

Bridging Technologies

Paul V. Braun

pbraun@illinois.edu

<http://braungroup.beckman.illinois.edu>

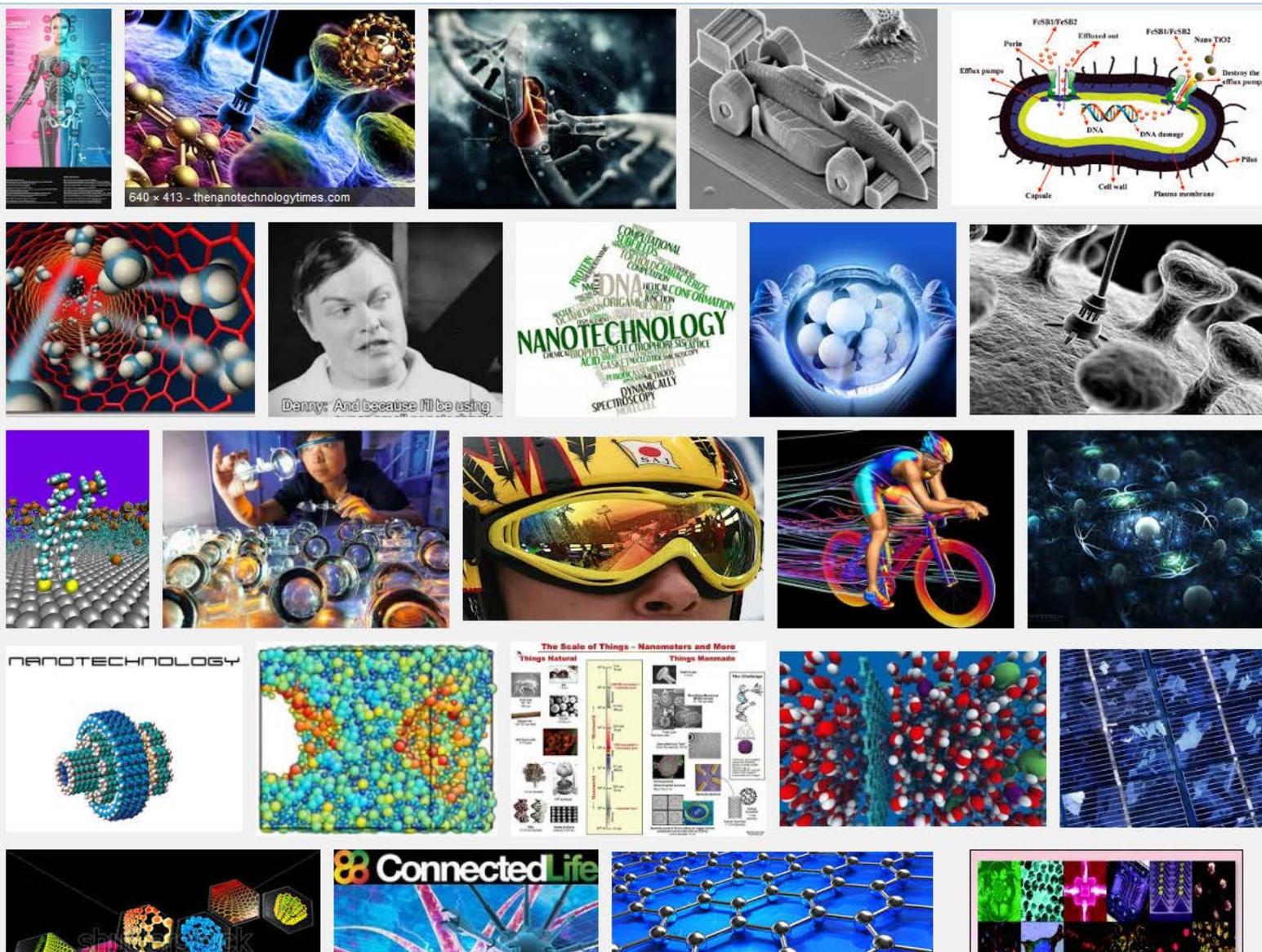
Department of Materials Science and Engineering,
Frederick Seitz Materials Research Laboratory, and Beckman Institute

University of Illinois at Urbana-Champaign, Urbana, IL

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“new form factors of matter lead to new functions”

Google: "Nanotechnology" images (June 7, 2012 – June 7, 2013)

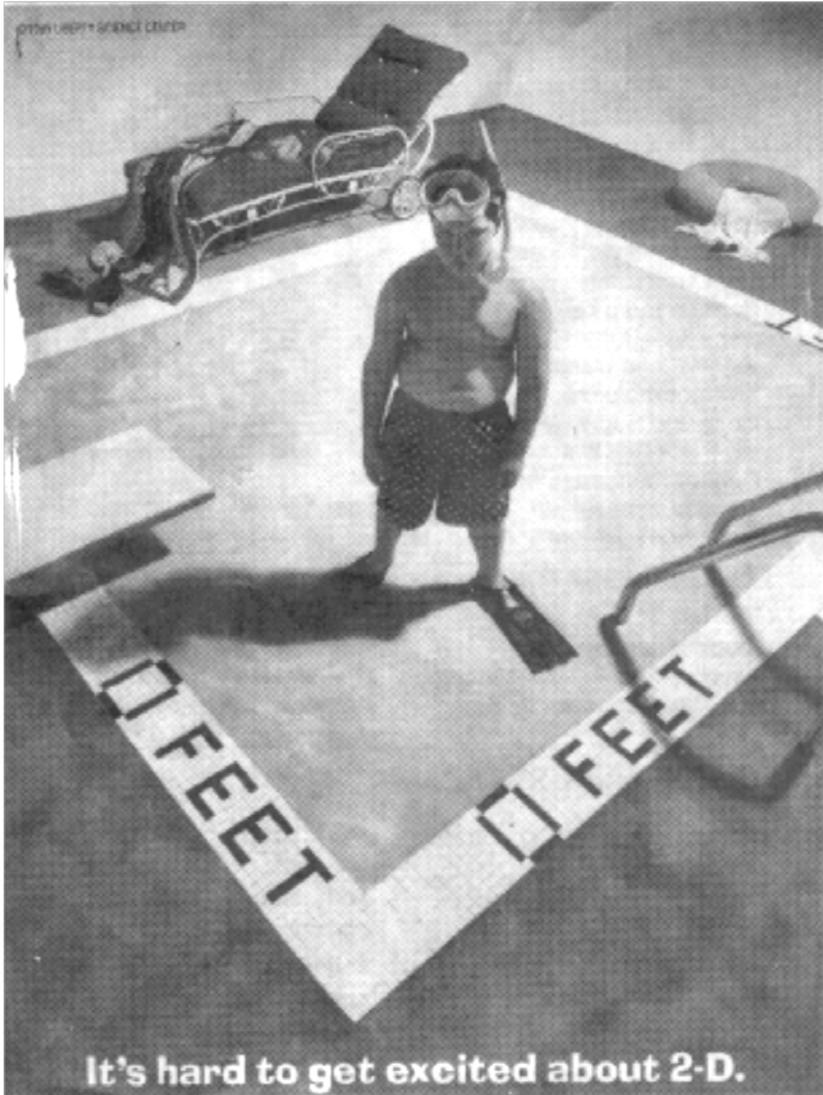


Products: 3
(all thin films)

Tiny things: 10+

How to reverse the ratio?

Going Beyond 2D



2D nanotechnology is virtually everywhere, and of critical importance. But high-value nanotechnology needs to expand its impact beyond microelectronics, thin films, and medicine.

Necessary to Bridge Technologies from nano to macro to make nanotechnology the value added component.

“It's hard to get excited about 2-D”

Macroscopic Nanotechnology

Examples of Macroscopic Nanotechnology

- **Carbon Black**
 - world production ~10,000,000 tons
 - Tires are ~50 wt% carbon black
 - Volume production since ~1900
- **Fumed Silica and Silica Fume**
 - used in cement and as polymer reinforcement
 - Volume production since ~1950
- **Clay**
 - used for 15,000 years
- **Alloys**
 - many metal alloys are nanostructured
- **Polishing media**
 - from low tech to high tech
- **Pigments**
 - e.g. for paints and coatings
 - many variations, often nanostructured
- **Energy Storage**

Most are commodities. Price only slightly greater than raw materials cost + energy input cost.

Existed long before “nanotechnology”.
Term “nano-technology” first used in 1974 by Taniguchi.

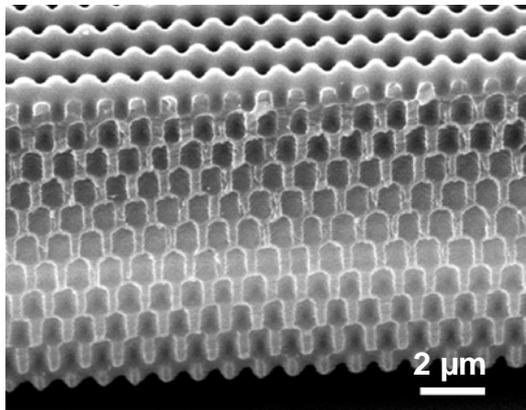
Translating Nanotechnology into Macroscopic Systems

May require some level of “bottom-up” (but not necessary exclusively so)

- Needs to be much more than just building blocks
 - add value through functionality
 - add value through substantive and broad IP (hard for building blocks)
- Must provide a paradigm change. Otherwise the steady rate of progress of established products will surpass the functions of the “nanotechnology”

Structural Complexity

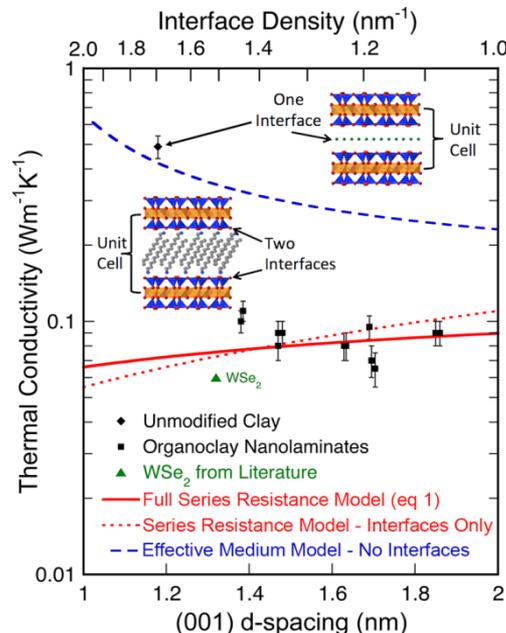
Interference lithography provides 3D structures in a single step



