



Materials Science & Technology

# Ecological Exposure Review of State of Science

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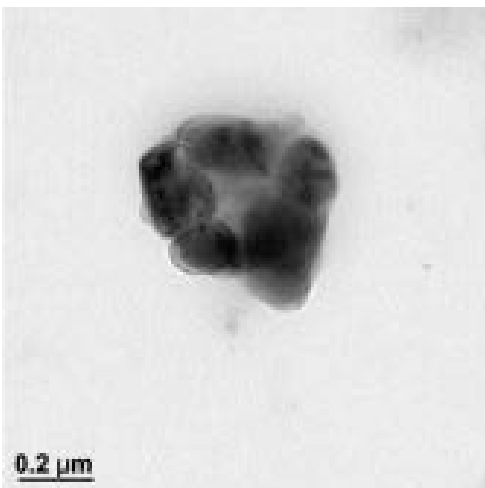
# Outline

- Life cycle approach
- Production and use
- Release
- Material Flow Modeling
- Ecological Exposure
- Environmental Risk Assessment

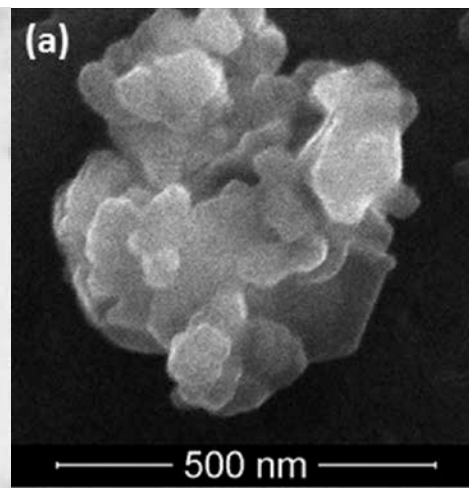
# Why not just measure exposure?

- Quantitative trace analysis of engineered nanomaterials in the environment is NOT yet possible
- Nanosized fraction  $\neq$  engineered NM
- Qualitative detection possible

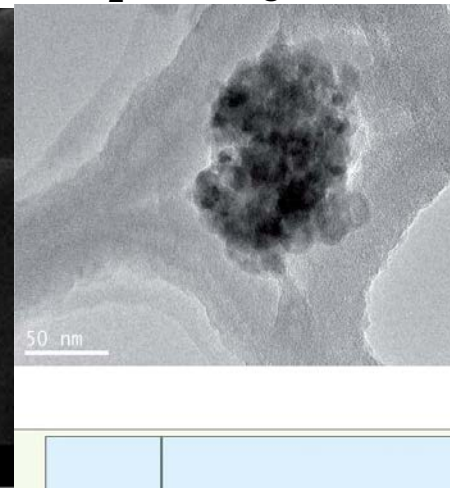
Kaegi (2008)  
TiO<sub>2</sub> in stream



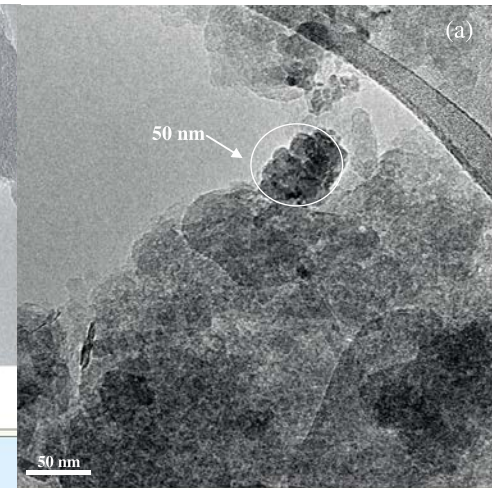
Gondikas (2014)  
TiO<sub>2</sub> in lake sediment



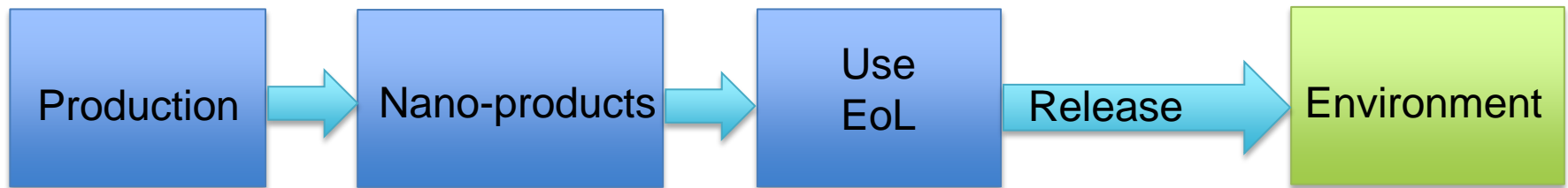
Westerhoff (2011)  
TiO<sub>2</sub> in sludge



Yang (2014)  
TiO<sub>2</sub> in soil



# The life cycle perspective on environmental exposure



## Information requirements

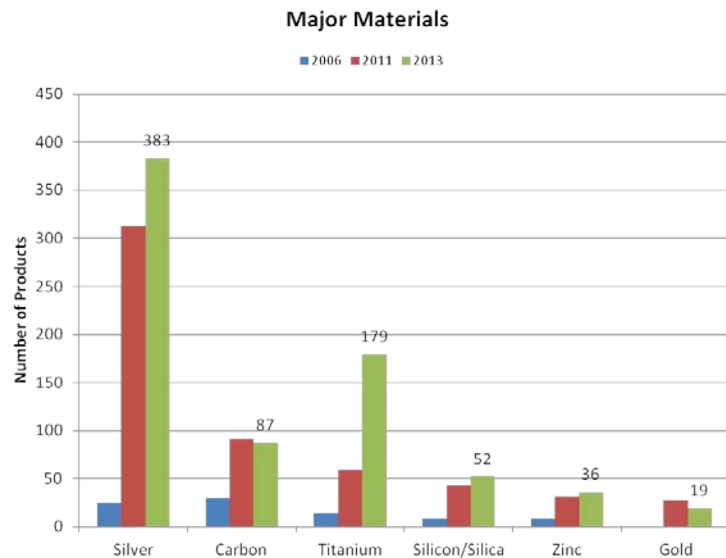
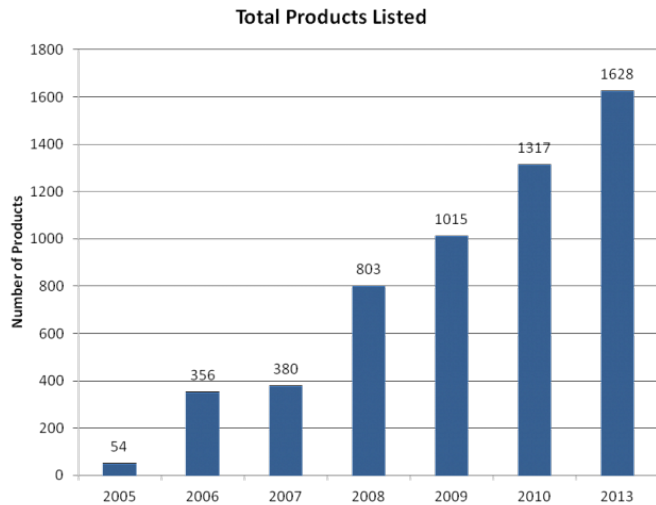
- Total amount
- Geographic distribution
- Relative share of product categories
- Life cycle as determinant
- Product type determines release potential
- Amount released
- Transformations
- Form released
- Real-world release
- Fate processes
- Geographic distribution
- Natural NM

# Production/ consumption, extrapolated to the EU (in tons/y)

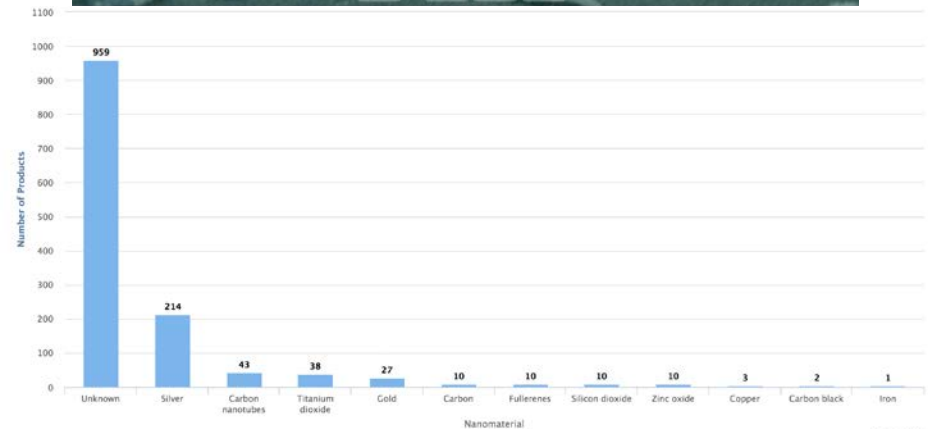
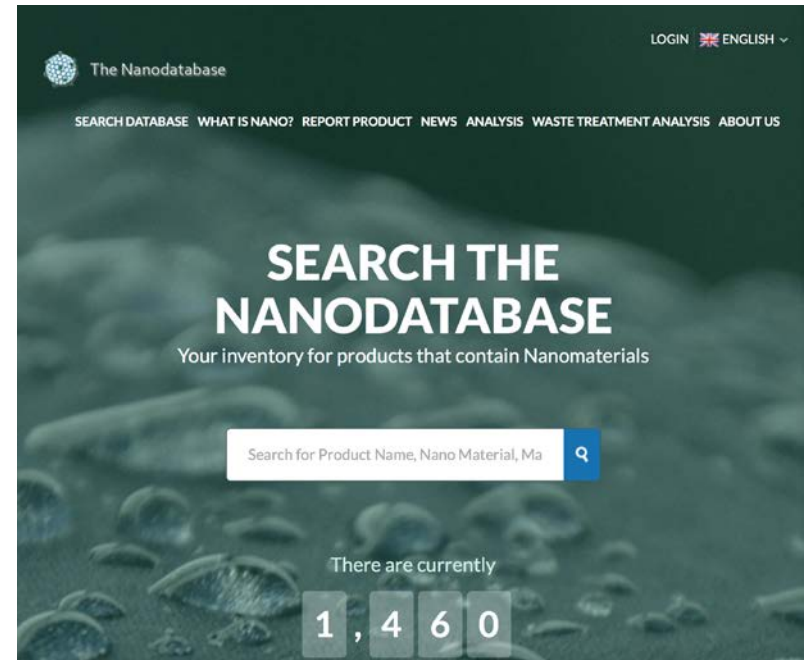
ENM	(Schmid and Riediker, 2008)	(Hendren et al., 2011)	(Piccinno et al., 2012)	(Keller et al., 2013)	(ANSES, 2013)	Sun et al., 2014
TiO <sub>2</sub>	11'500	8'600-42'000	550	20'000	92'000	10,000
Ag	82	3-20	6	100	0.006	30
ZnO	1,900	-	55	7,900	1,900	1,600
CNT	26	60-1,200	550	740	-	380
C <sub>60</sub>	-	2-90	0.6	-	< 100	20
CeO <sub>2</sub>	-	40-770	55	2,300	700	-
Al-ox	0.1	-	550	8,100	15,000	-
Fe-ox	9,700	-	550	9,700	6,100	-
SiO <sub>2</sub>	2,000	-	5500	22,000	990,000	-
Nanoclays	-	-	-	2,400	<100	-
Cu	-	-	-	46	< 100	-
Quantum dots	-	-	0.6	-	-	-

# Product databases

## Woodrow Wilson Database



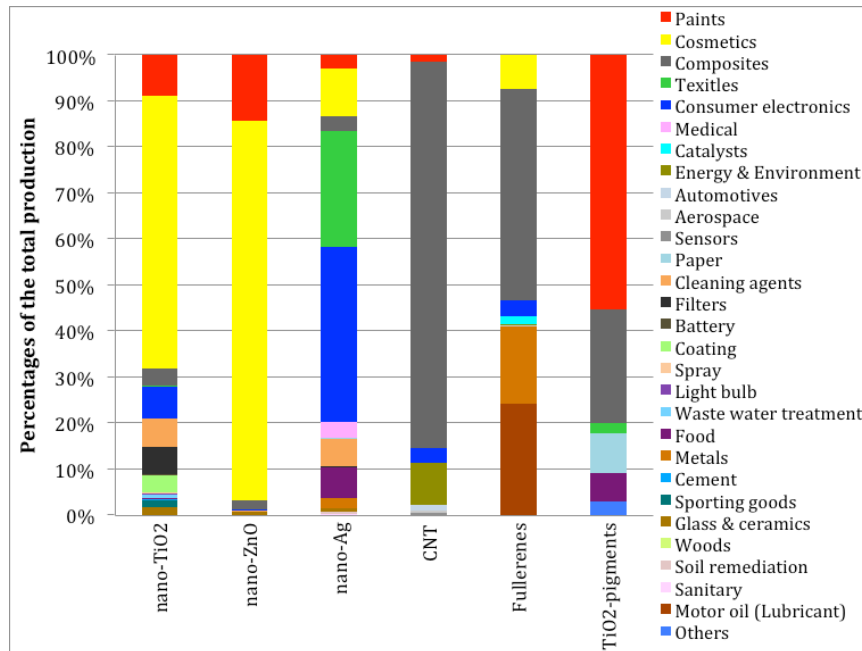
## Danish Nano-Database



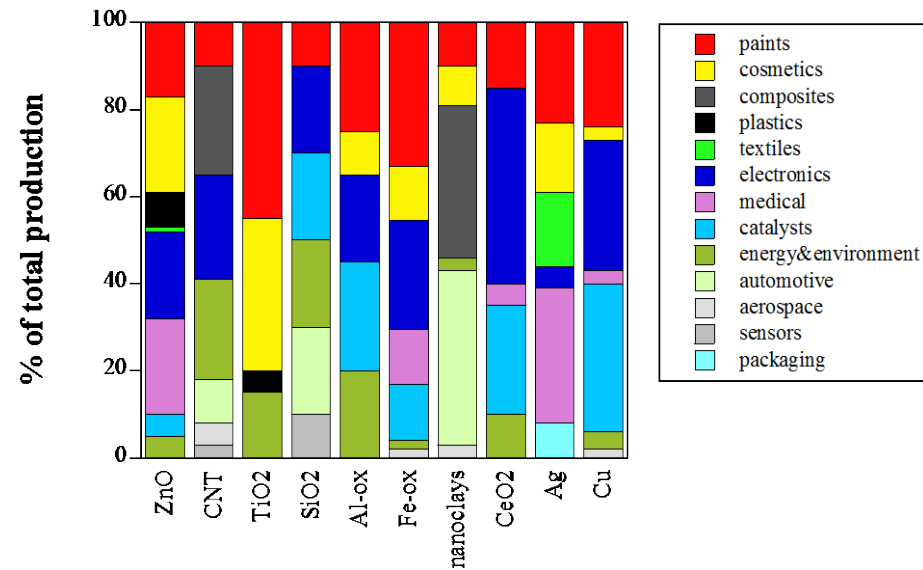
# Survey of industry: Uses of nanomaterials

