesson Learned- Abrasion Test**

• Commercial rotary abraser can be used for nanoparticle release study, but commercial abrading wheels that are composed of a polymer binder and inorganic abrasives  $\rightarrow$  release their own particles  $\rightarrow$  not suitable



• NIST-made deep cross-patch (MW2) or sandblasted (MW4) noncorrosive stainless steel (e.g., 316 SS) wheels having a root mean square (RMS) surface roughness between 5 µm and 7 µm, are suitable for reproducibly abrading in water and in air for coatings and paints containing nanoparticles.







### **Lesson Learned- Abrasion Test**

• Laser scanning confocal microscopy (LSCM) in combination with image analysis relatively fast method for quantifying the number and size distribution  $A_1, O_3$ oxide/inorganic particles accumulated on abraded surfaces having particle size  $\alpha$ than 100 nm (detection limit).





Al2O3

To identify the particles on surface  $\rightarrow$  SEM/EDX, ICP-OES SEM images: particles from Abraded Surfaces (TEM grid pressed against the surface)

### **Case-Study: SiO<sub>2</sub>-PU Exterior Coatings**

- $\triangleright$  Neat PU and 5 % (by mass) nanosilica in PU (commercial, containing UV absorbers)
- $\triangleright$  Nanosilica (surface treated) in suspension
- ▶ Exposed on NIST SPHERE at 50° C and both dry (0%RH) and humid (75% RH) conditions (PU:  $T_q = 40.4 \pm 3^\circ$  C)

**Release Pathways:** 

### **Polymer matrix degradation via UV exposure**

- **Simulated rain test**
- **Abrasion test**

### **Characterization**

- Chemical Degradation (rates, mechanism)- FTIR, UV-vis, and XPS
- Surface Morphologies (AFM, SEM, EDXS)
- Release: amount & rate by ICP-OES

## **Nanoparticle Release Process and Collection**

Taber rotary abraser (ASTM D 4060-14, organic coatings)







**NIST SPHERE High Throughput, High Intensity UV Chamber** 









#### Abrasion parameters:

- MW2 metallic wheels
- Fixed loading
- 100 cycles

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### **Chemical Changes and Moss Loss**





Rates of chemical degradation and weathering-induced mass loss of commercial PU nanocoating (ENC) were lower than those of the neat PU, indicating that surface-treated silica nanoparticles had photostabilized the PU matrix.

### **Surface Morphological Changes**



 $\triangleright$  Silica nanoparticles were observed to accumulate and cluster on the nanocoating surface with increasing UV exposure time and eventually release from the nanocoating.

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### **Released Si mass collected from simulated rain test**



#### **Surface morphology & mass loss before and after abrasion**



#### **Surface morphology after abrasion – at different UV exposure times**



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- $\triangleright$  Silica nanoparticles were observed to accumulate and cluster on the nanocoating surface with increasing UV exposure time and eventually release from the nanocoating.
- $\triangleright$  The trends (as a function of exposure time) of released Si mass collected from simulated rain process and the mass loss & total surface particle counts from abrasion process are similar.

#### **Simulated rain process**

0

0.5

1

1.5

**Released Si Mass (µg)**

2

2.5

3







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### **Current Issues - for discussion**

*Concern: Harmful effects of surface-exposure and release* of nanomaterials during the life cycle of polymer nanocomposites?

- **How to capture released particles?**
- **Evidence of particle release – detection?** Can you detect discrete nanoparticles?
	- $\triangleright$  High resolution microscopy –SEM/TEM –labor intensive
	- $\triangleright$  ICP element analyses
	- **▶ Others**

#### • **The size and form of released particles?**

- $\triangleright$  Size: range from "nano" to "micro" depends on release mechanism
- $\triangleright$  Form: free nanoparticle? nanoparticles embedded in polymer matrix?
- Can we distinguish between agglomerates and aggregates of nanoparticles?
- **What are the best methods available to answer these questions? Reference?**
- Experimental data are needed for assessing and managing potential EHS risks of nanoparticles release during nanocomposites' life cycles.
- Need guidelines and protocols!