Campbell, et al. *Nature* 2000
Yang, Wiltzius, et al. *Chem. Mater.* 2002

Heat Transfer

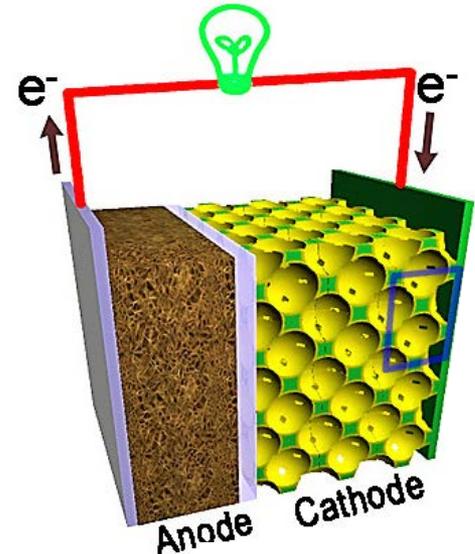
Ultralow thermal conductivity in self-assembled 3D structures



Losego, Braun, et al. *Nano Letters* 2013

Rechargeable Batteries

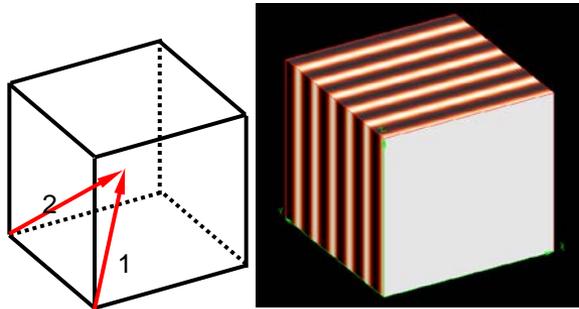
Ultrahigh power density through electrode nanostructuring



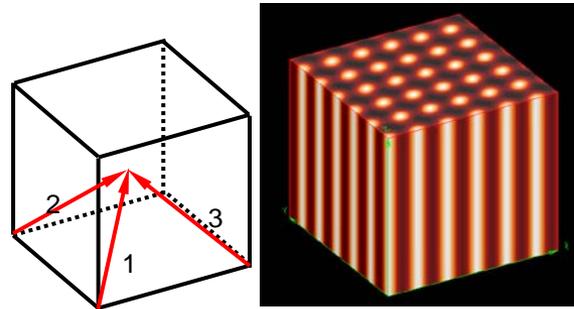
Zhang, Yu, Braun *Nature Nanotech.* 2011

Multibeam Holography: Macroscopic Nanostructuring

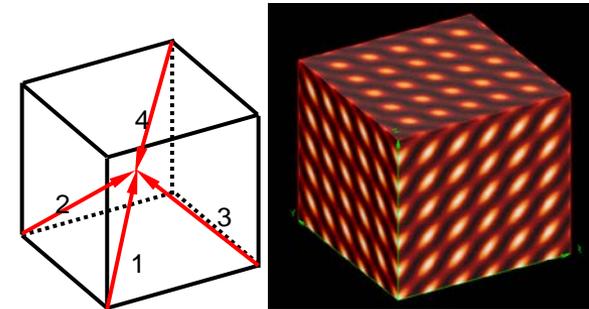
2 beams (1D)



3 beams (2D)

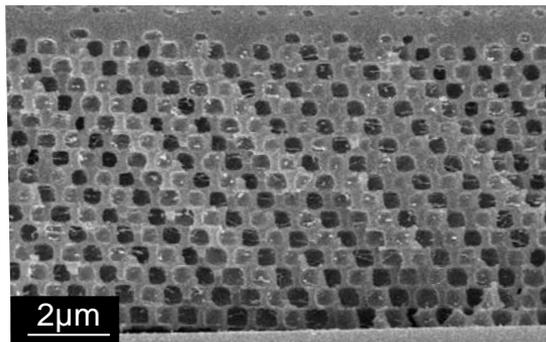


4 beams (3D)

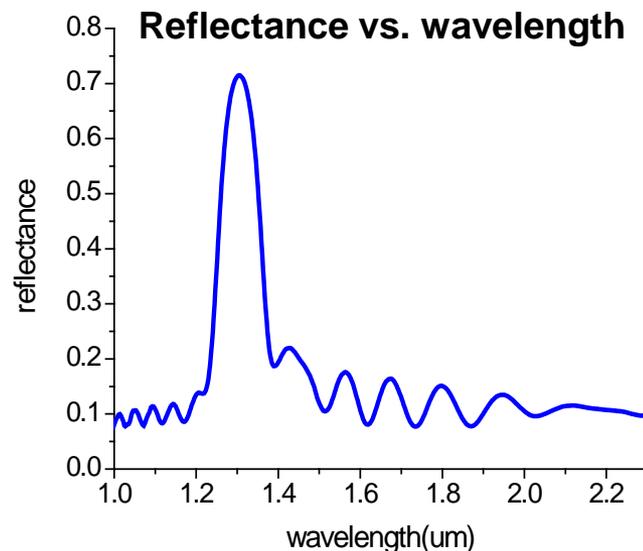


$$I = \underbrace{\sum_j E_j^2}_{\text{power}} + \underbrace{\sum_{i < j} 2E_i E_j \cos \theta_{ij}}_{\text{polarization}} \underbrace{\cos[(\mathbf{K}_i - \mathbf{K}_j)\mathbf{r}]}_{\text{wavevector}} \underbrace{+ \varphi_{0i} - \varphi_{0j}}_{\text{phase}}$$

- beam geometry
- wavelength
- refractive index

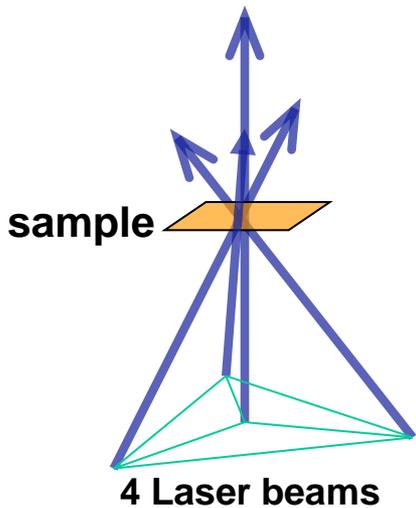


SEM cross-section

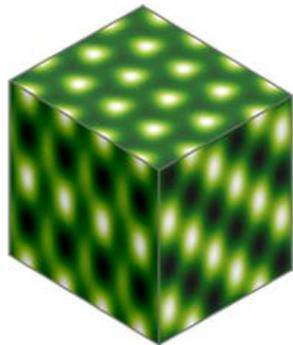


Multibeam Interference Patterning of Materials

Low cost, large area periodic structures



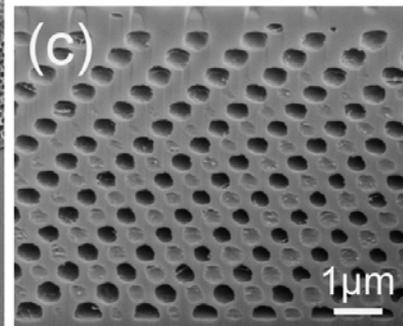
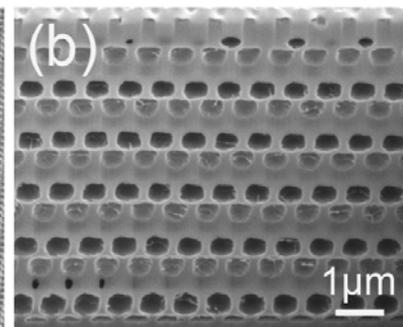
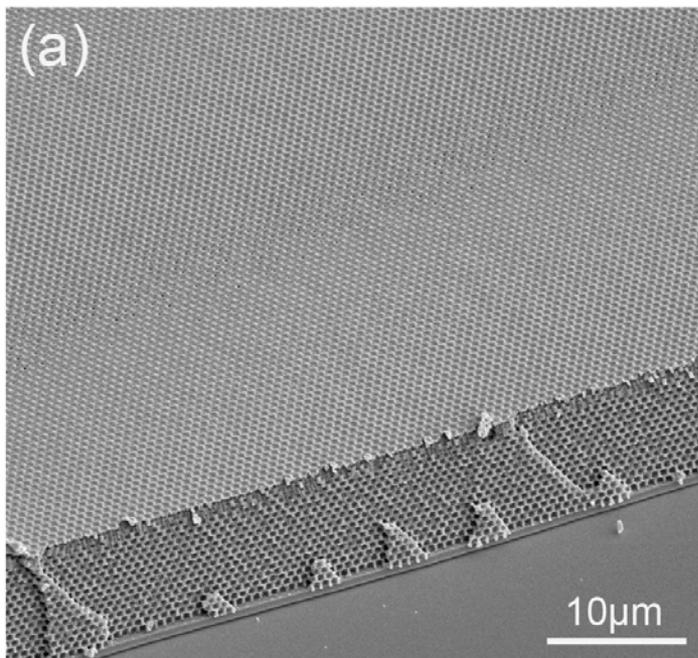
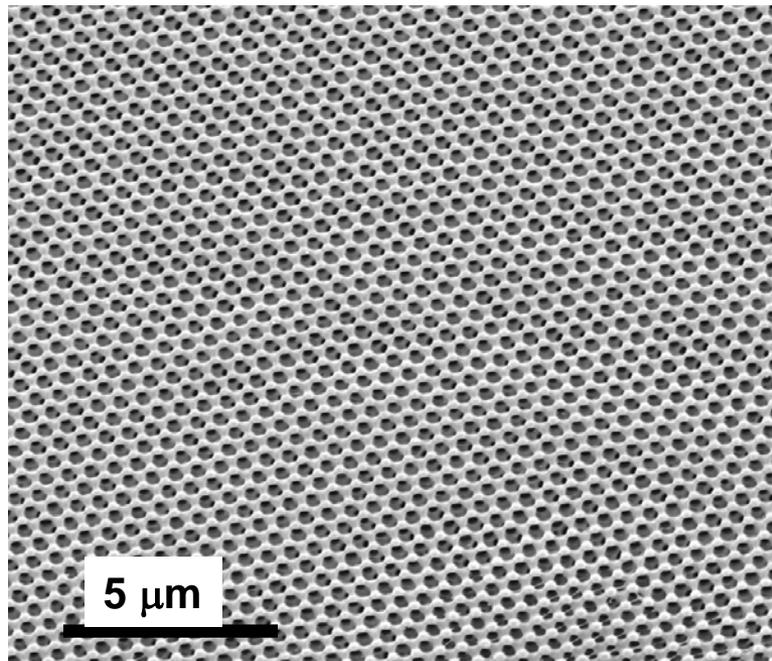
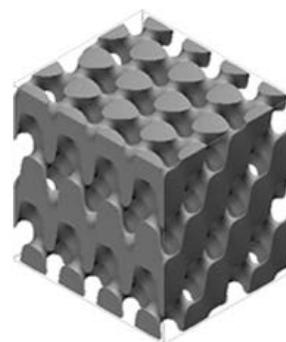
Interference pattern