Nanomaterial	Product Group	% of total use
<b>Nano-TiO<sub>2</sub></b>	<i>Cosmetics (incl. sunscreens)</i>	70 – 80
	<i>Coatings &amp; cleaning agents</i>	< 20
	<i>Plastics</i>	< 20
	<i>Paints</i>	10 – 30
	<i>Cement</i>	1
	<i>Others</i>	< 10
<b>Nano-ZnO</b>	<i>Cosmetics (incl. sunscreens)</i>	70
	<i>Paints</i>	30
<b>CNTs</b>	<i>Composites &amp; polymer additives</i>	20
	<i>Materials</i>	80
	<i>Composites</i>	50
	<i>Batteries</i>	50
<b>Fullerenes</b>	<i>R&amp;D</i>	80
<b>Nano-Ag</b>	<i>Paints, coatings &amp; cleaning agents</i>	10 – 30
	<i>Textiles</i>	30 – 50
	<i>Consumer electronics &amp; conductivity</i>	10 – 20
	<i>Cosmetics</i>	20
	<i>Medtech</i>	20
	<i>Anti-microbial coatings</i>	80 – 100

# Product distribution for nano-product categories



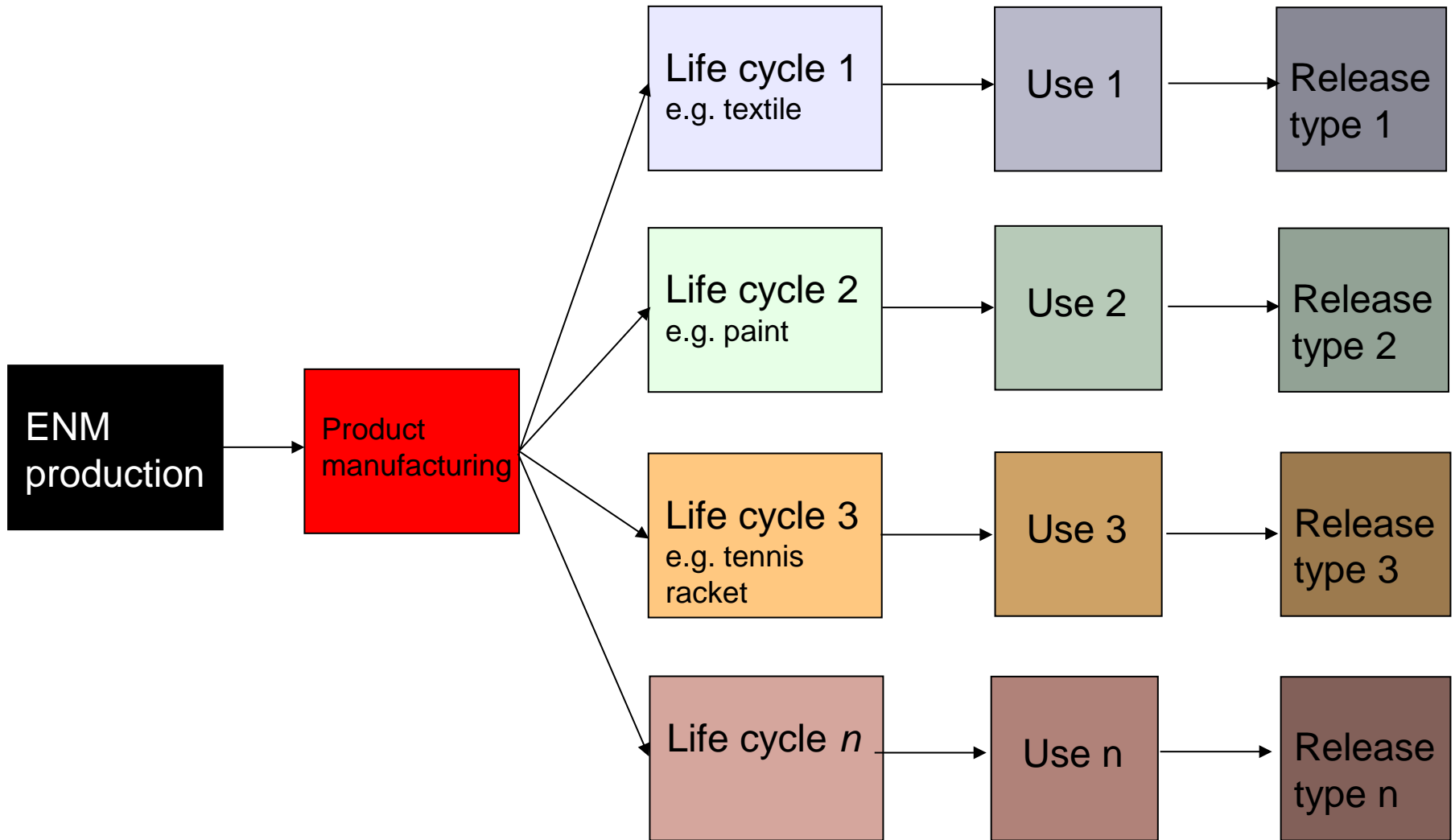
Sun et al., (2014) Environ. Pollut. 185: 69-76



Keller et al., (2013) J. Nanopart. Res. 15:1-17



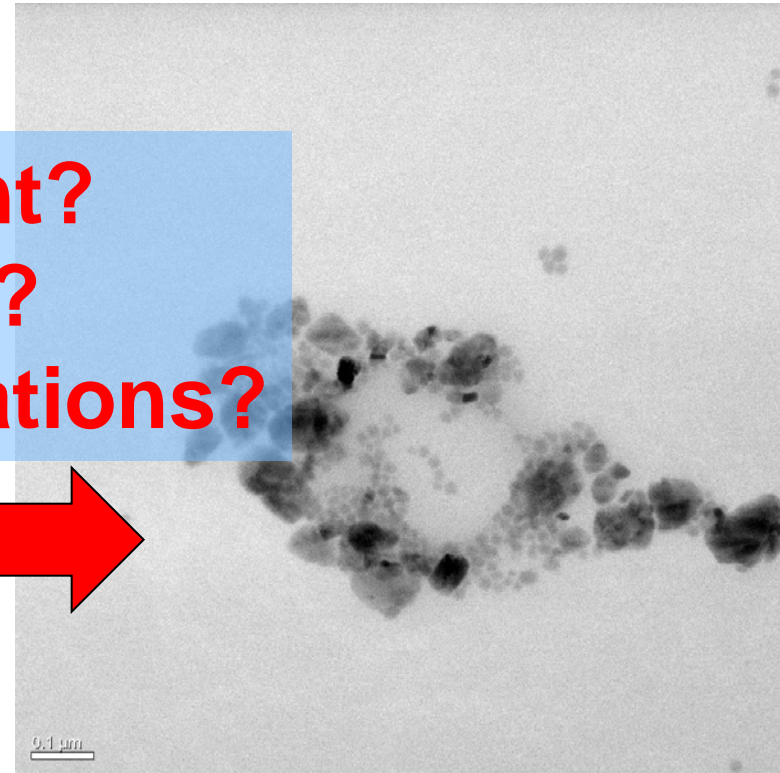
# Why care about the life cycle?



# Release determines exposure



**Amount?  
Form?  
Transformations?**



# Release during accidents



<http://www.lalsace.fr/actualite/2011/10/10/vieux-thann-des-sacs-d-oxyde-de-titane-tombent-d-un-camion-sur-la-rn66#jimage=1446B4B6-C4F4-42B8-8F20-052BEA52339E>

ALSACE



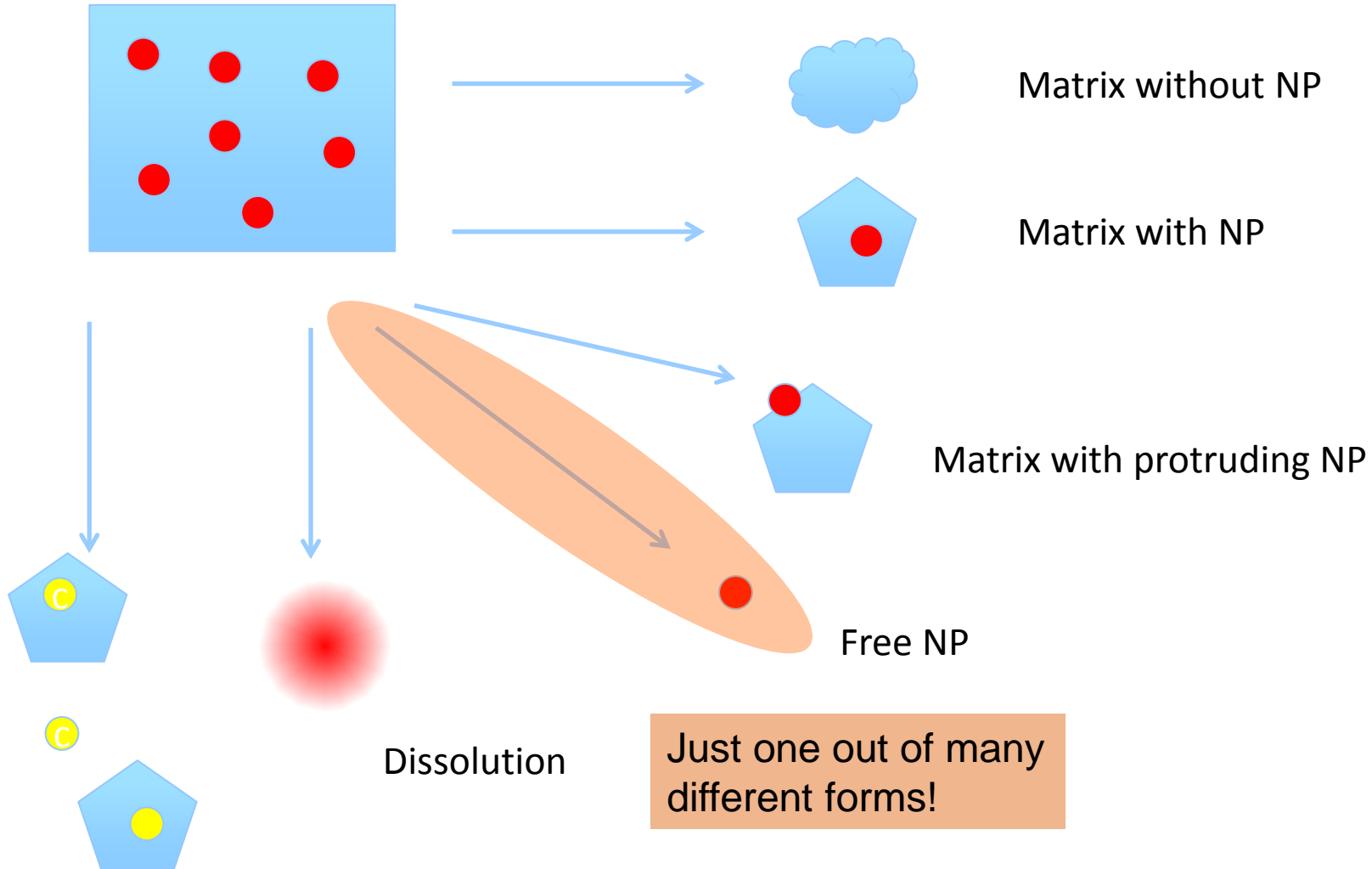
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# Release during accidents

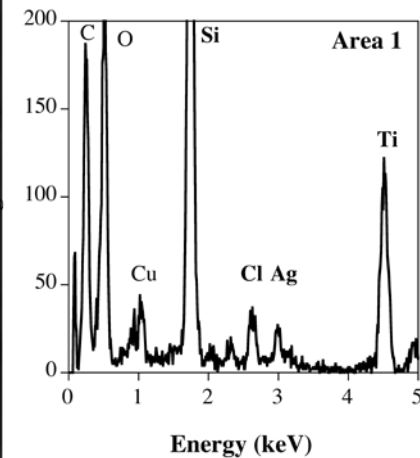
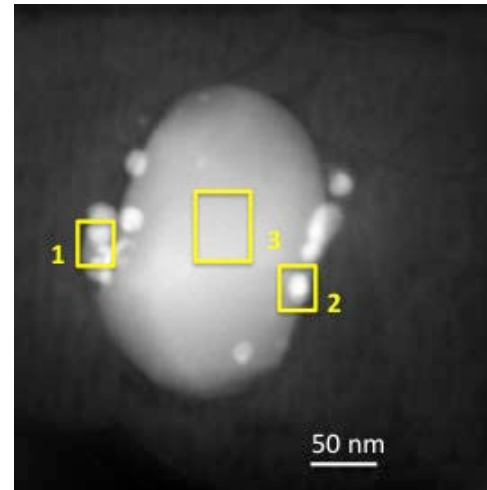
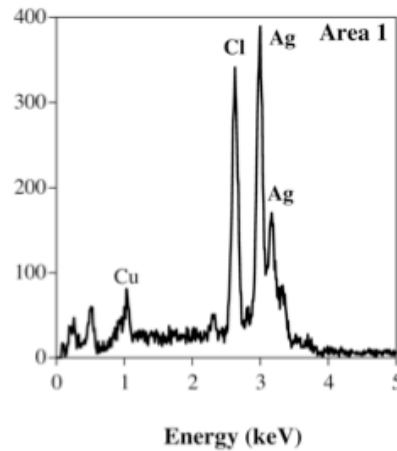
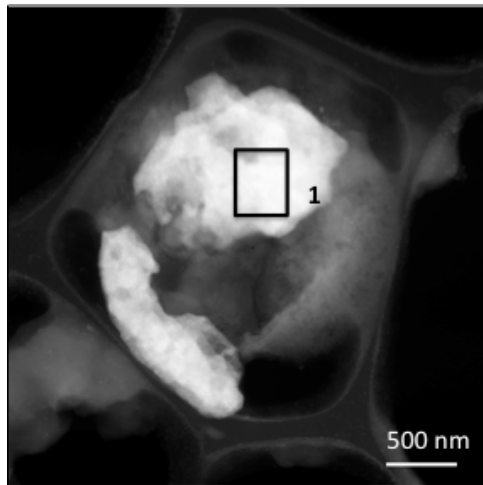
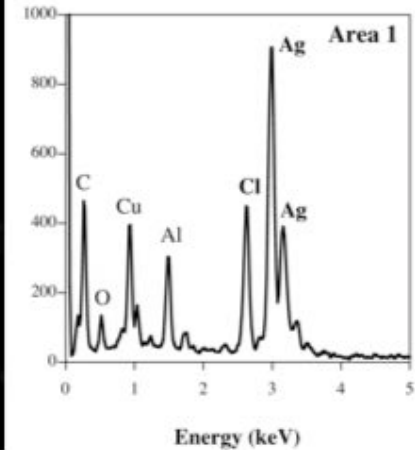
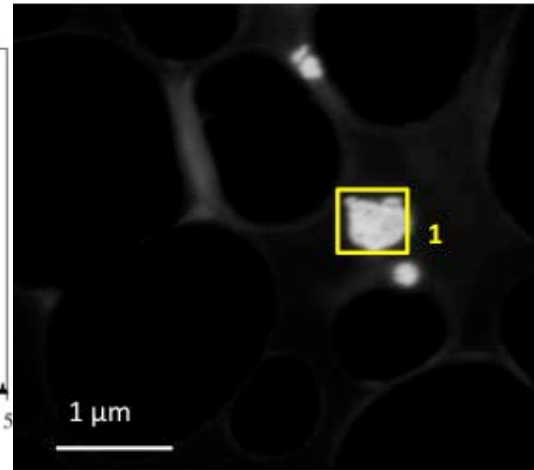
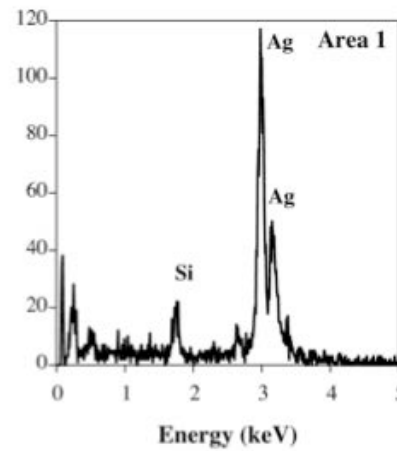
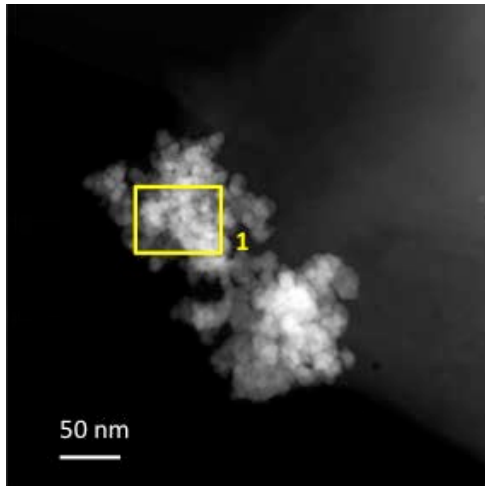


<http://www.lalsace.fr/actualite/2011/10/10/vieux-thann-des-sacs-d-oxyde-de-titane-tombent-d-un-camion-sur-la-rn66#jimage=1446B4B6-C4F4-42B8-8F20-052BEA52339E>

# Characterization of released materials



# Characterization of released Ag

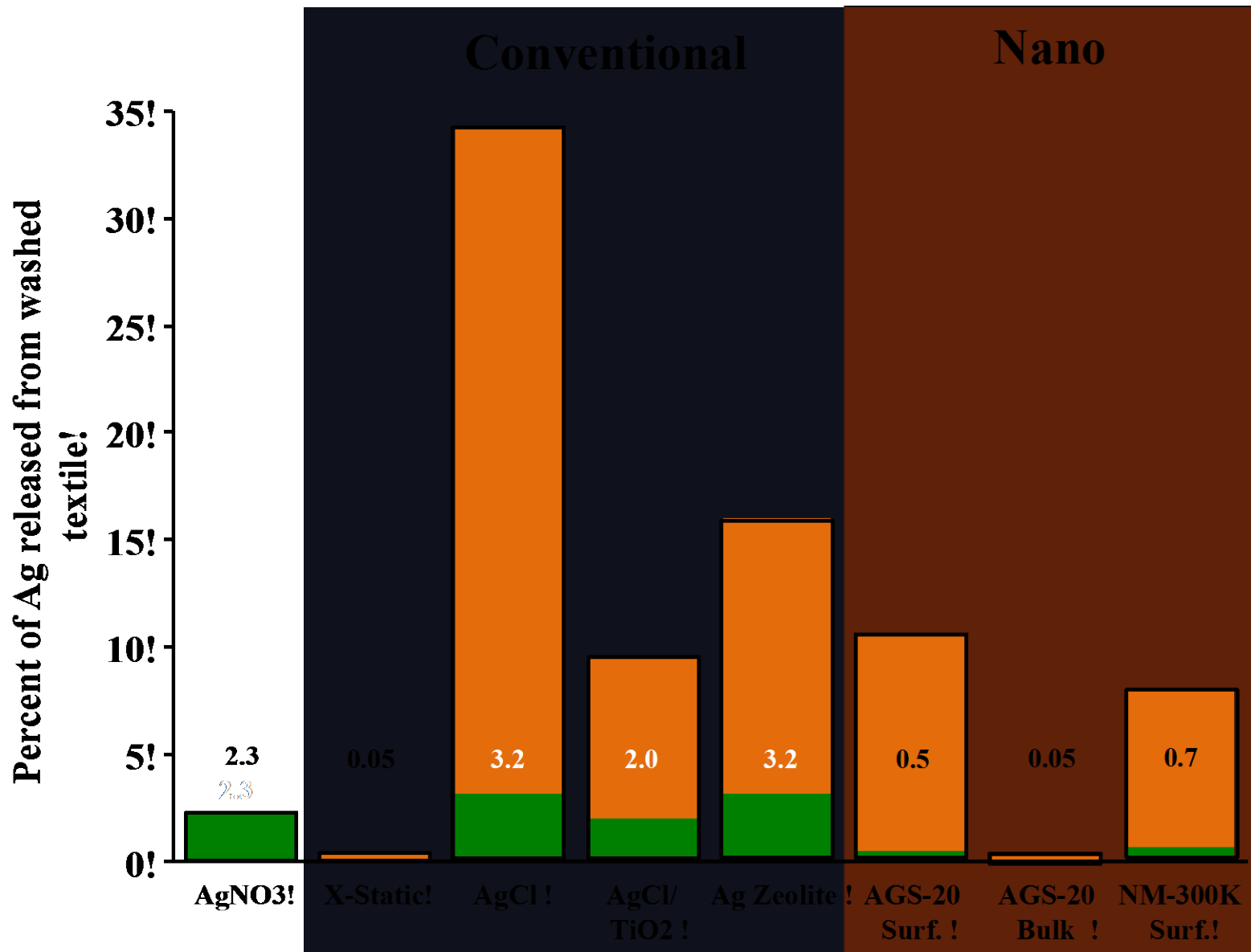


# Ag speciation of original and washed textile

Ag integrated in fiber  
 Nano-Ag  
 Ag ions  
 Nano-Ag  
 AgCl

Textile	Ag-NPs	AgCl-NPs	Ag <sub>2</sub> S-NPs	Ag oxide	Ag phos.	Ag nitr.	Ag sulf.	Ag-zeolite	R factor
<i>Unwashed</i>									
T1			16 (1.6)			52 (0.7)	14 (0.7)	18 (1.3)	0.0005
T4	55 (1.7)	11 (3)	32 (5)						0.0001
T5	43 (1.5)	17 (3)	35 (3)	5 (0.8)					0.0001
T6	12 (3)		16 (5)		67 (6)			5 (0.5)	
T7	36 (1.7)	36 (3)	14 (2)				14 (1.7)		0.0005

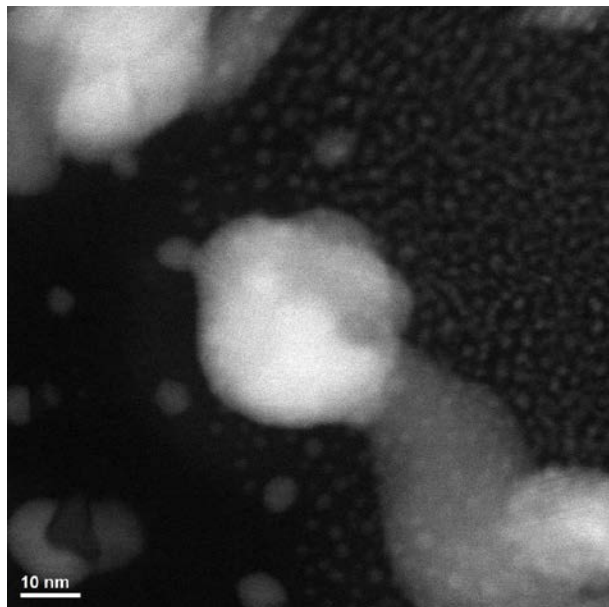
# Nanomaterial Release from Textiles



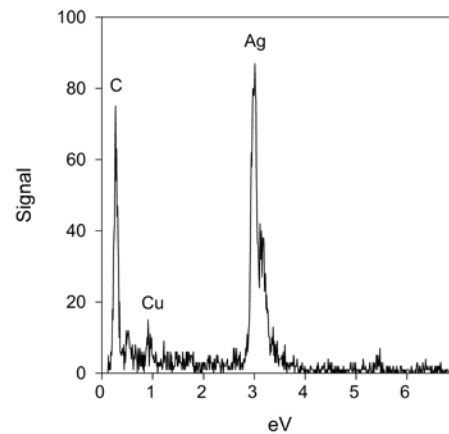
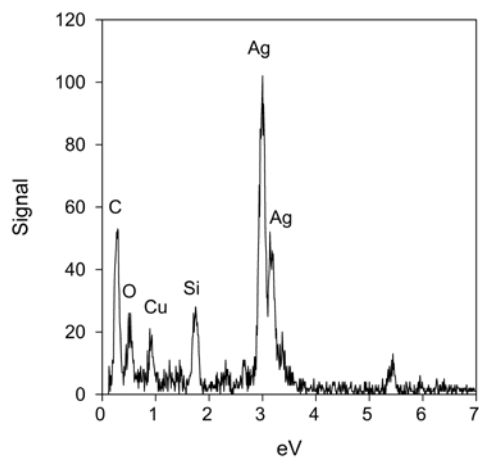
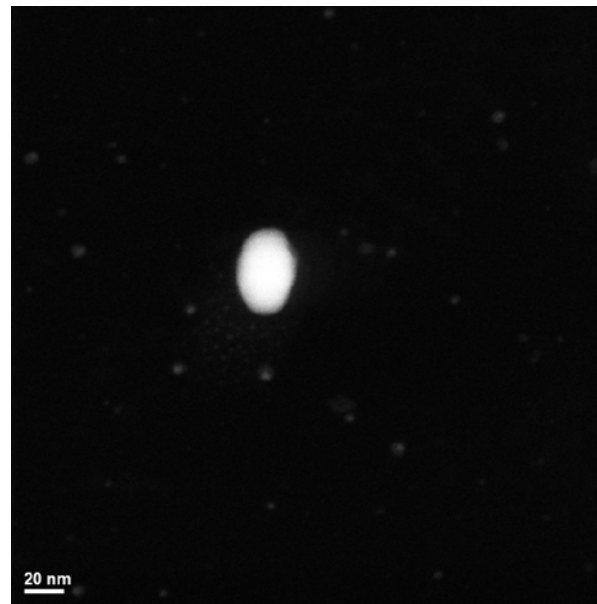