Remove unexposed material



Resulting Structure

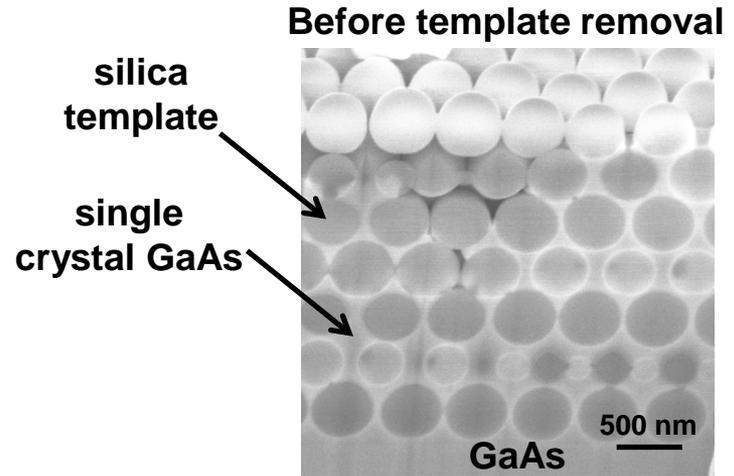
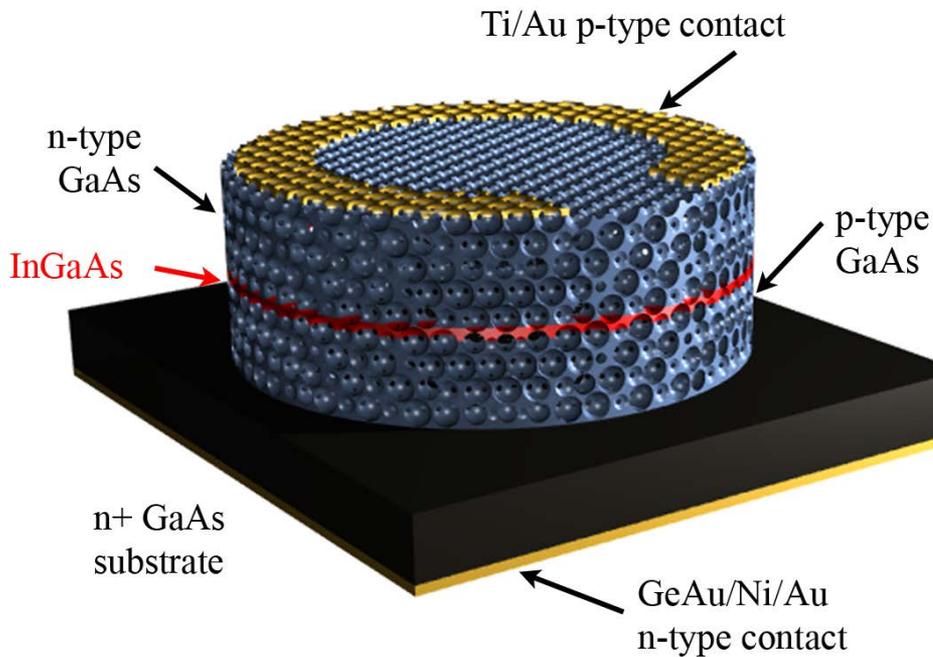


Polymer Photoresist (SU-8)

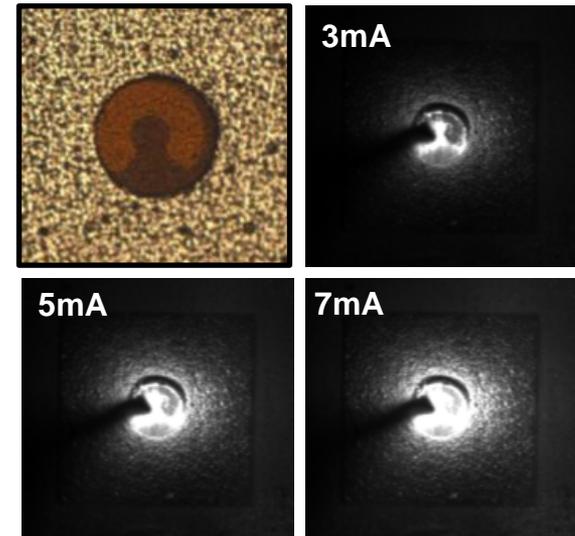
3D Photonic Crystal-Based LED

First example of:

- * Electrically pumped emission from a 3D PC LED
- * Epitaxial growth of porous 3D semiconductors



Electrically Pumped IR Emission

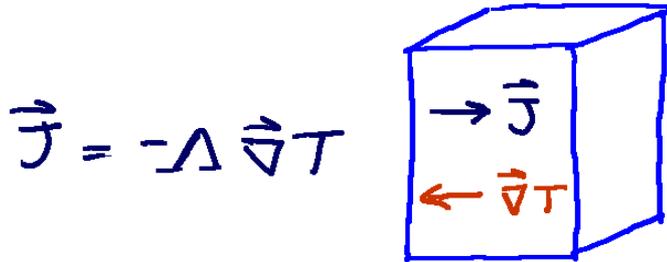


Required merging a common 2D semiconductor growth with a 3D self-assembled template

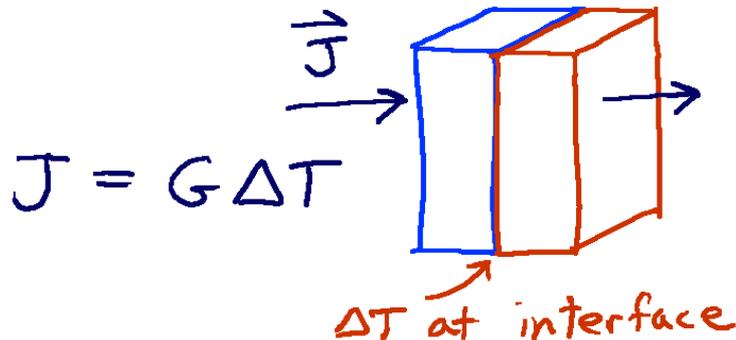
Limits to Thermal Conductivity (in solids)

How does nanotechnology relate to thermal conductivity?

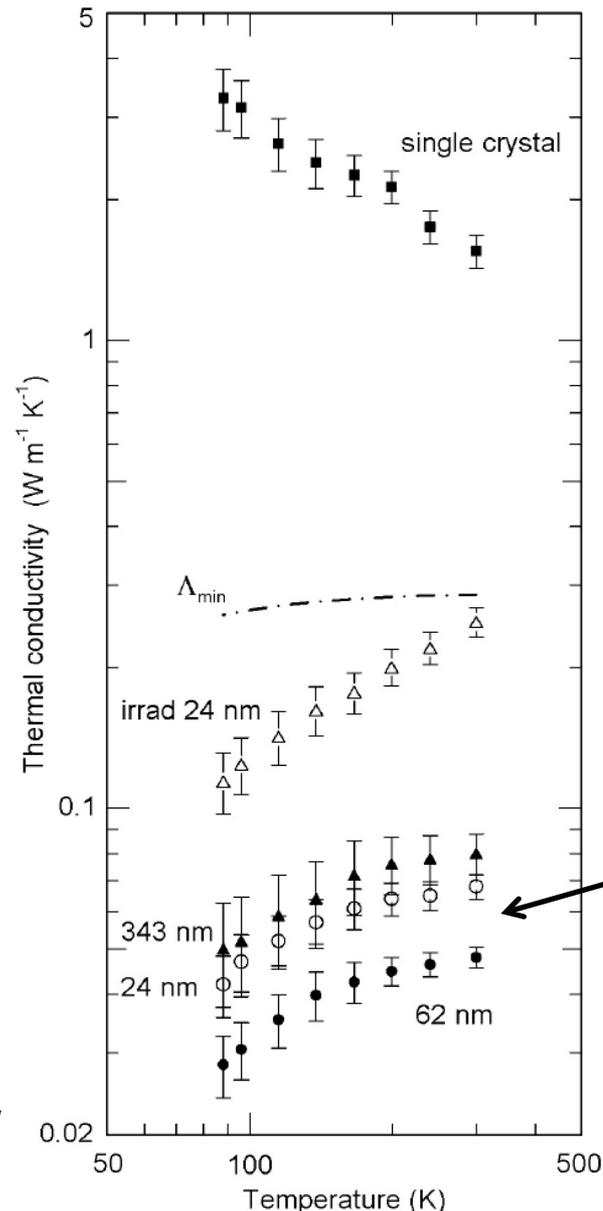
- Thermal conductivity Λ is a property of the continuum



- Thermal conductance (per unit area) G is a property of an interface



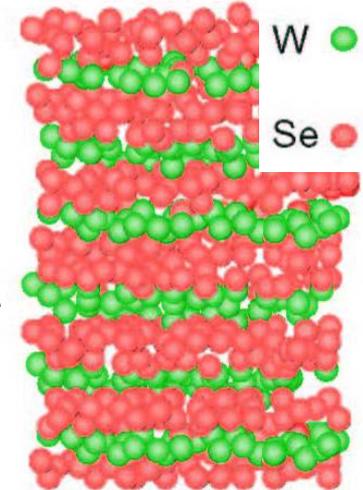
So, as the density of interfaces increases, thermal conductivity decreases. Interfaces important only for very high interface densities ($\sim 1 \text{ nm}^{-1}$)



Potential Impact

- Thermoelectrics
- Insulation
- Thermal switch
- Thermal isolation
- Thermal solar

Layered WSe_2
(vacuum deposited)