# Formation of nano-Ag

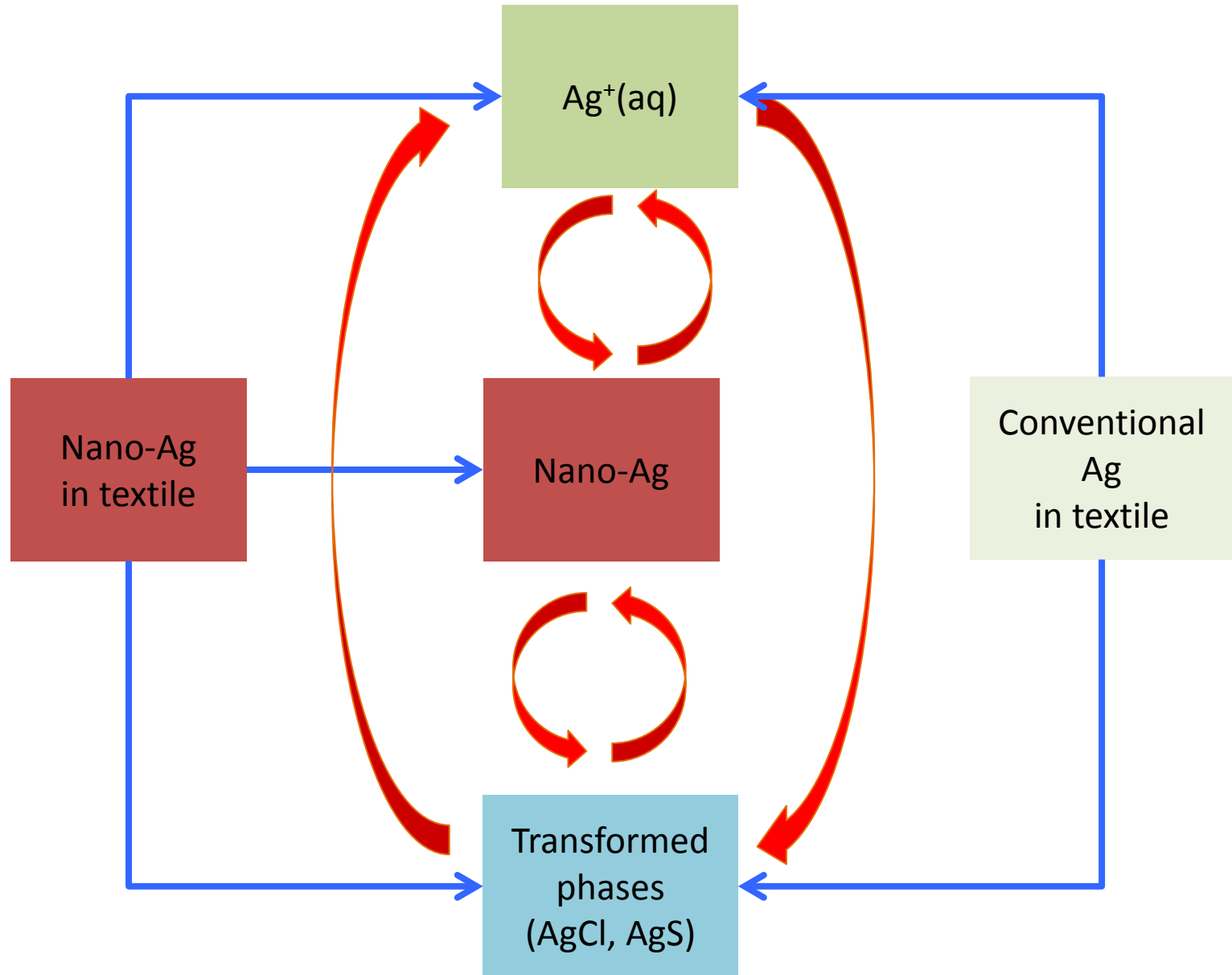
## Dissolved Ag



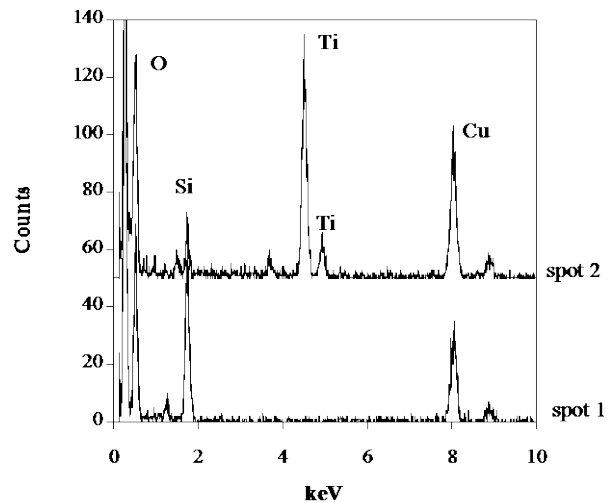
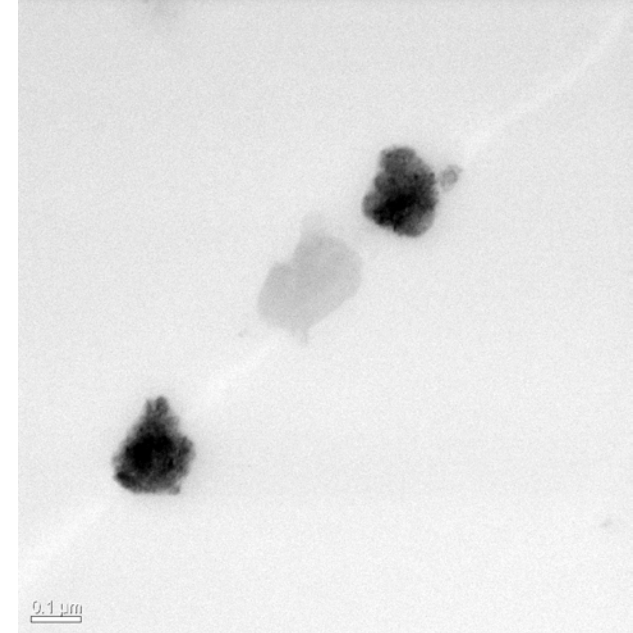
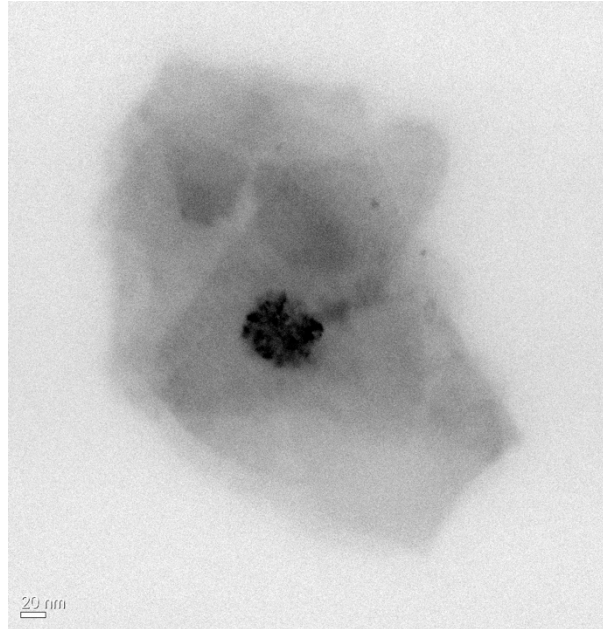
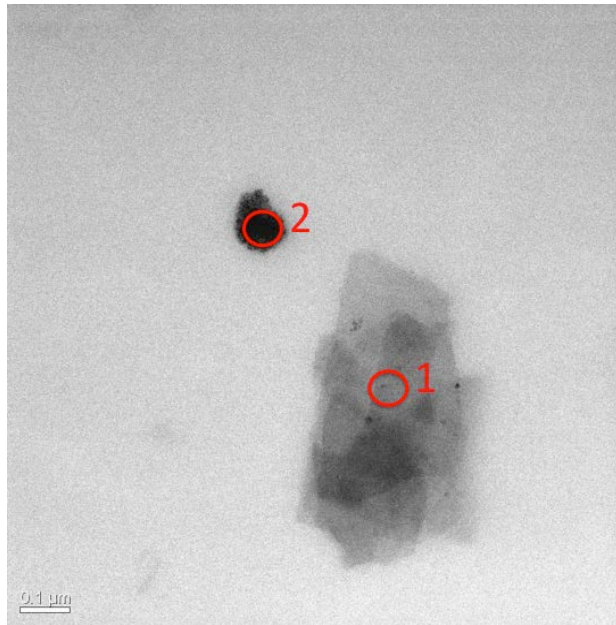
## Ag zeolites



# Transformations of Ag released from textiles

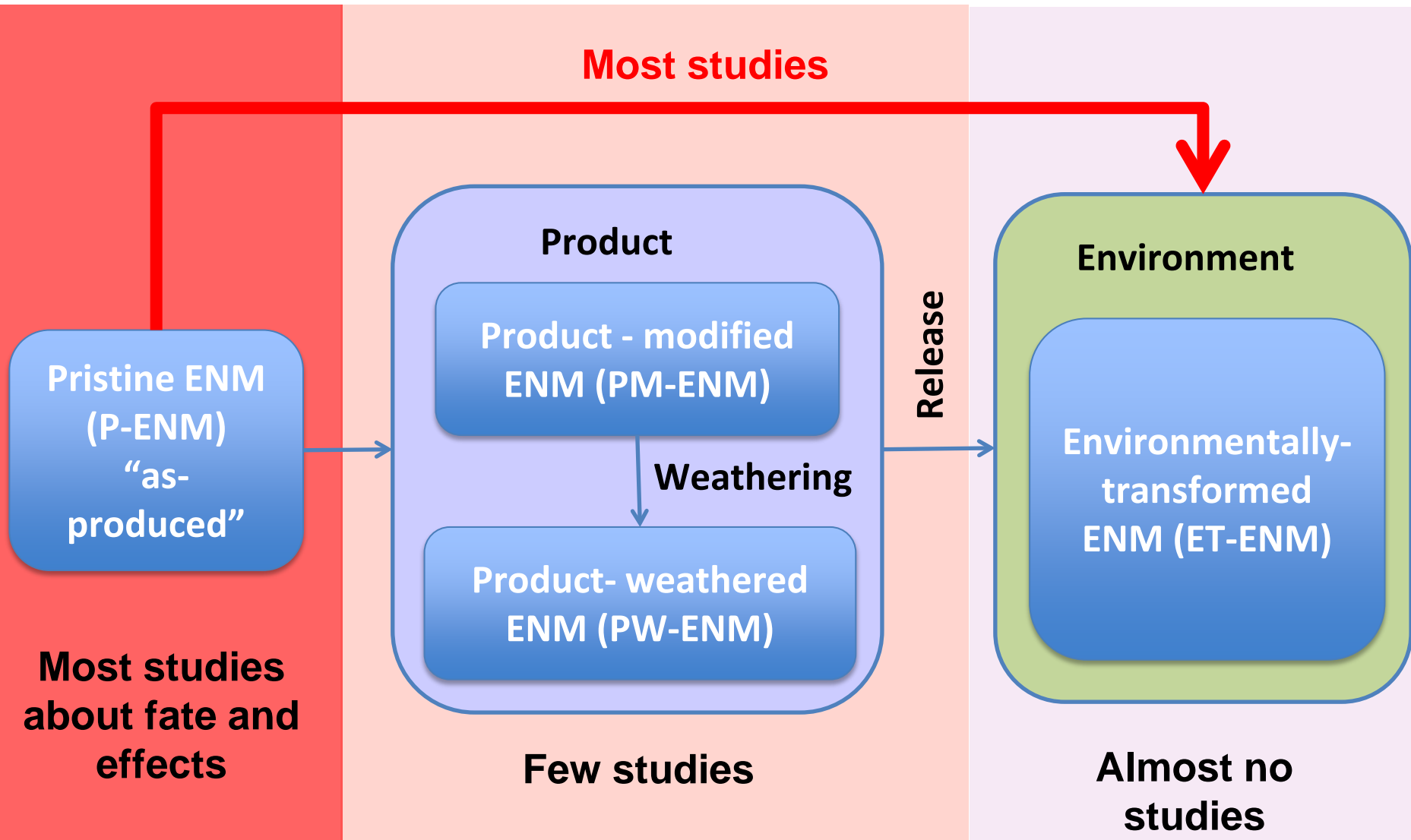


# Characterization of released $\text{TiO}_2$



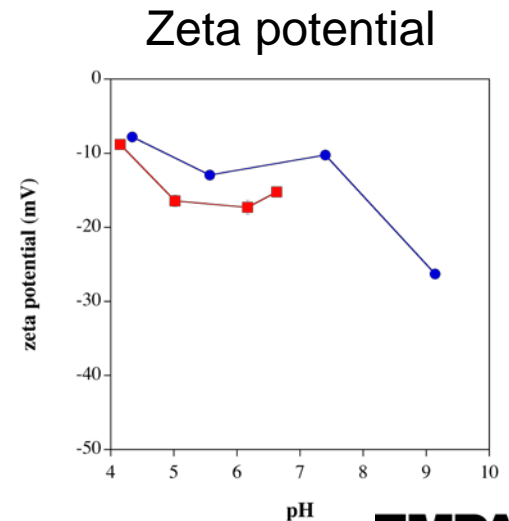
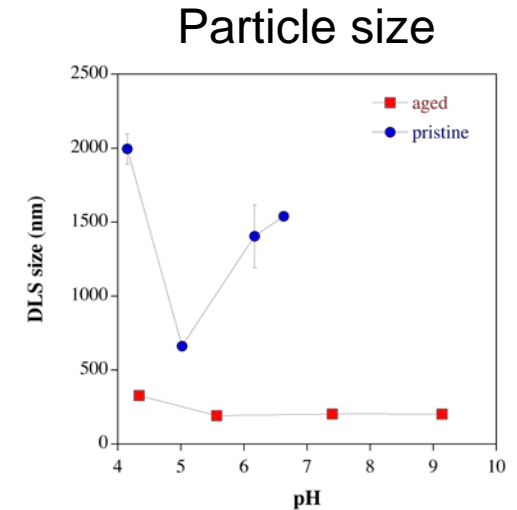
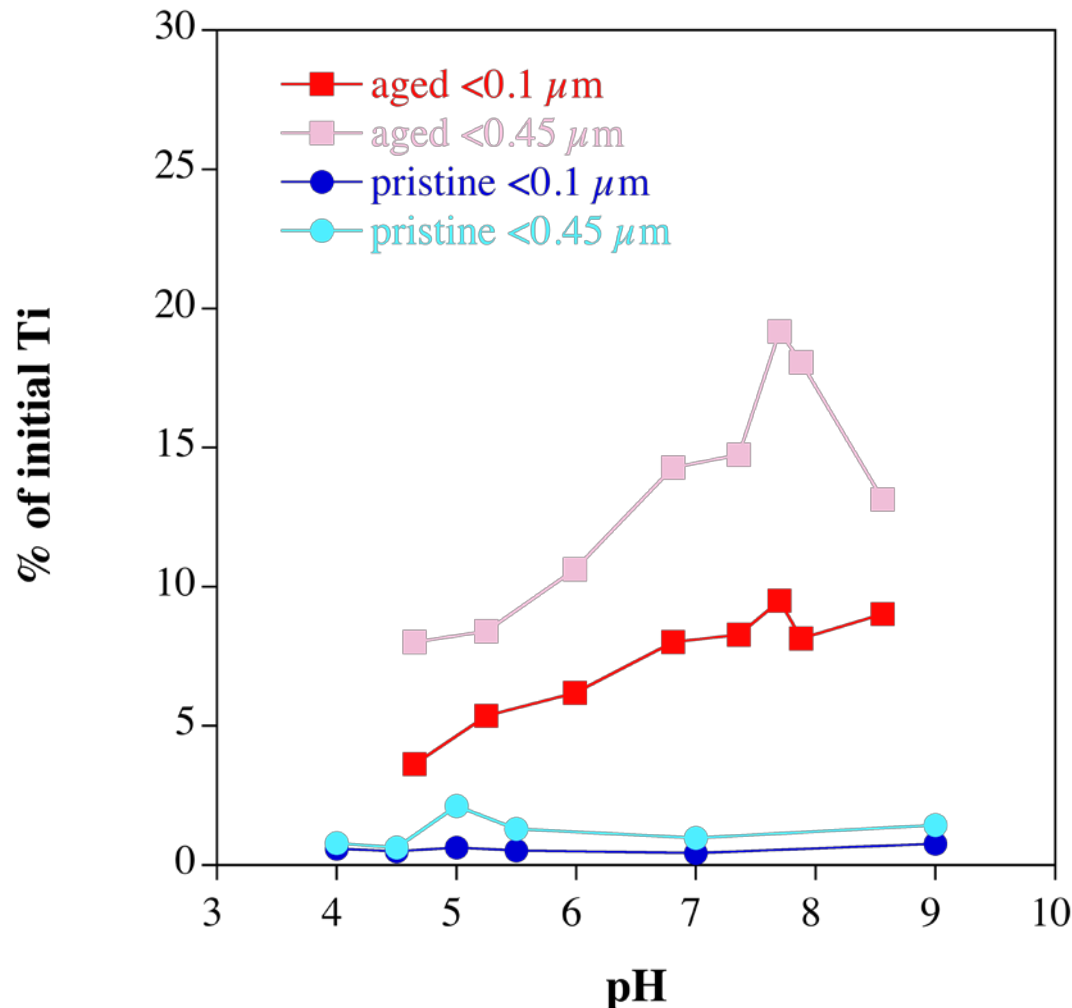
Al-Kattan et al., (2014) *Environ. Sci. Technol.*, 48: 6710–6718

# Release and transformation

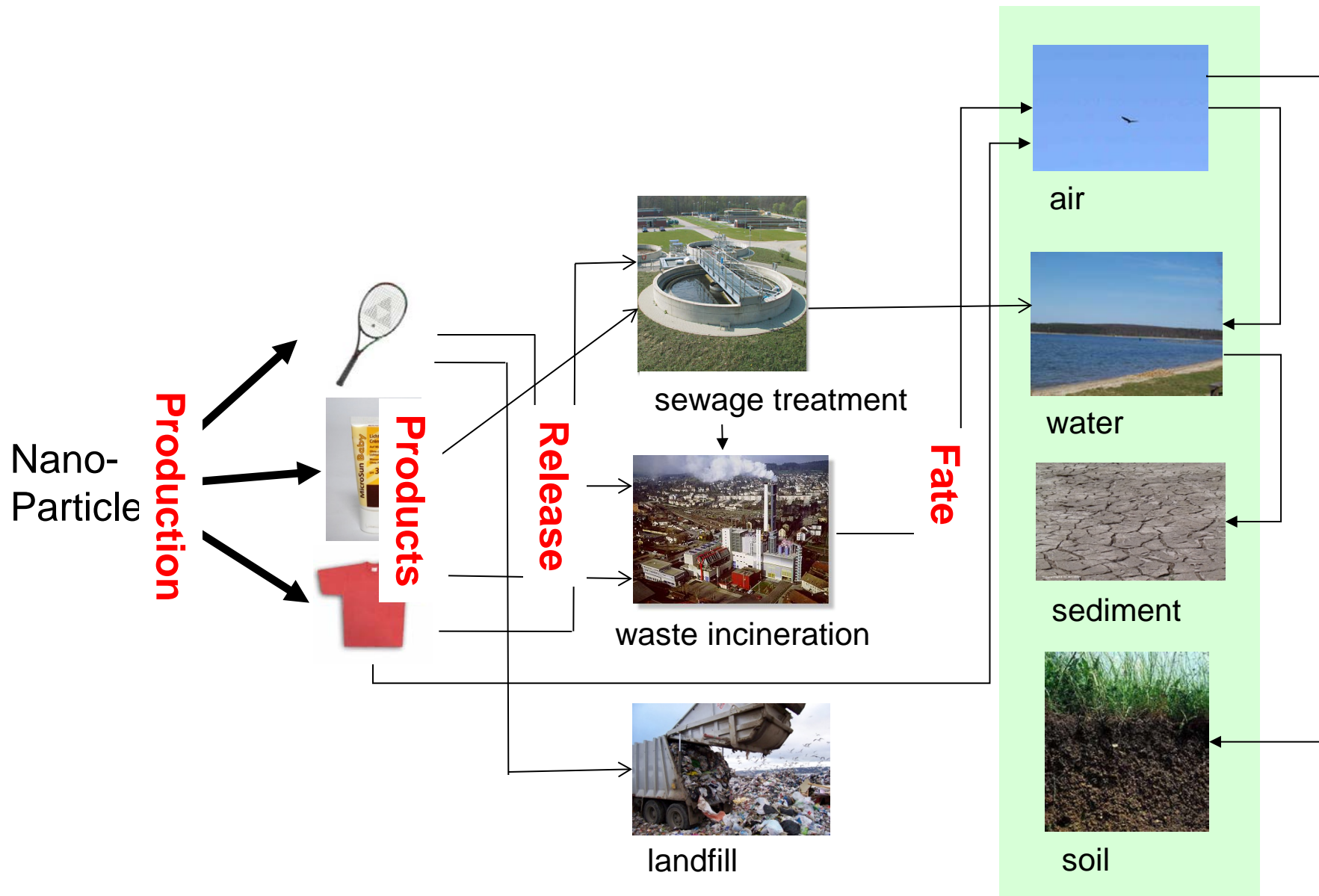


# Environmental behavior of released TiO<sub>2</sub>

Stability of released TiO<sub>2</sub> in 1 mM Ca, compared to pristine TiO<sub>2</sub>

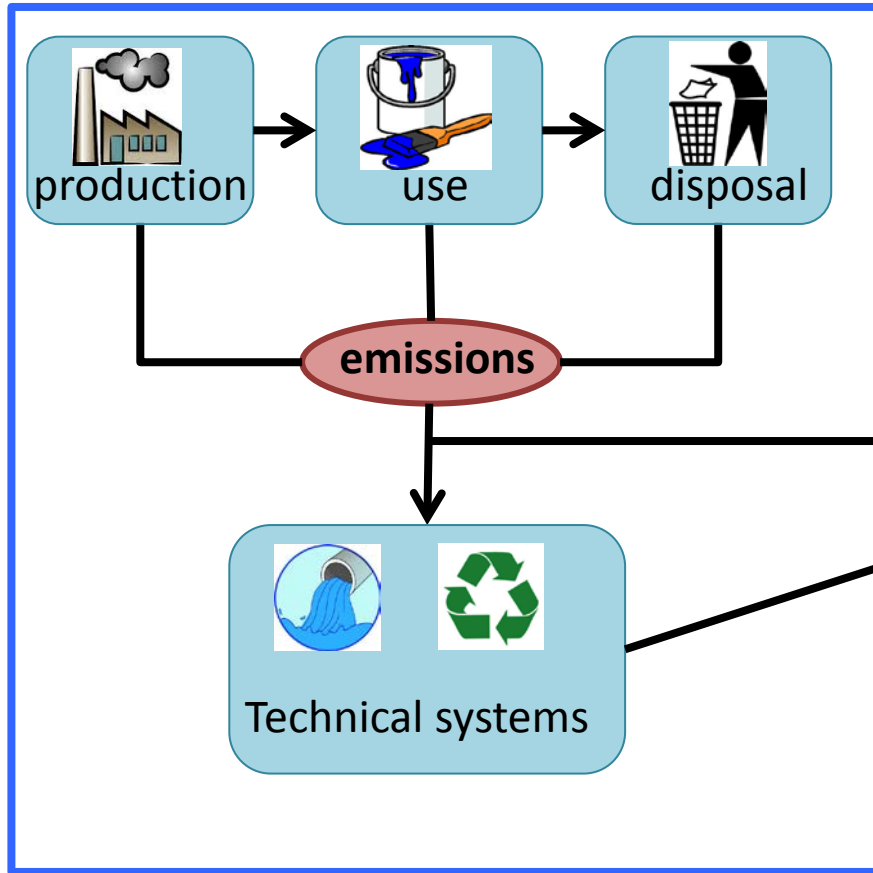


# Modeling environmental concentrations

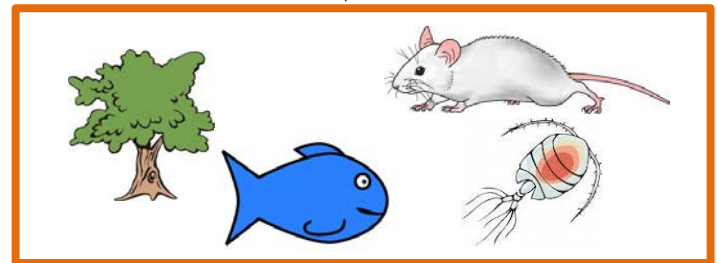
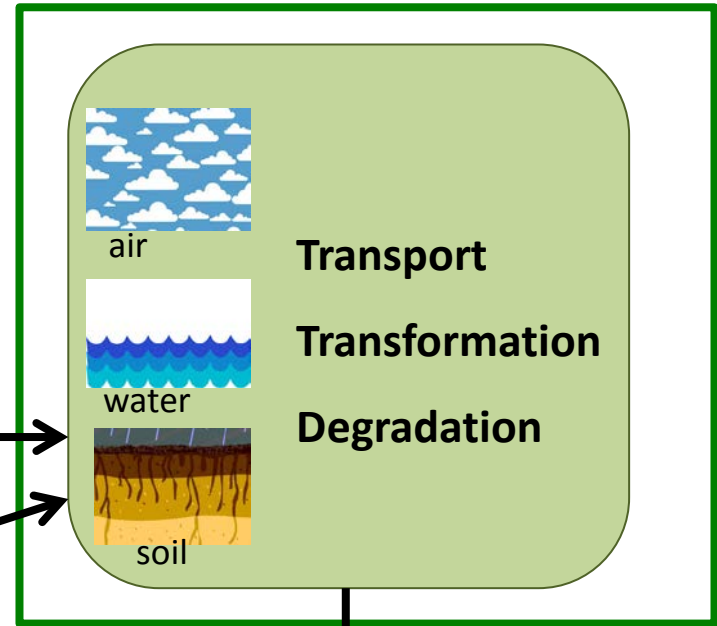


# General concept of exposure models

## Material Flow Modeling (MFA)

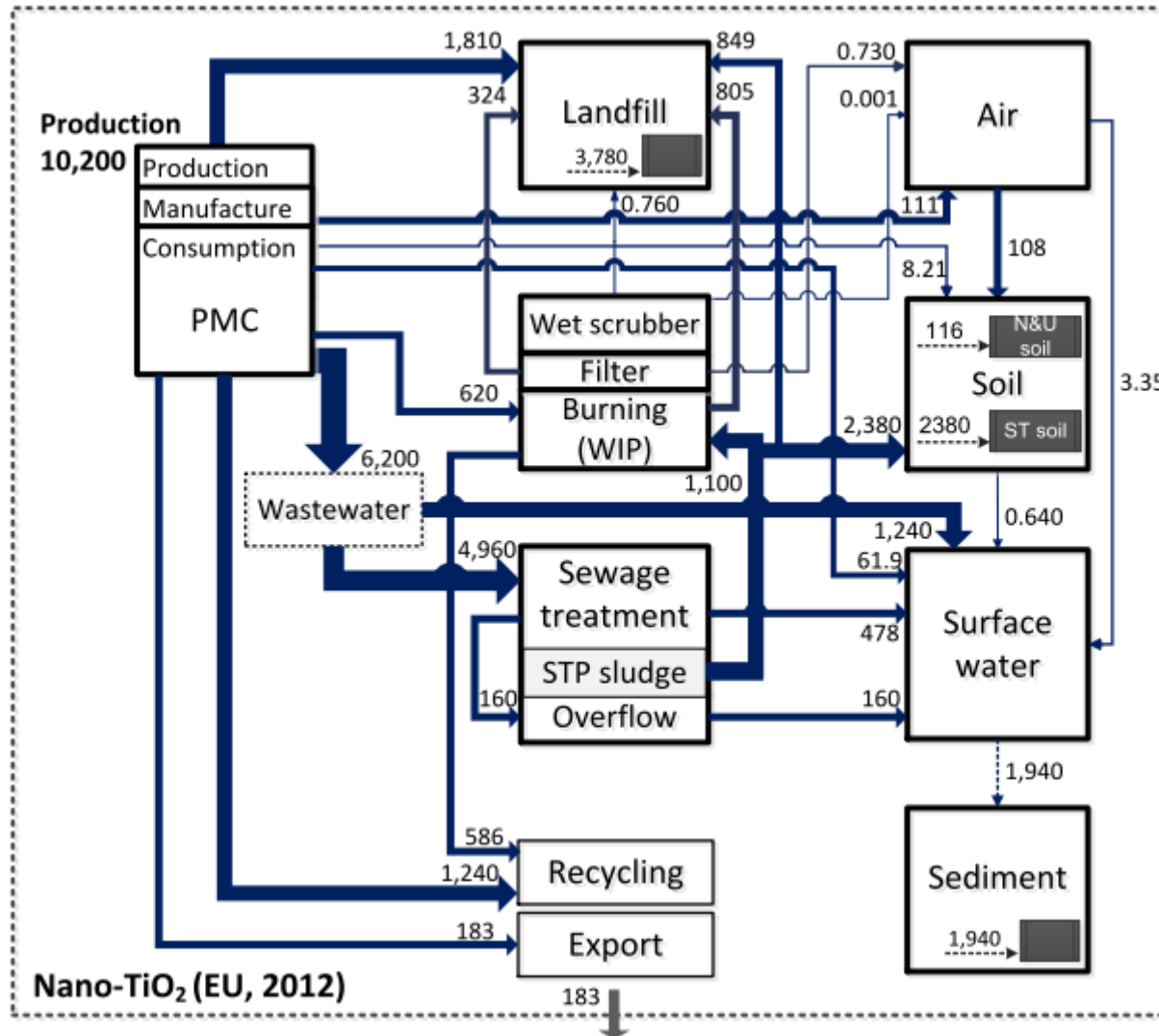


## Environmental Fate Modeling (EFM)



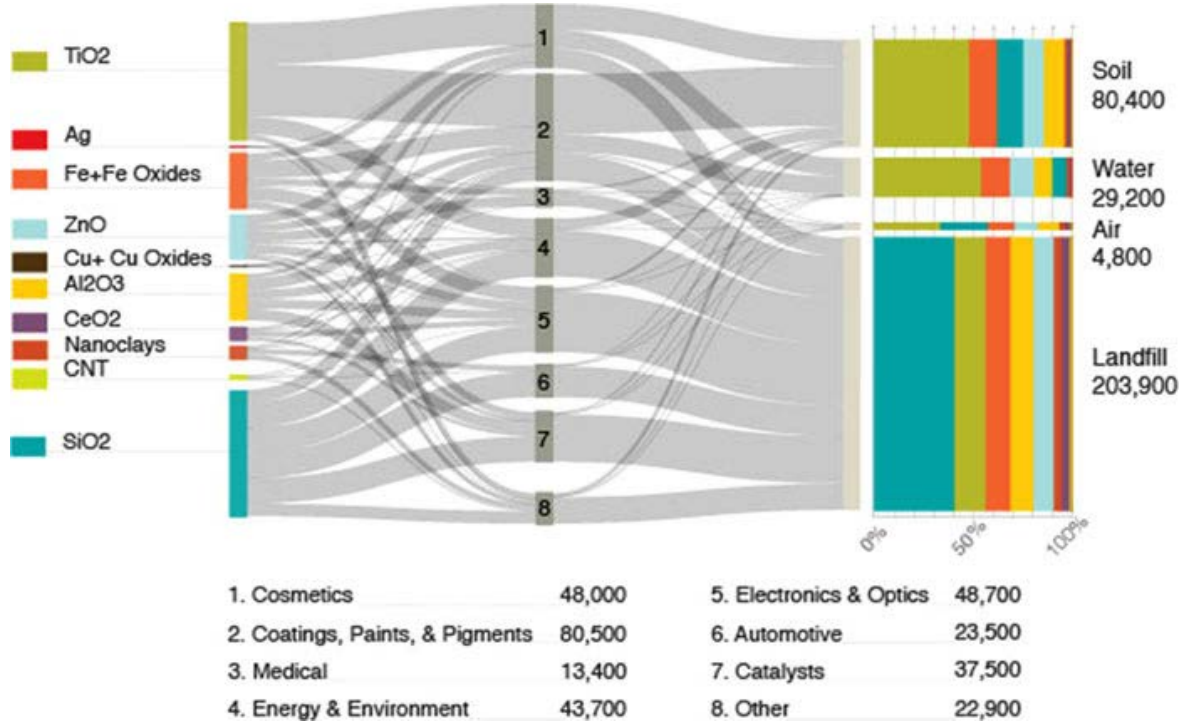
## Effect modeling

# Material-flow model for nano-TiO<sub>2</sub> in the EU (mode values in tons/year)





# Global Flows



- Source: one market research report
- No uncertainty considered
- Very simple transfer factors
- No transformation