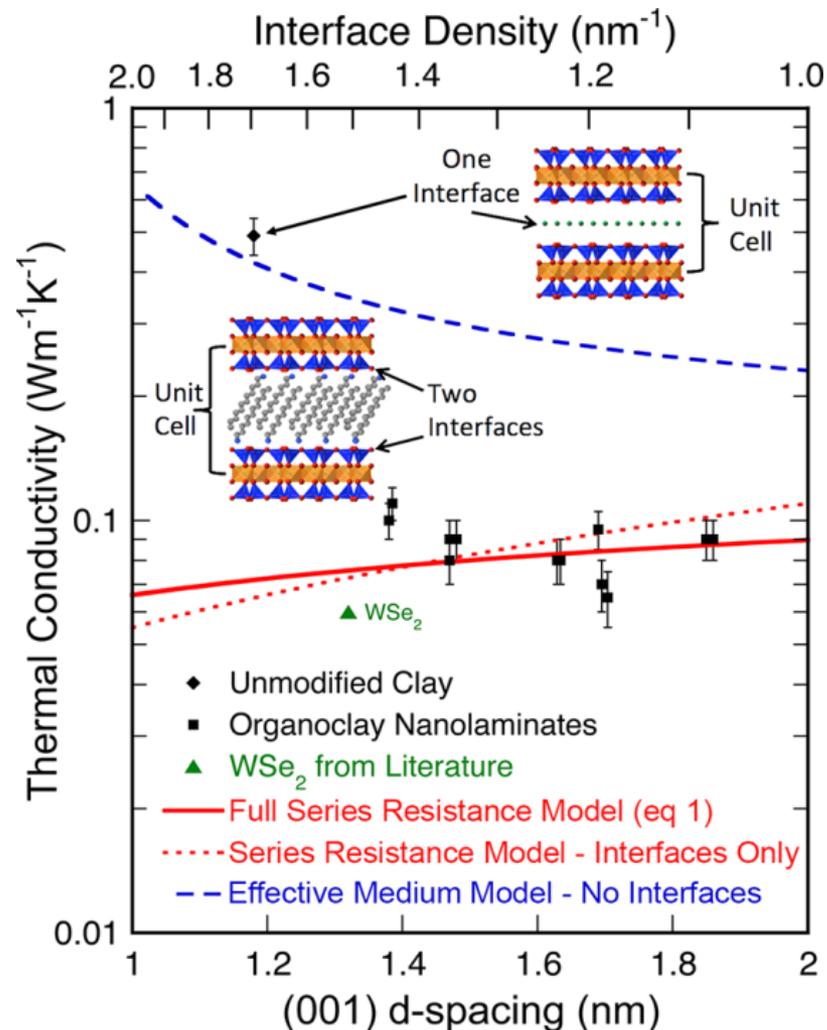
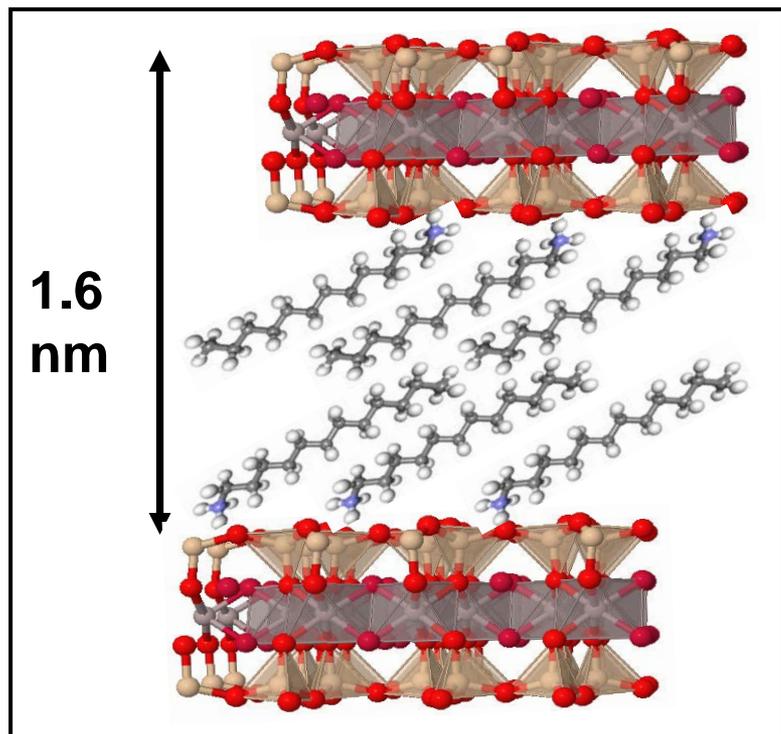
Cahill Science 2007

Ultra-low Thermal Conductivity: Towards Macroscopic Systems

Layered Clays

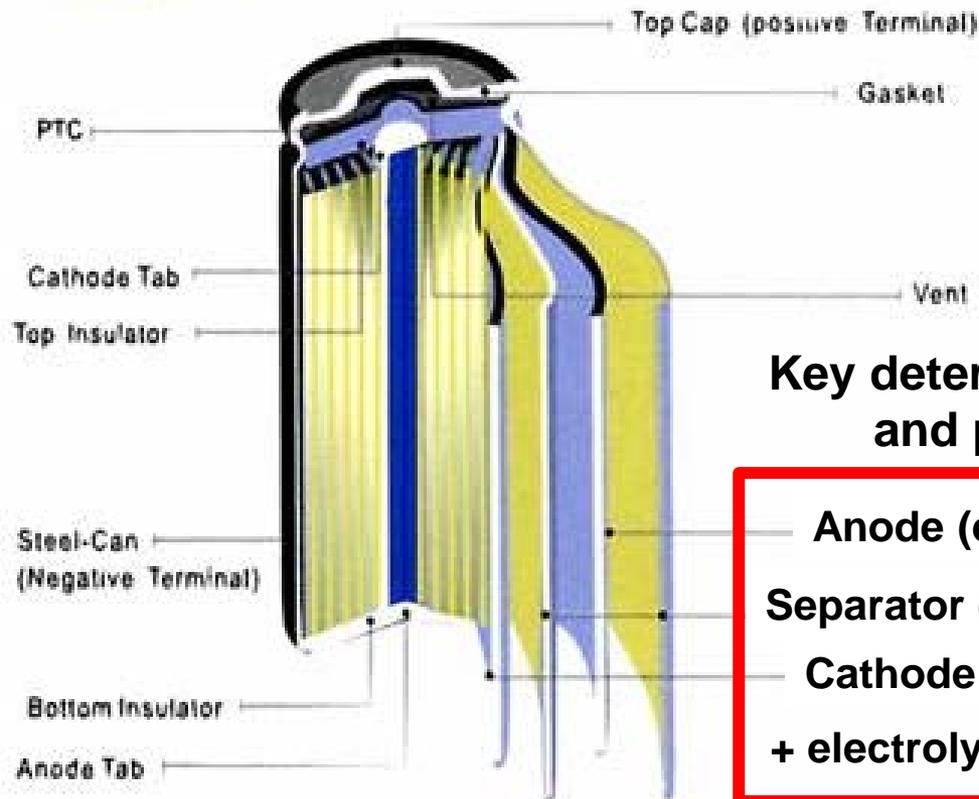
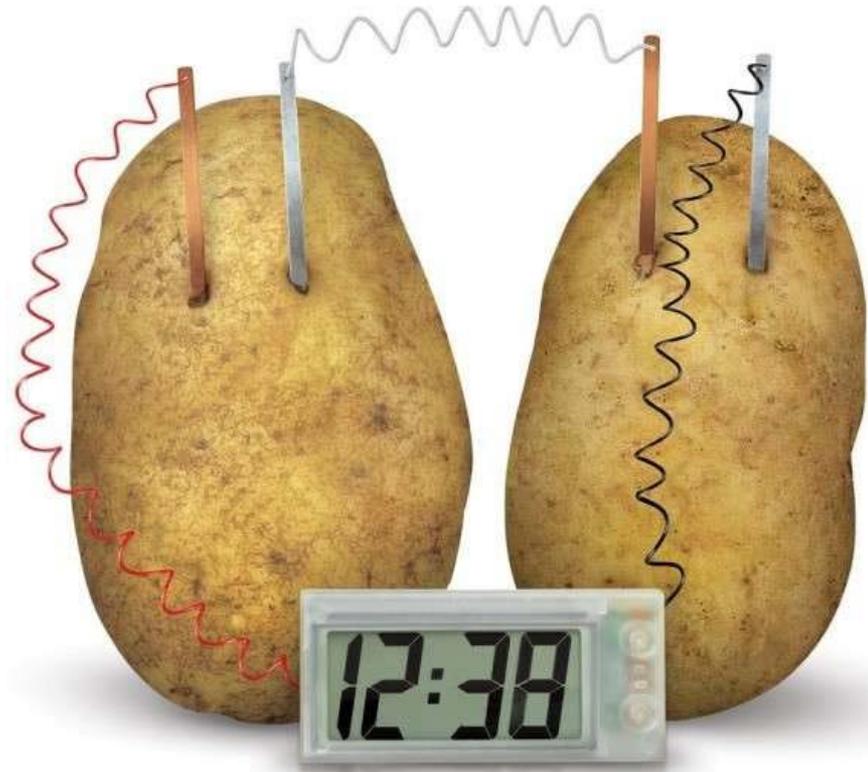
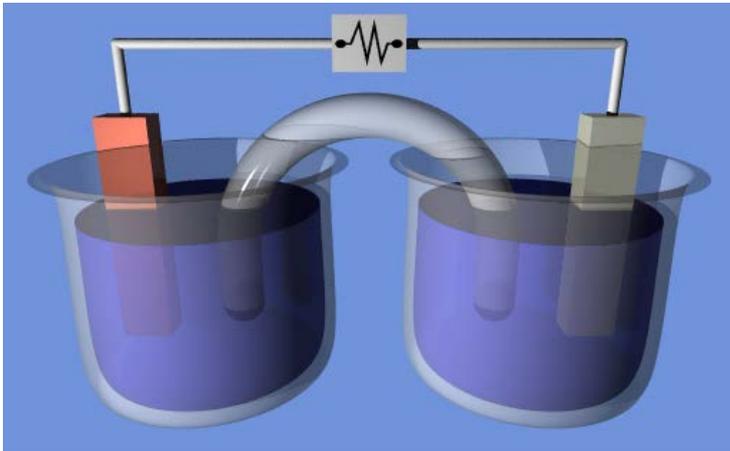
- ~1 interface/nm
- Self-assemble into layered structures
- Produced worldwide in high volume

Organically Modified Layered Clay



Nanostructuring results in a thermal conductivity which is 50% of a typical plastic and 10% of glass

Batteries (always “Bulk” materials)

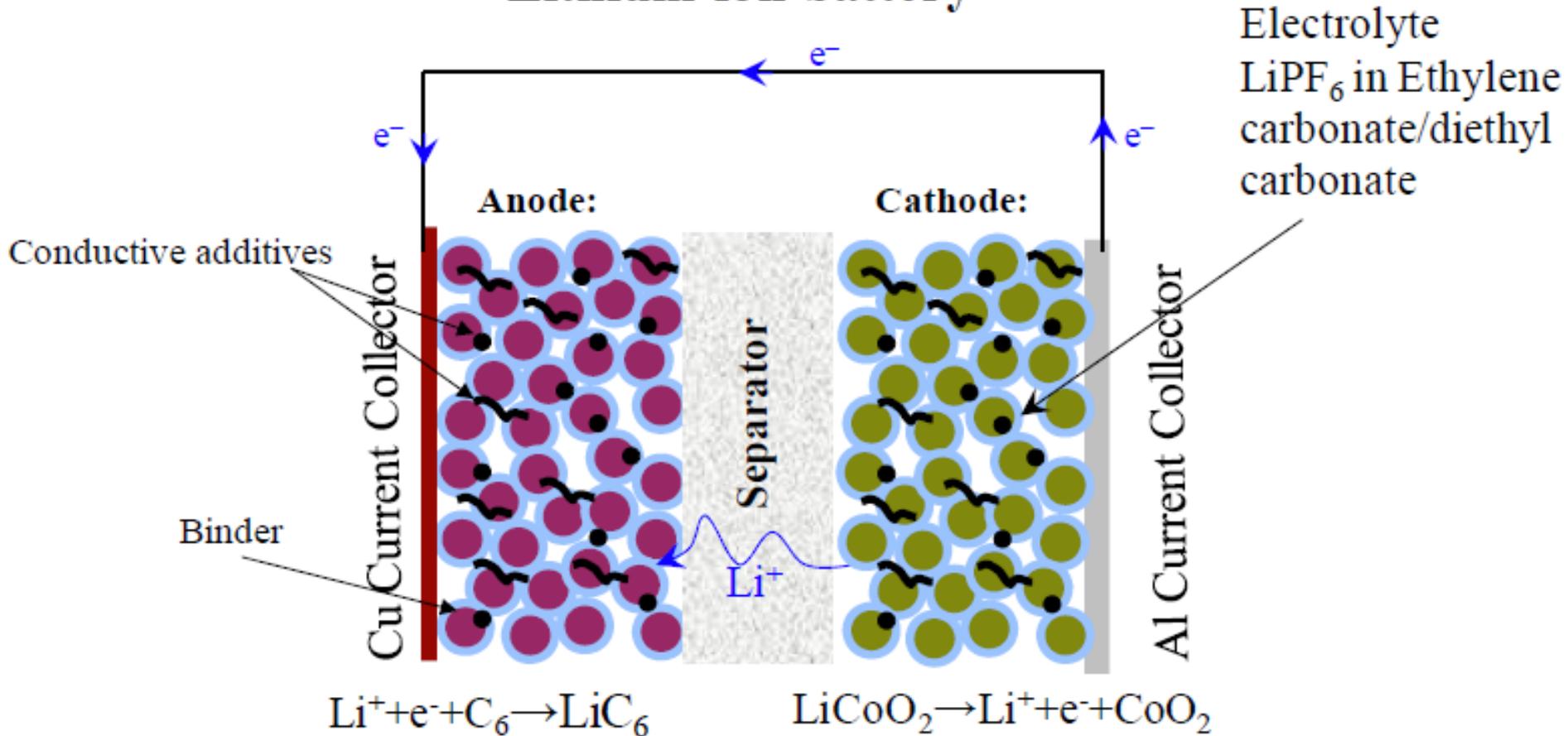


Key determinants of energy and power density

- Anode (e.g. carbon)
- Separator (porous polymer film)
- Cathode (e.g. LiMnO_2)
- + electrolyte (solid or liquid)

Inside a Li-ion Battery

Lithium-ion battery



Continuous ion and electron transport pathways in electrodes critical
Provided by pore network and conductive additives

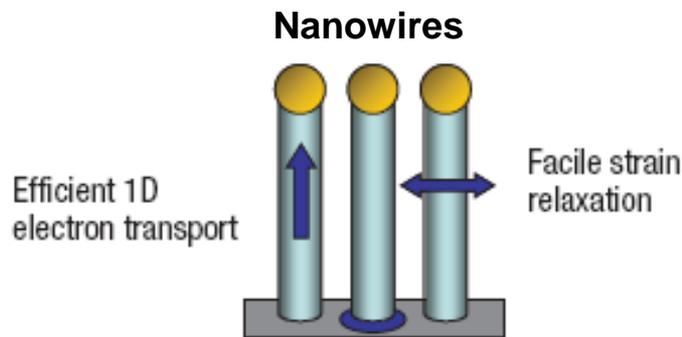
Pathways to Increase Power & Energy Density

1. New Materials

2. Structure Design

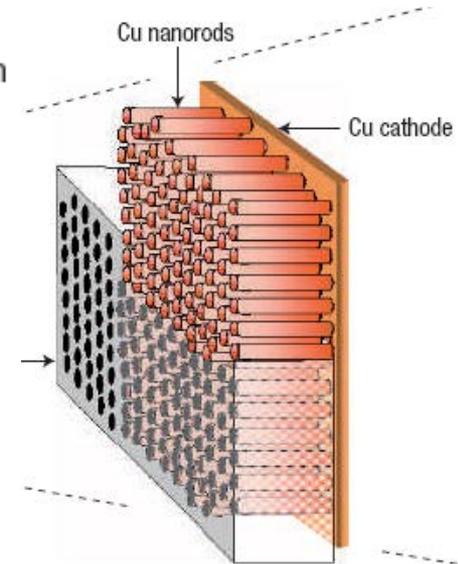
- 3D electrode architecture
- Large surface area
- Thin film of active materials

REDUCE TRANSPORT LENGTHS



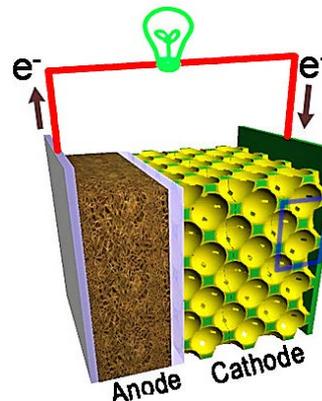
Good contact with current collector
Nature Nanotechnology 2008, 3, 31

Nano-pillar current collectors

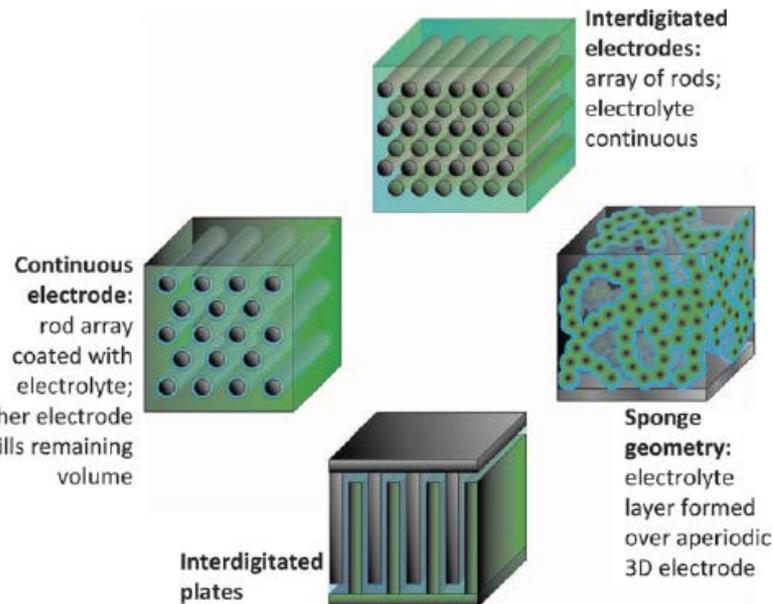


Nature Materials 2006, 5, 567

3D Bicontinuous Electrodes



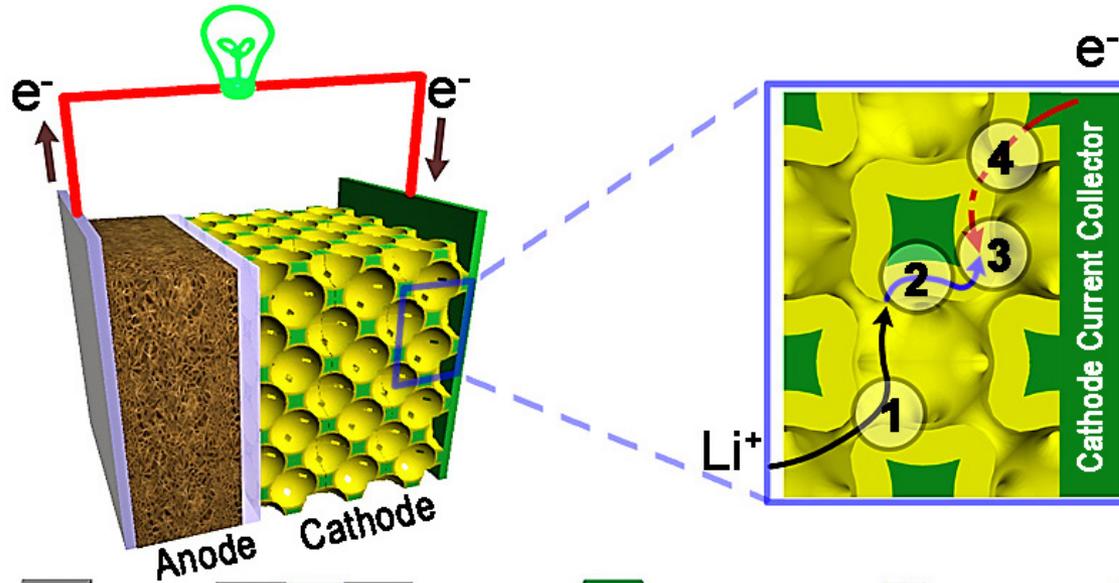
Zhang, Yu, Braun *Nature Nanotech.* 2011



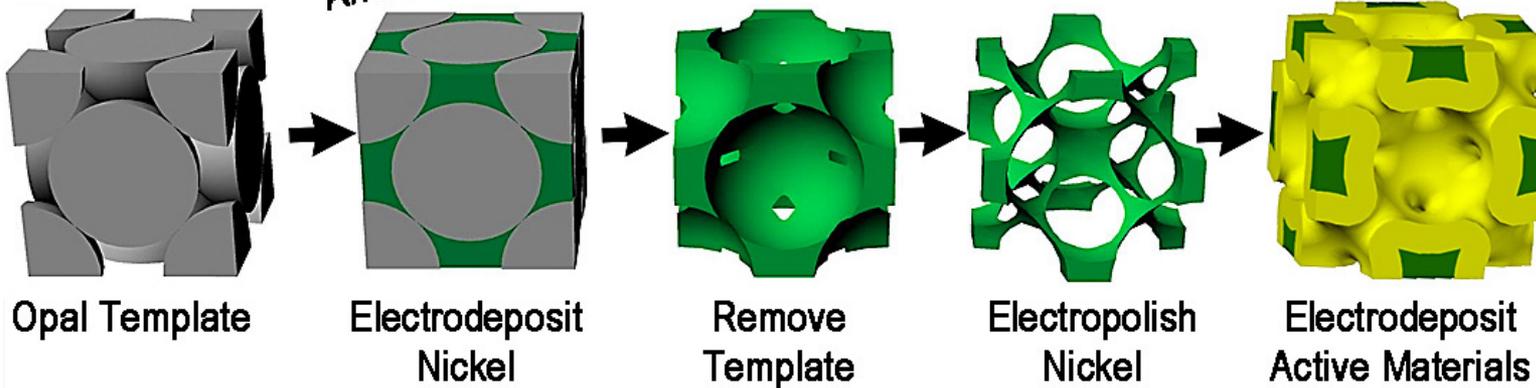
Chemical Society Rev. 2009, 38, 226

Bicontinuous Battery Electrode (Cathode)