TiO<sub>2</sub>

# From flows to concentrations

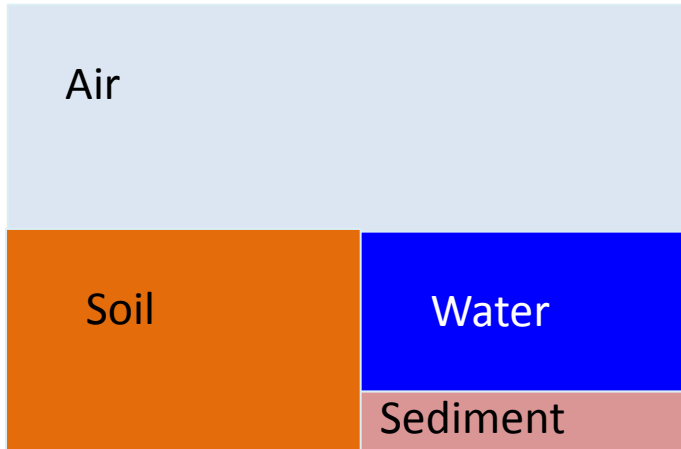


Table R.16-12: Proposed model parameters for regional model

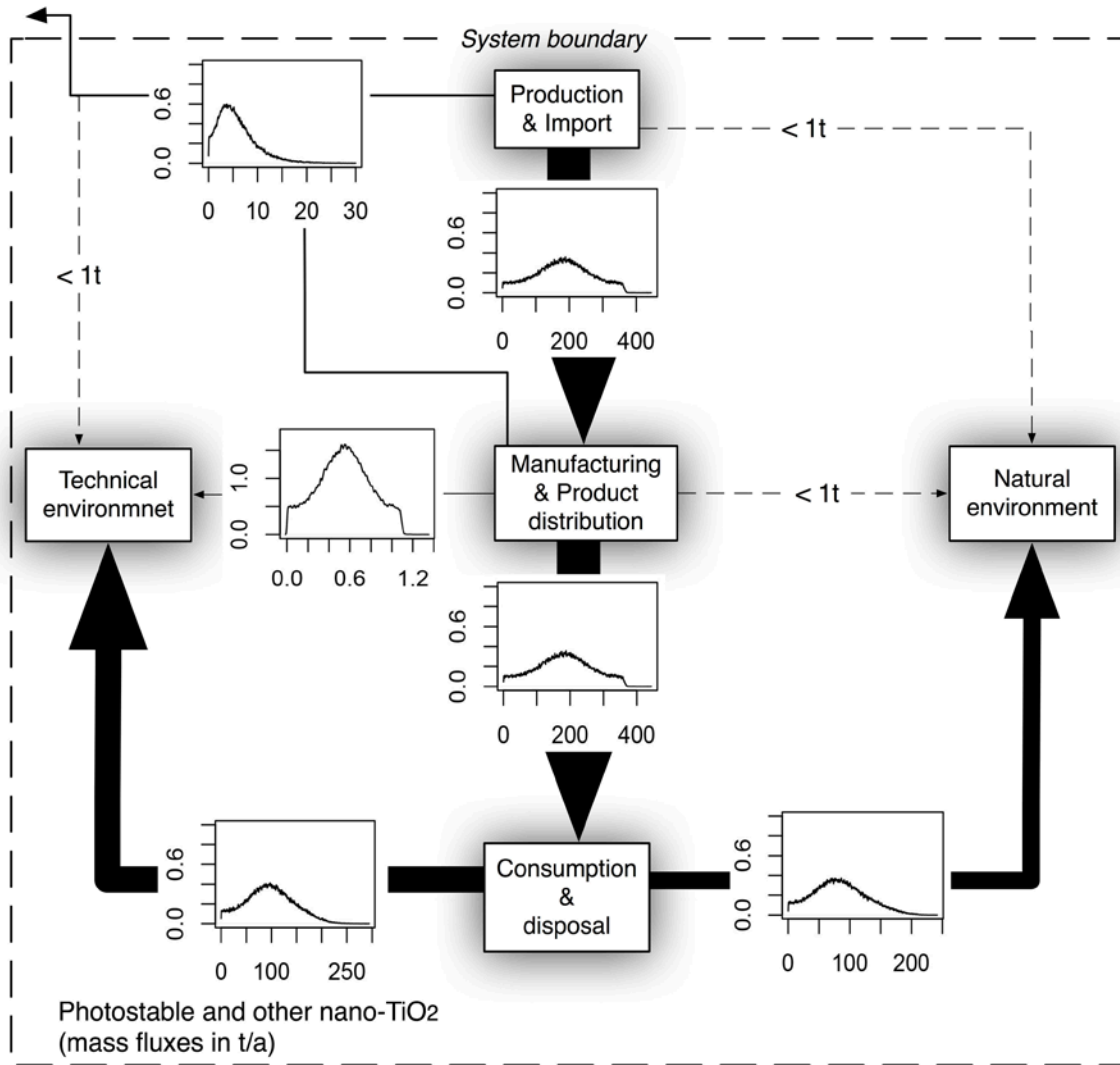
- Well mixed compartments
- Standard volumes assumed
- Source: REACH guidance

# Environmental exposure

## Example nano-TiO<sub>2</sub>

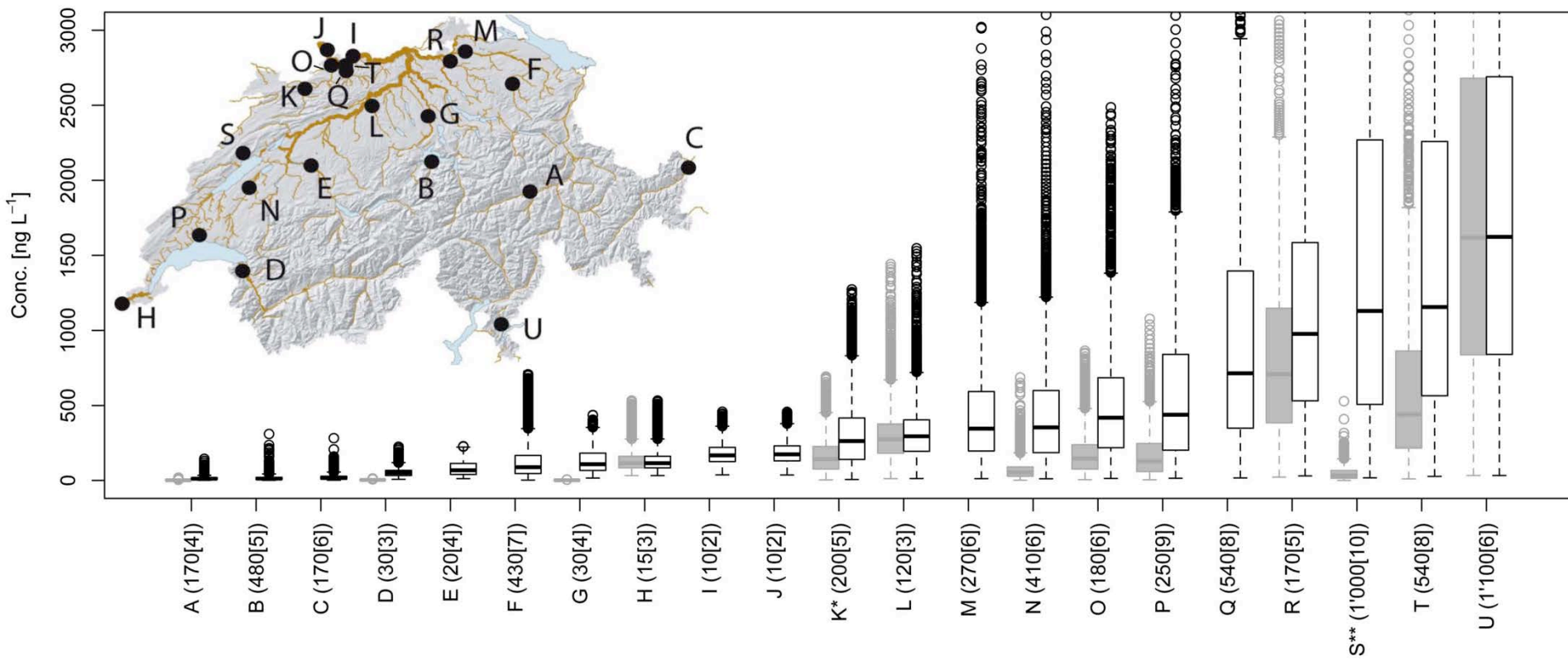
	EU			
	Mode	Q <sub>0.15</sub>	Q <sub>0.85</sub>	
<b>STP Effluent</b>	16	13	110	µg/L
<b>Surface water</b>	0.53	0.40	1.4	µg/L
<b>Sediment</b>	1.9	1.4	4.8	mg/kg·y
<b>STP sludge</b>	170	150	540	mg/kg
<b>Natural and urban soil</b>	0.13	0.09	0.24	µg/kg·y
<b>Sludge treated soil</b>	1200	940	3600	µg/kg·y
<b>Air</b>	0.001	0.000	0.001	µg/m <sup>3</sup>
<b>Solid waste</b>	12	8.3	20	mg/kg
<b>WIP bottom ash</b>	120	82	230	mg/kg
<b>WIP fly ash</b>	150	110	310	mg/kg

# Probabilistic modeling

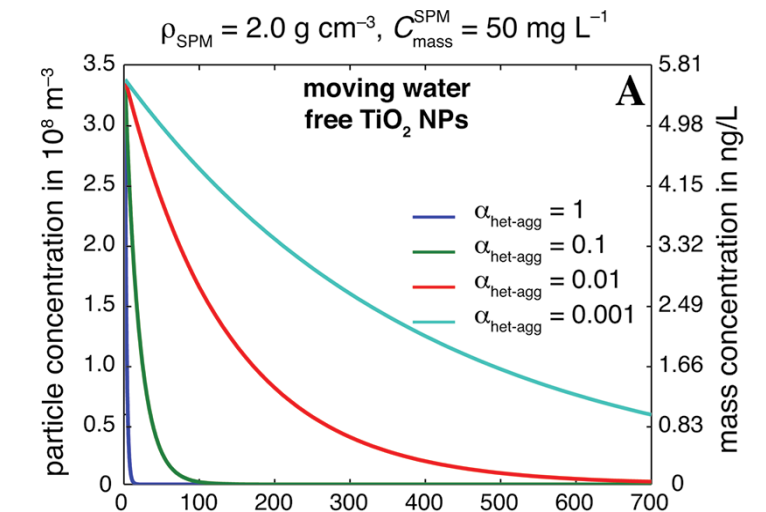
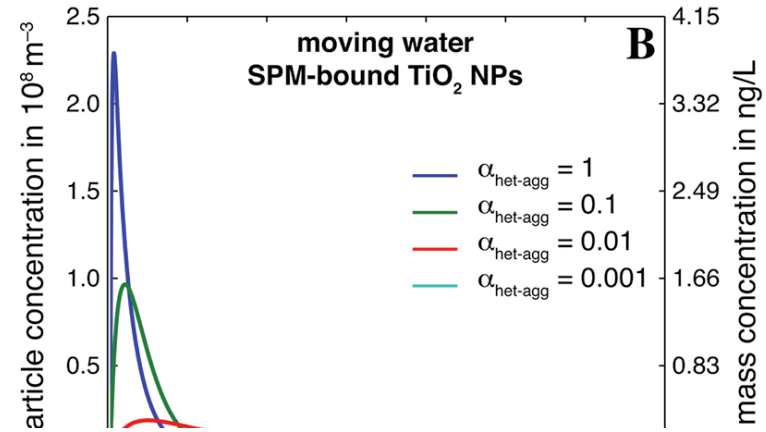
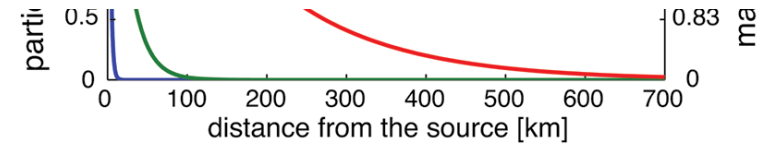
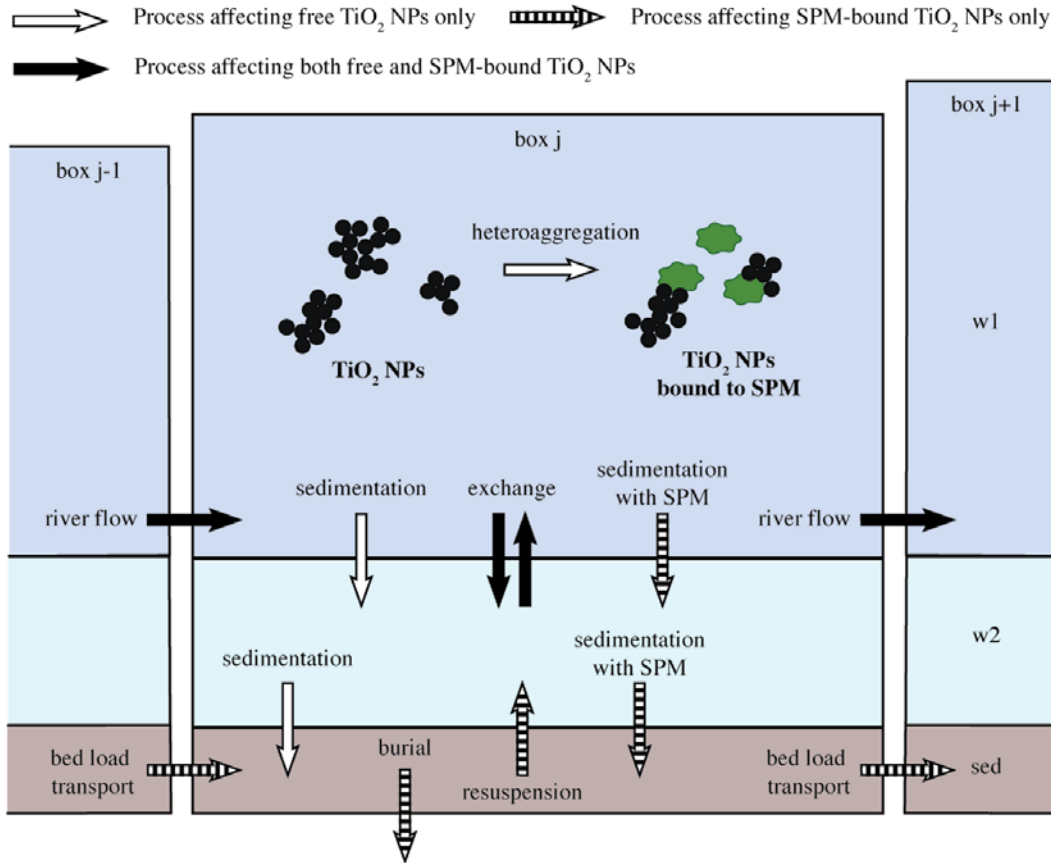


# Geographic variability

## Nano-TiO<sub>2</sub> based on 20-year flow data

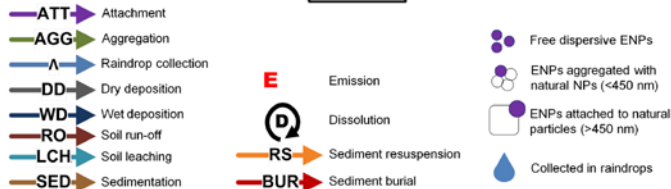
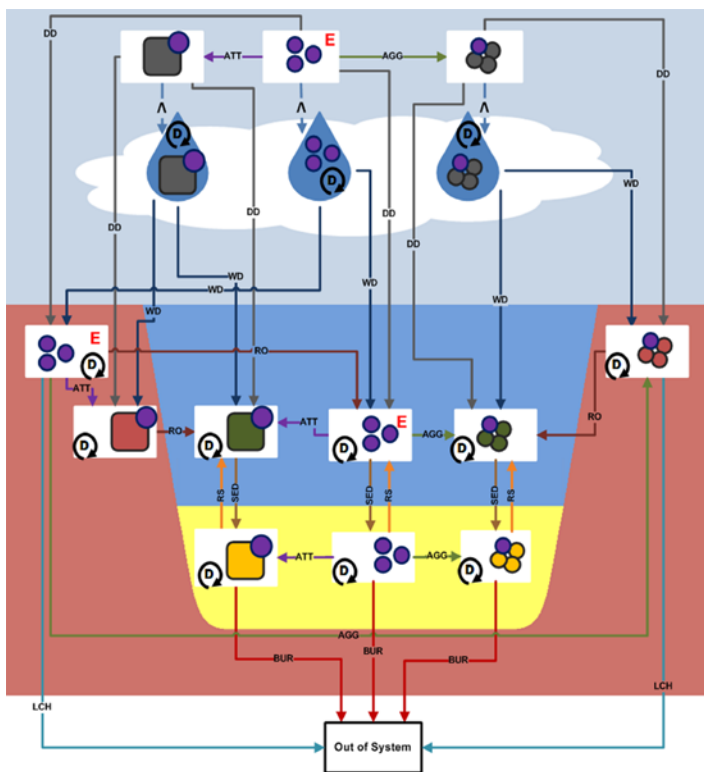