Wrap a surface into a 3D structure maximizing kinetics and capacity

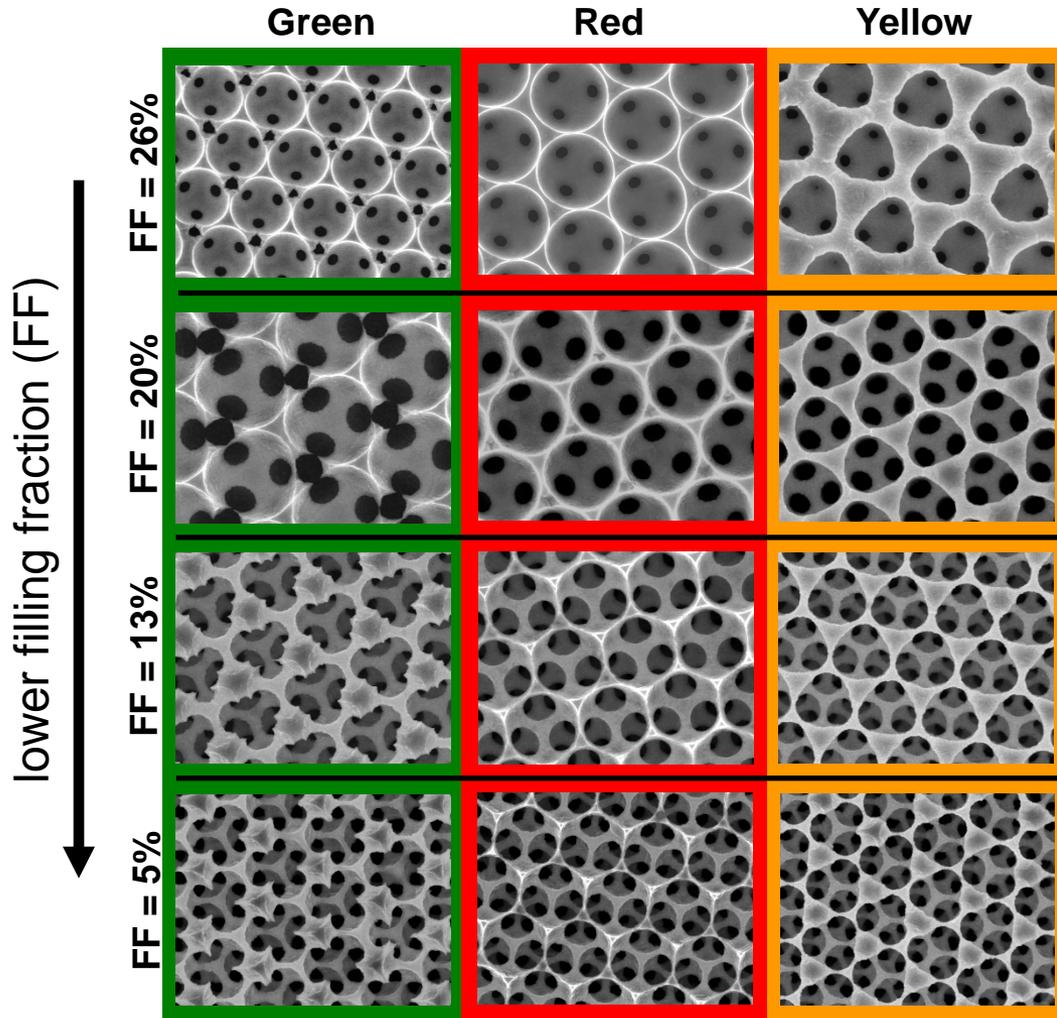


- 1) Facile ion transport through connected pore network
- 2) Short solid-state ion diffusion length
- 3) Rapid electrochemical processes (large surface area)
- 4) Low resistance electrical network

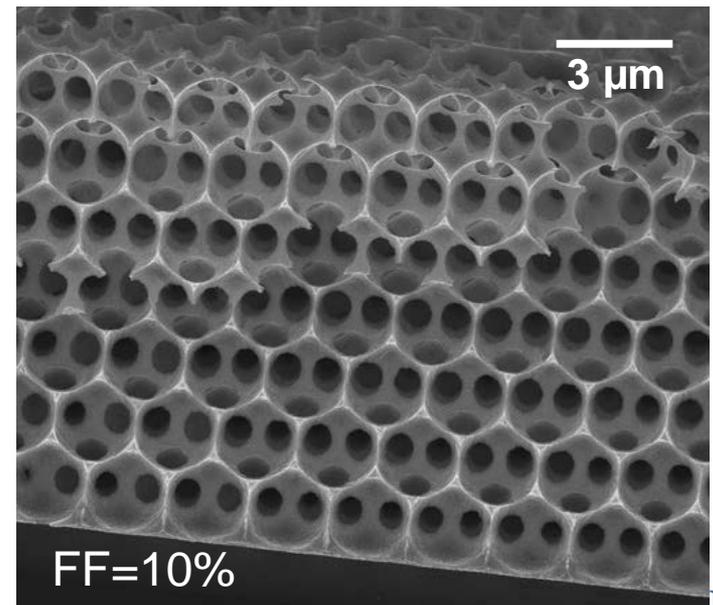
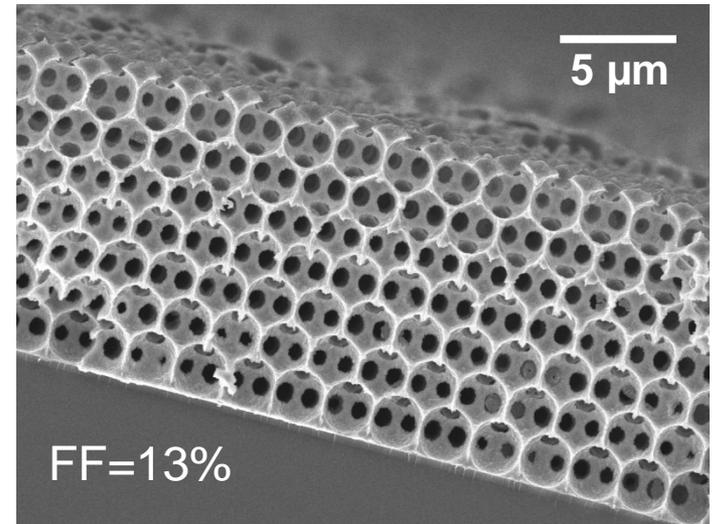


3-D Metal Foams – The Electrode Support

95% Air Metal Foams Possible



Side Views

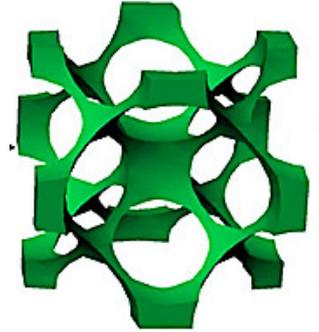
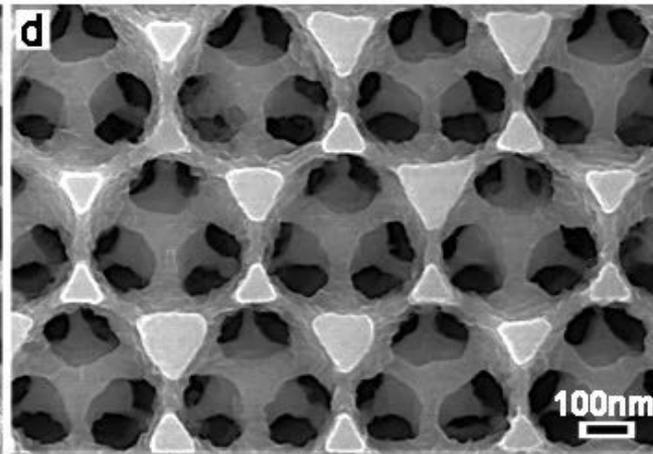
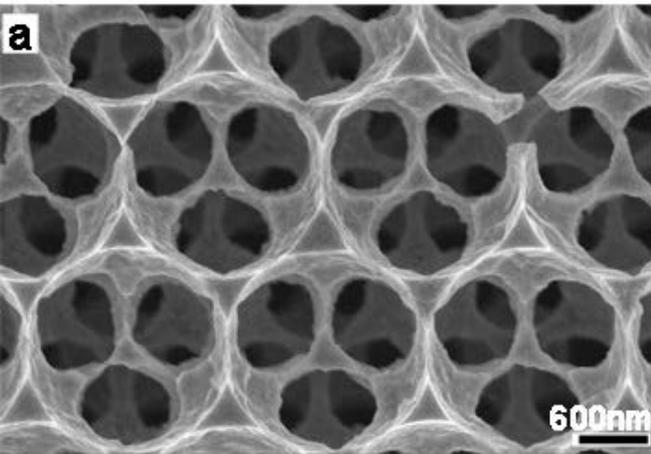


Formed via self-assembly followed by electroplating

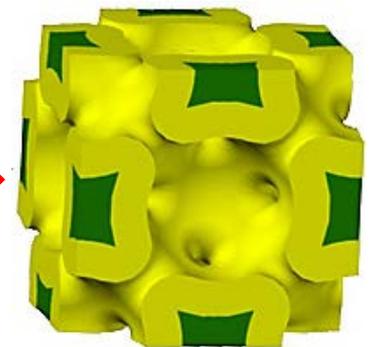
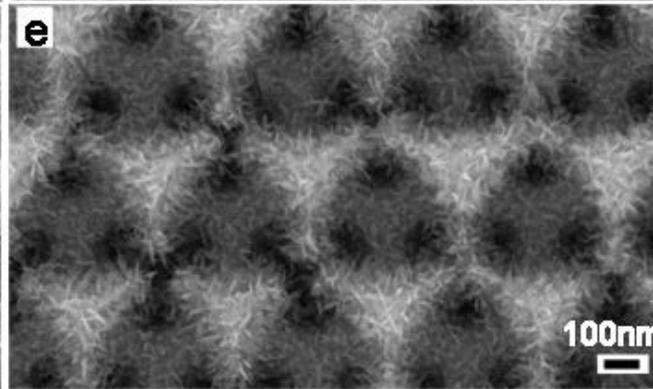
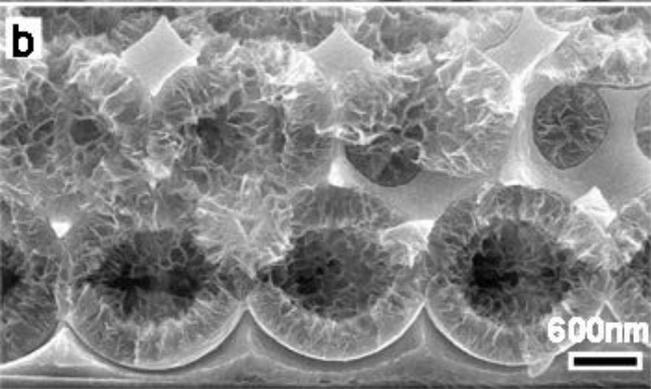
Bicontinuous Battery Electrode (Cathode)

NiMH (NiOOH)

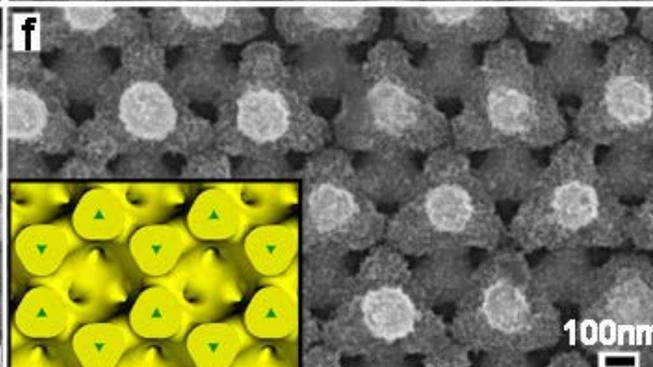
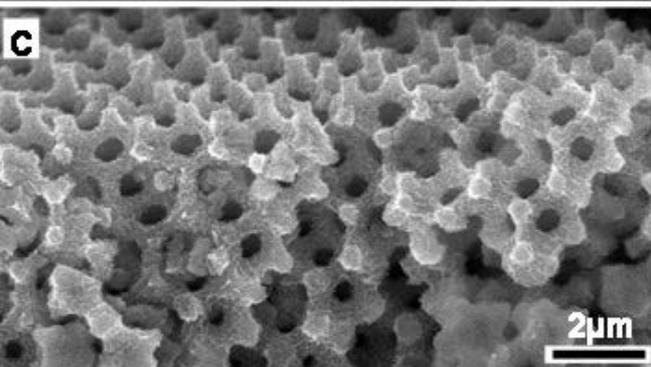
Li-ion (LiMnO₂)



Metal Framework



Coated Framework

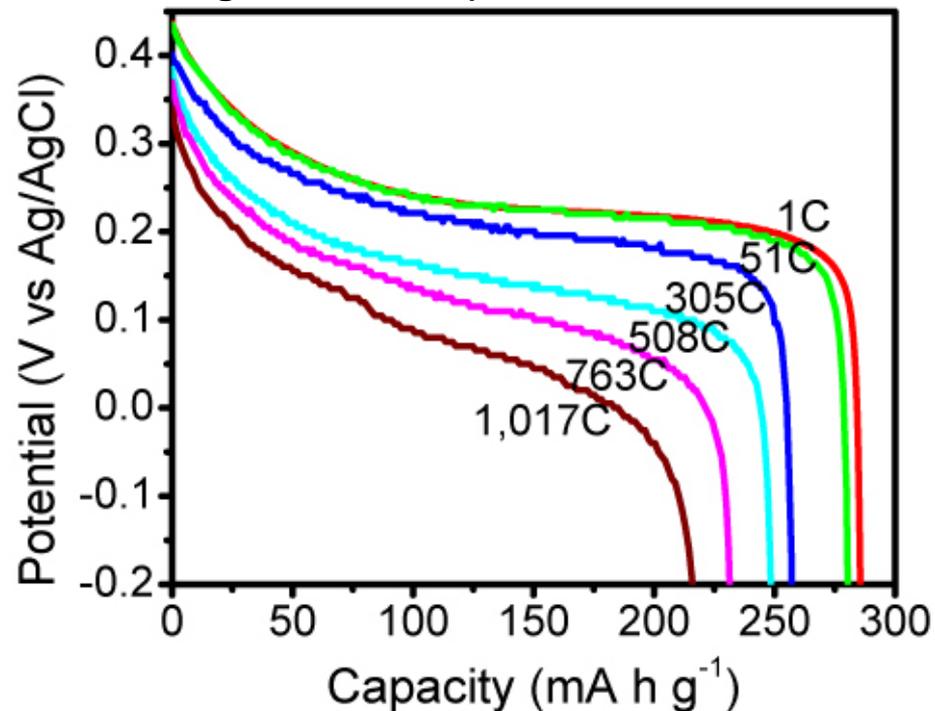


Ultrafast Discharge Characteristics (high power)

Nickel Metal-Hydride Cathode

75% capacity retention at 1000C discharge!

(1C is the current required to fully discharge the battery in 1 hour, 1000C is the current required for a full discharge in ~3.6 sec.)

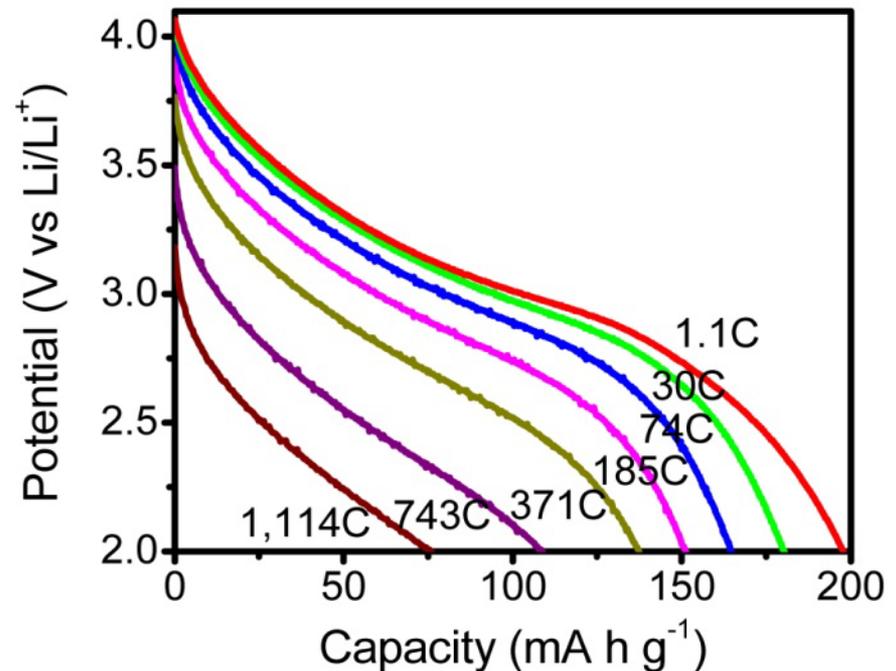


In a typical NiMH cathode, capacity drops dramatically above ~25C

Electrochem. Solid-State Lett.
2007, 10, A56-A59

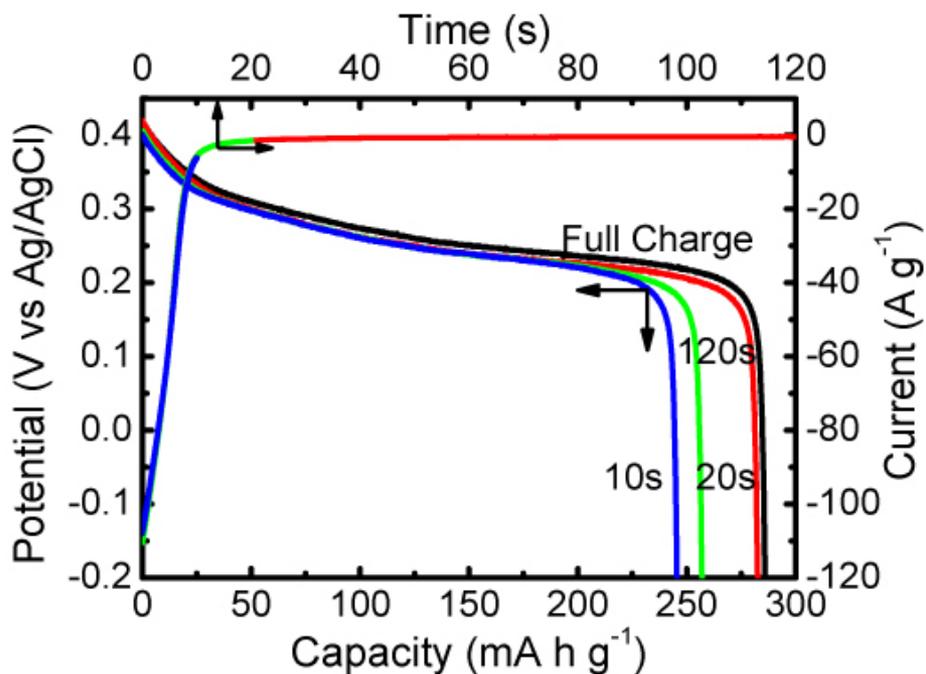
Lithium-ion Cathode

Significant capacity retention at 371C
(complete discharge in ~10 sec.)