# Environmental fate modeling: Rivers

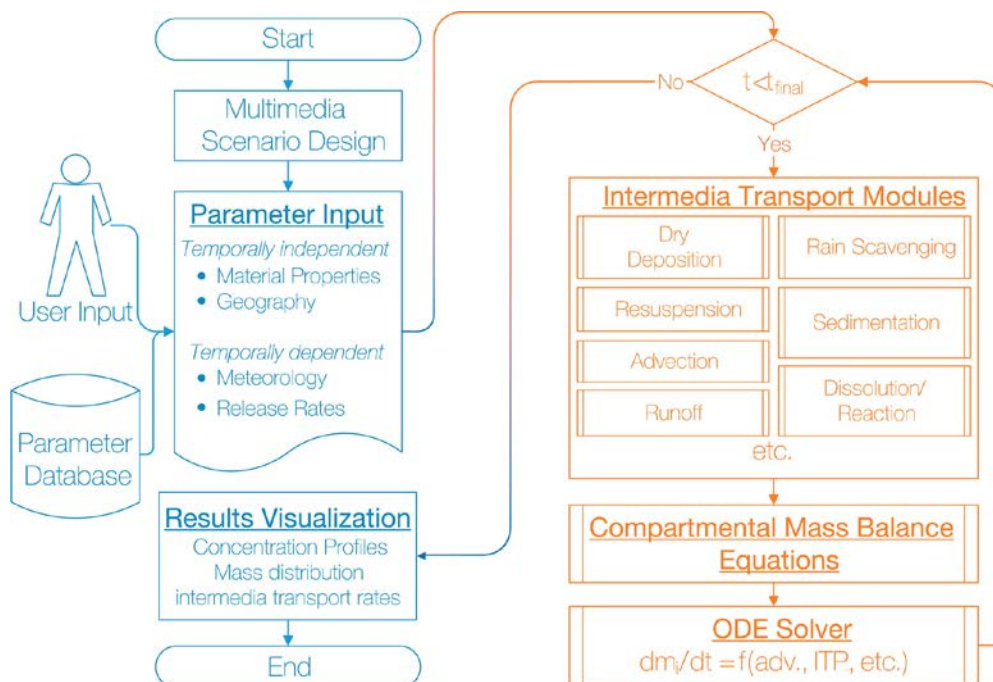


# Complete environmental fate modeling

## SimpleBox4Nano



## MendNano

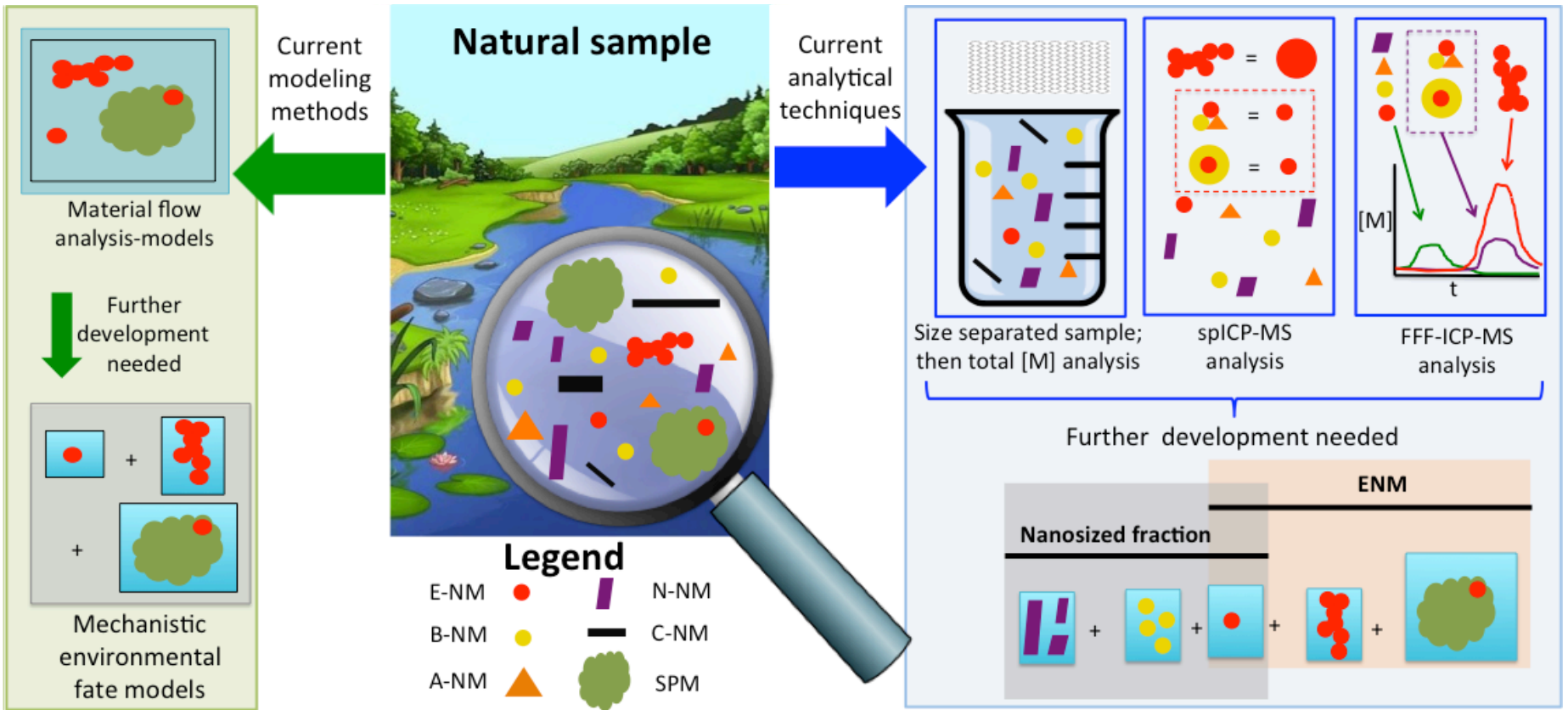


# Coupled MFA and fate model

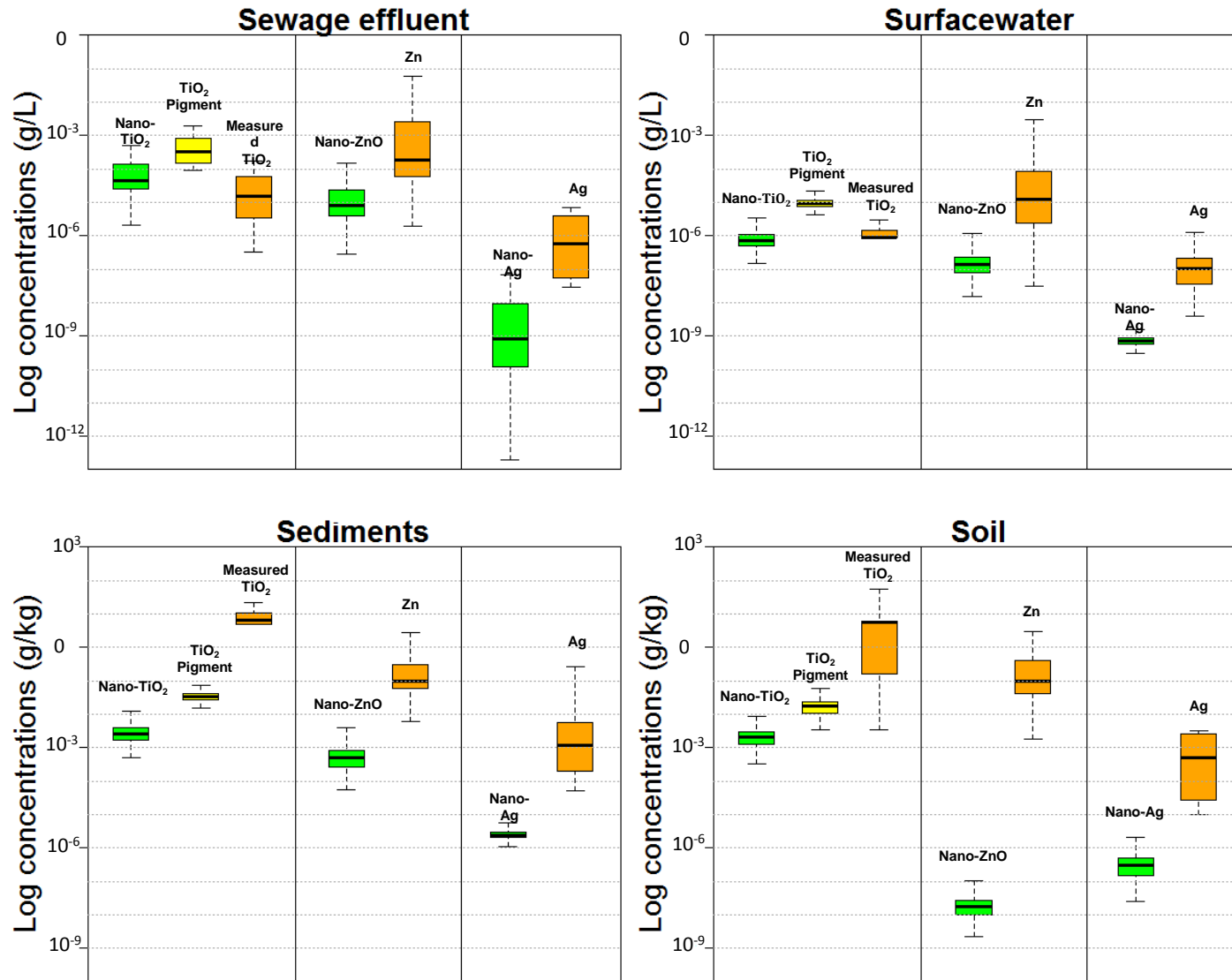




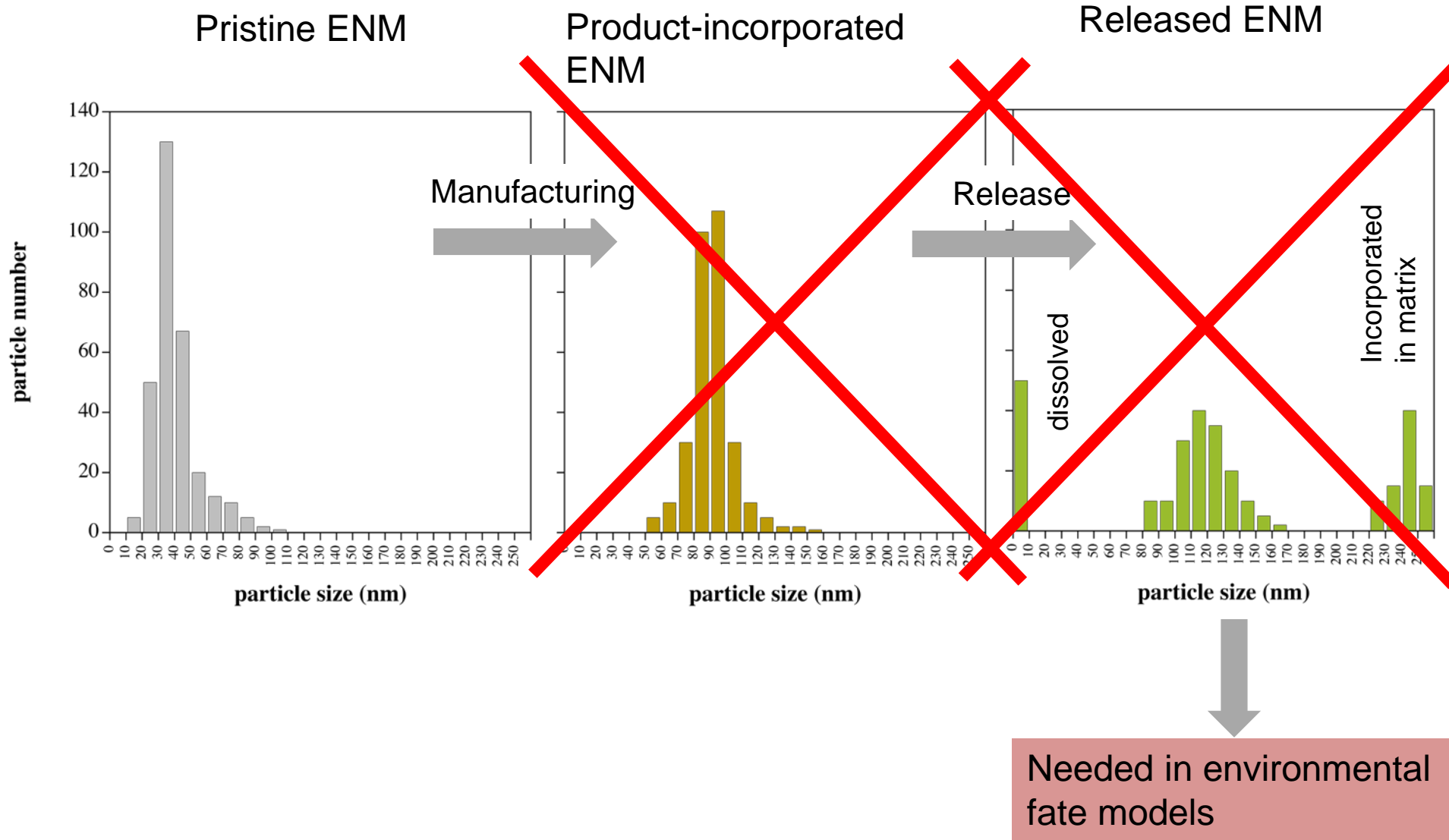
# Validation



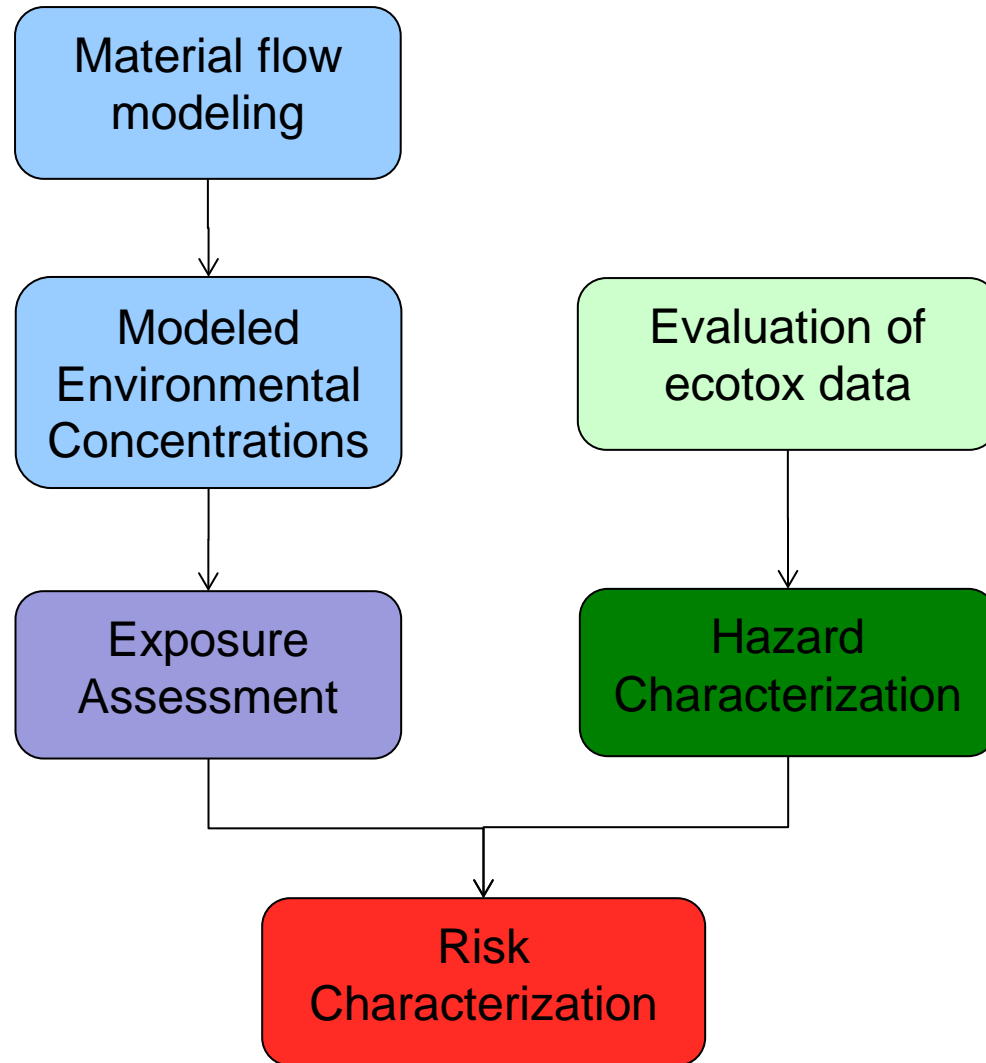
# Comparison nano and conventional metals



# Needs for fate modeling

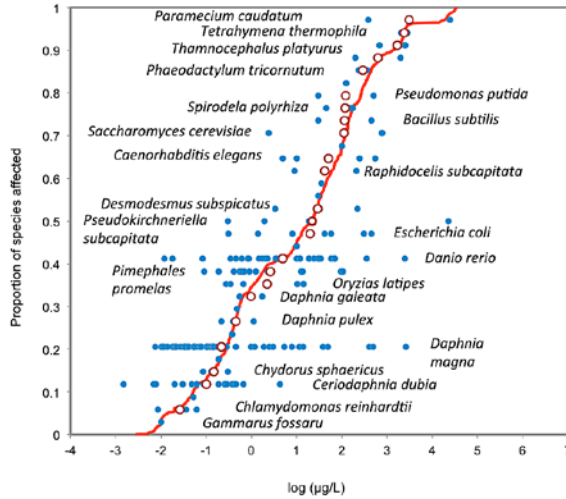


# Risk Characterization of ENM

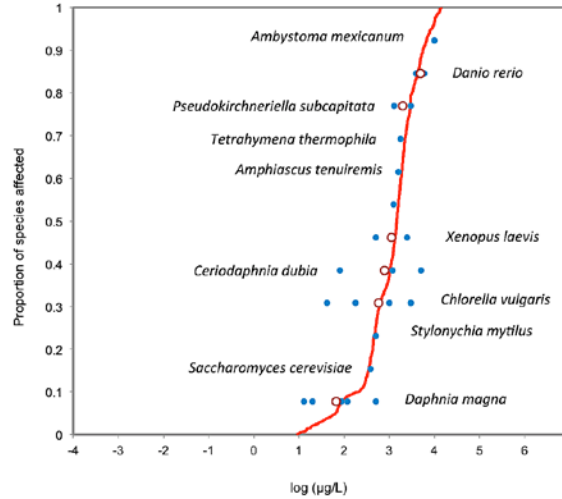


# Hazard characterization (freshwater)

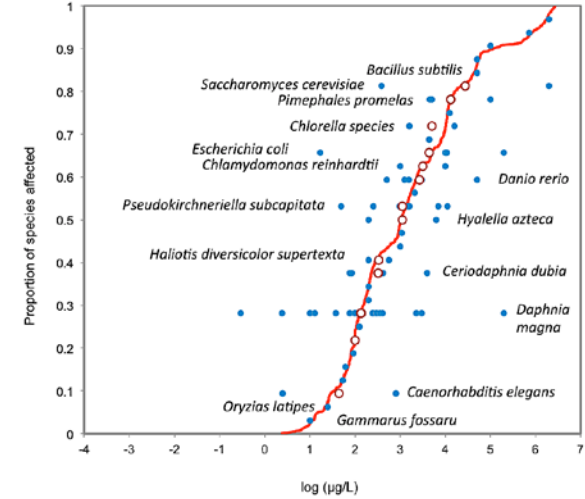
## Nano-Ag



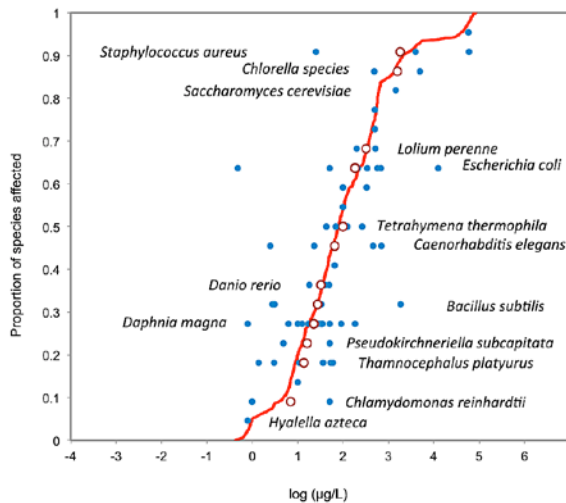
## CNT



## Nano-TiO<sub>2</sub>



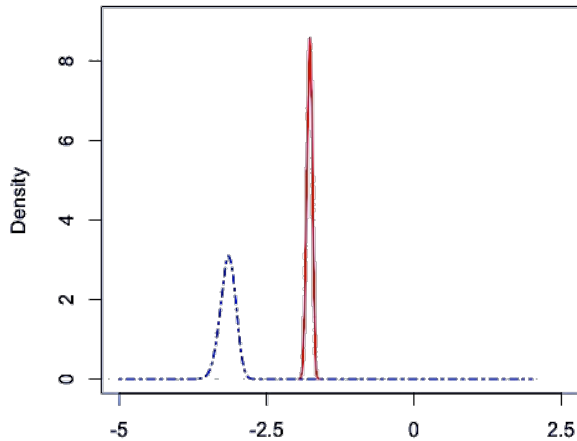
## Nano-ZnO



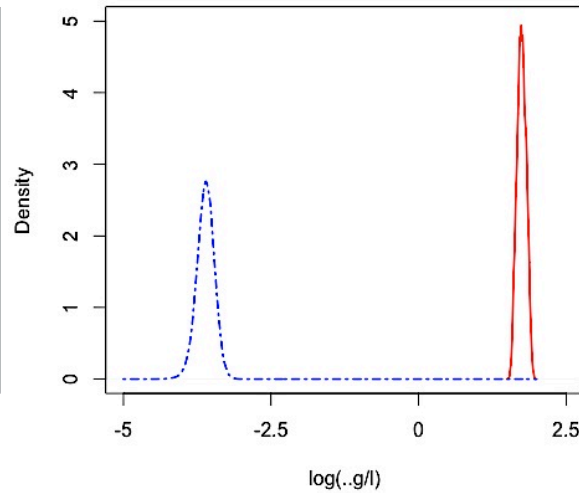
- Species sensitivity distributions (SSD)
- State: 2014
- Available for water, soil, (sediment)
- Used to derive PNEC from HC5

# Risk characterization (freshwater)

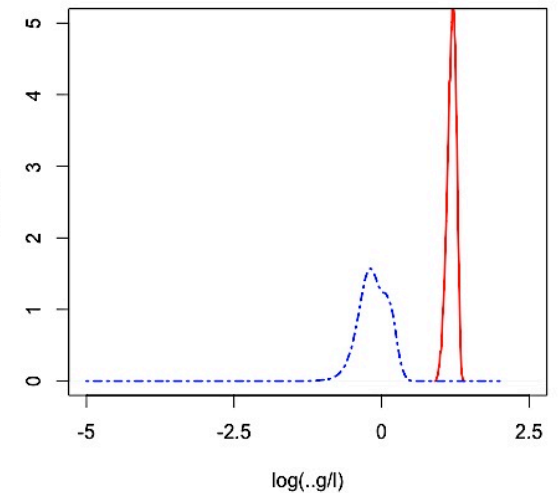
Nano-Ag



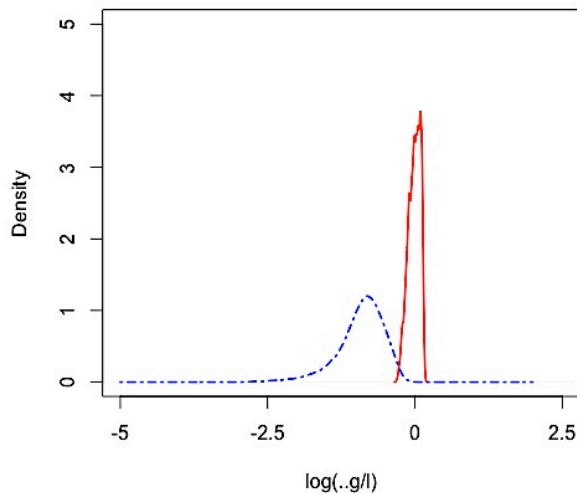
CNT



Nano-TiO<sub>2</sub>



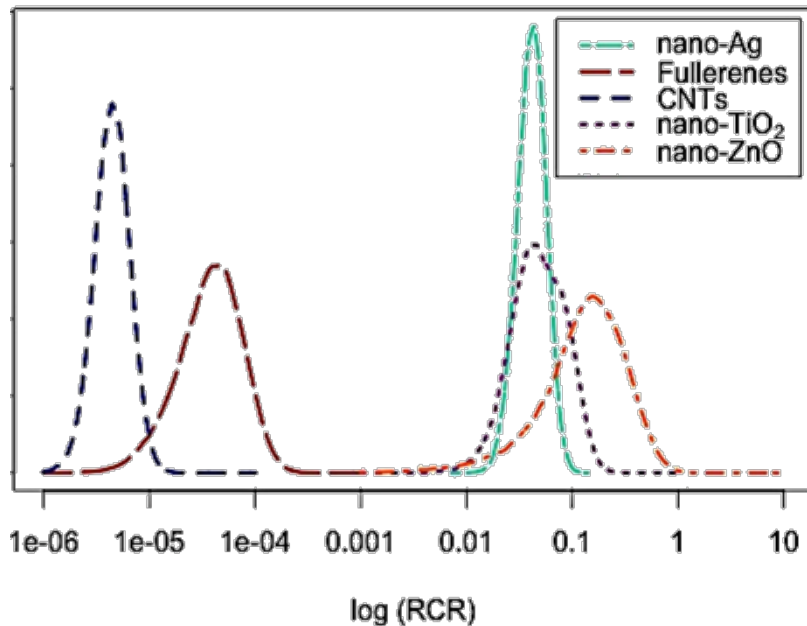
Nano-ZnO



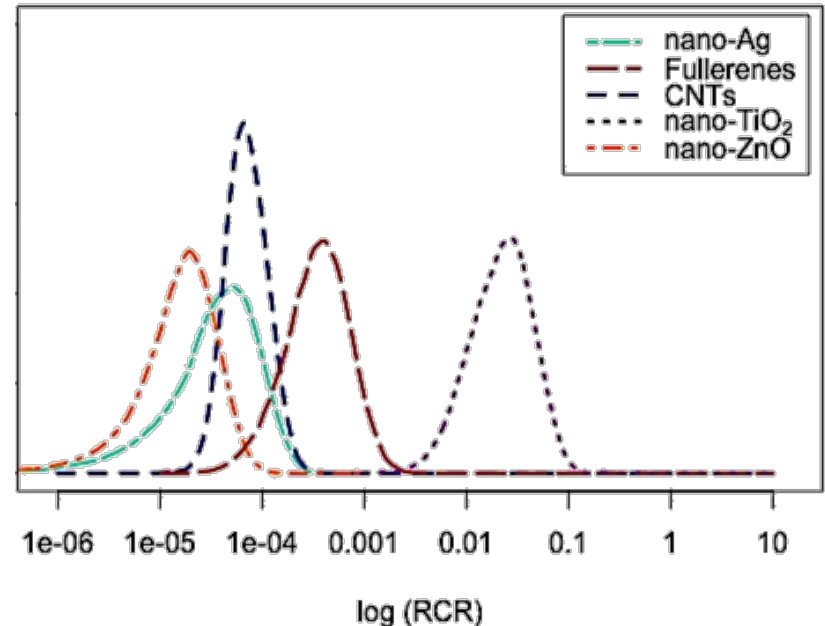
- Predicted environmental concentration (PEC) from Sun (2014)
- Predicted no effect concentrations (PNEC) from SSD

# Risk characterization

Freshwater



Soil



$$\text{Risk Characterization Ratio (RCR)} = \text{PEC/PNEC}$$

# Conclusions

- Life-cycle-based modeling can provide predictions of flows to the environment
- Knowledge about use and release of ENM is scarce
- Release is the link between products and exposure
- Flow data can be converted to environmental concentrations
- Coupling of flow and fate modeling is possible, but size distribution input data are not available
- Validation of modeled data by measurements is not yet possible
- Comparison of exposure concentrations with effect data can be performed



Thank you for your attention!

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