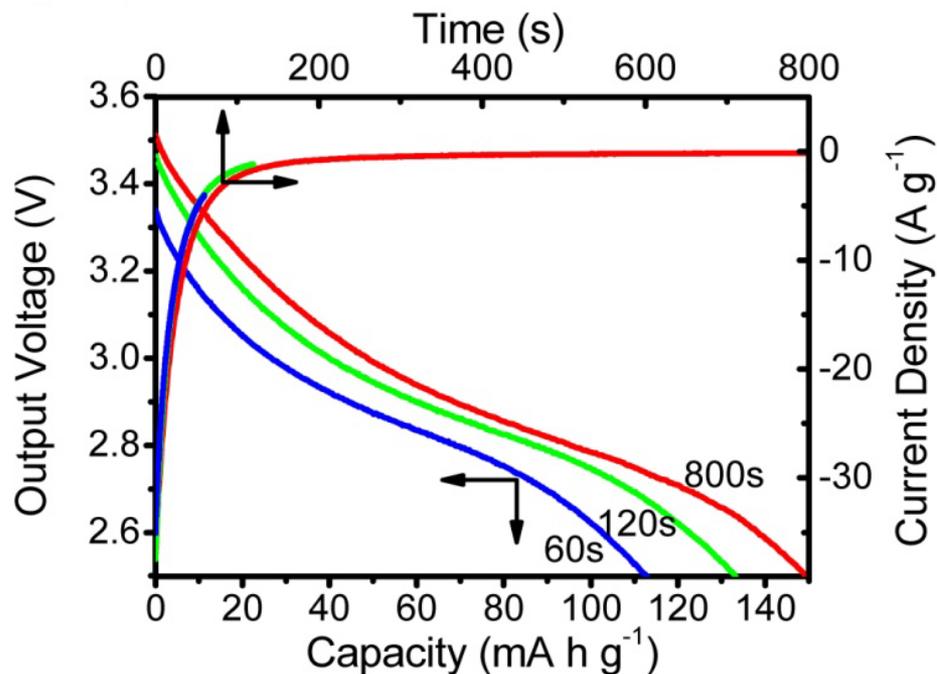
Ultrafast Charging

**NiOOH
cathode**



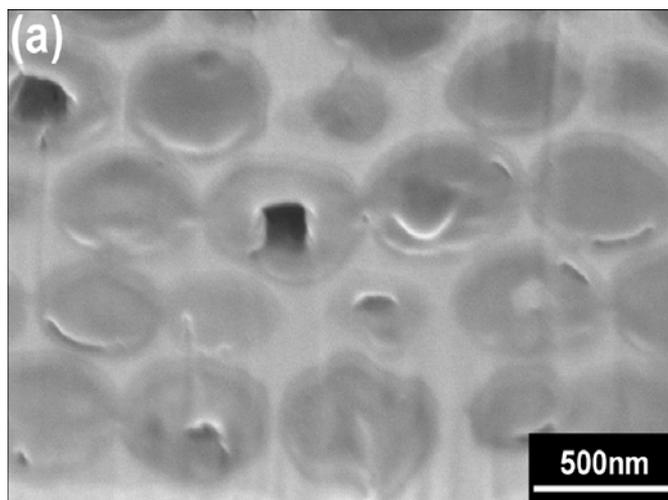
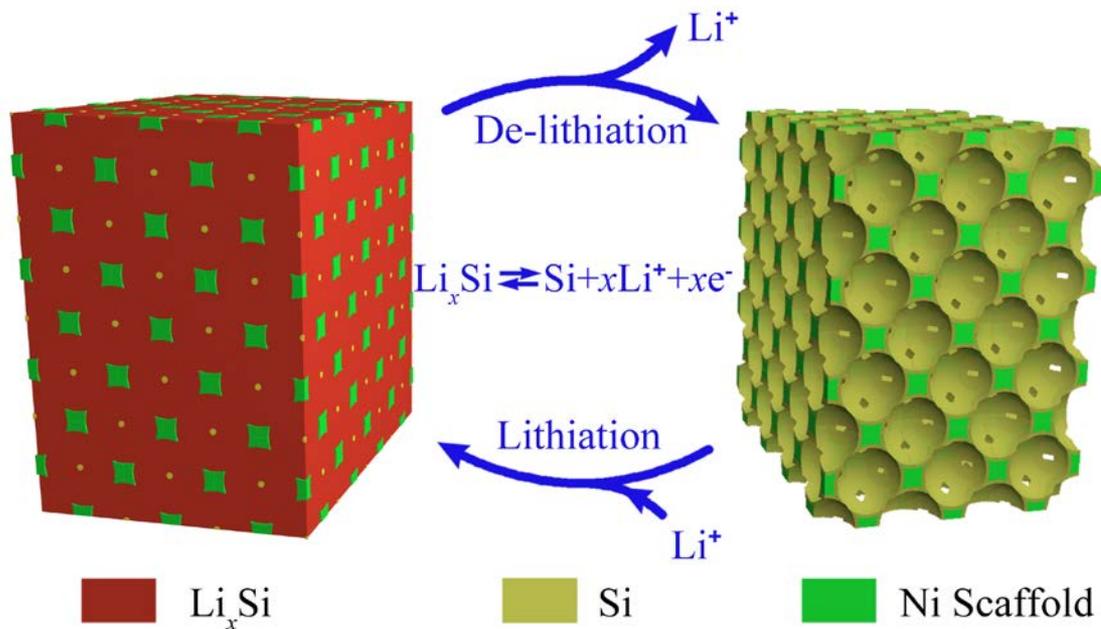
**Potentiostatic charging (0.4V vs Ag/AgCl)
3C discharge
Nearly complete charge after 120 sec.**

**Nanostructured Lithiated
MnO₂/Graphite Cell**

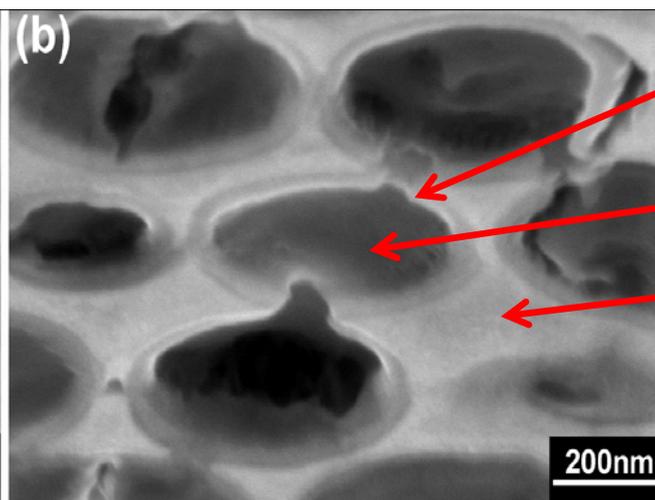


**Potentiostatic charging (3.5V)
3C discharge
Nearly complete charge after 120 sec.**

New Materials: Bicontinuous Silicon Anodes

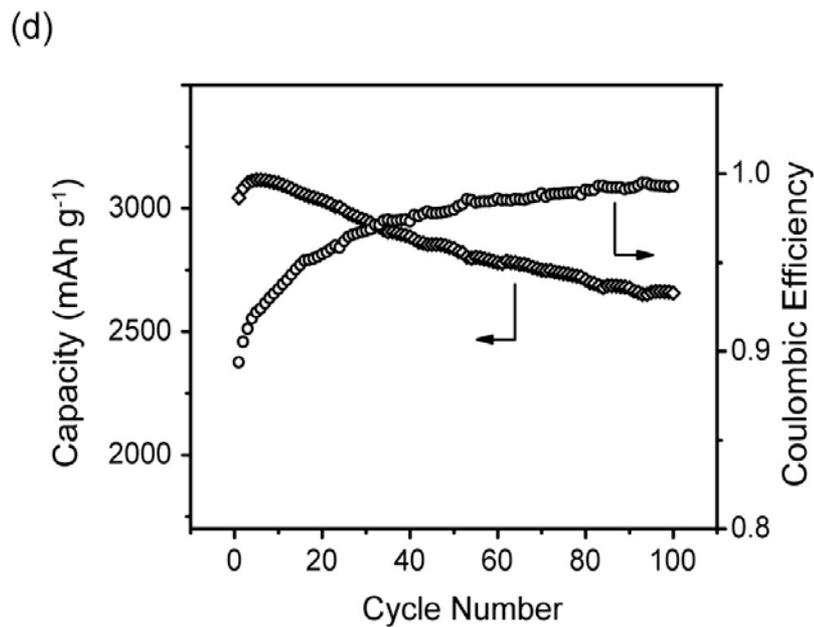
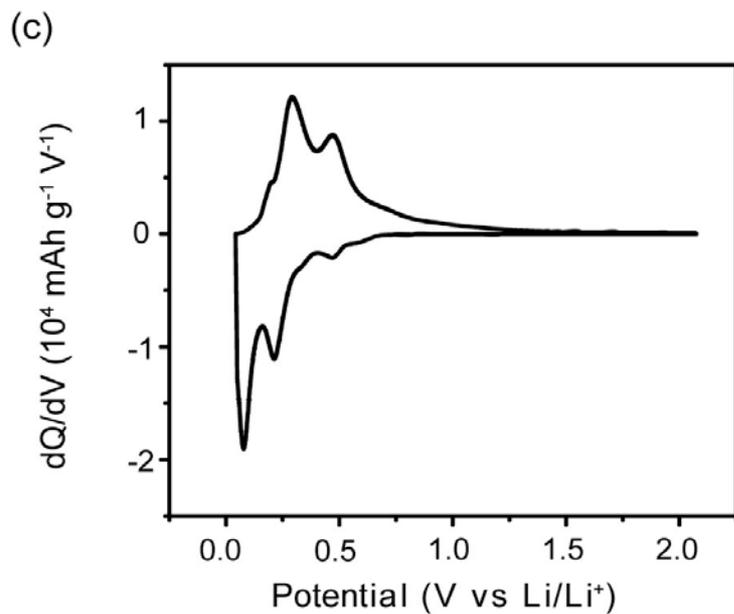
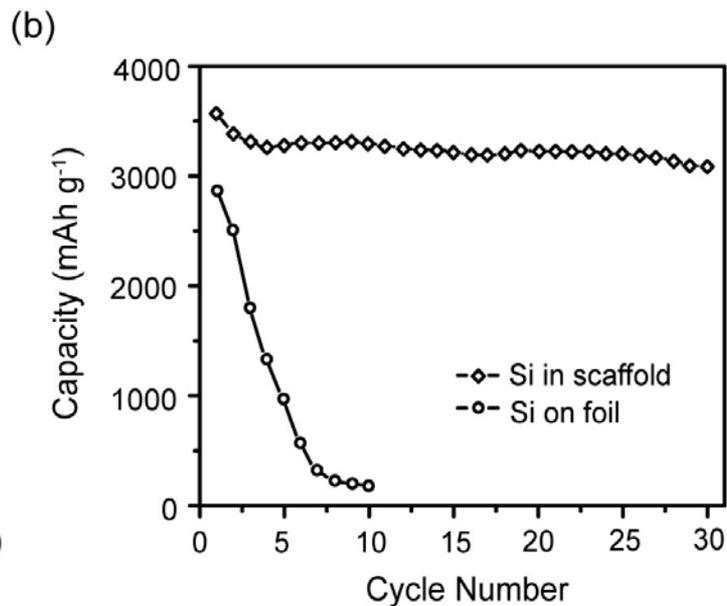
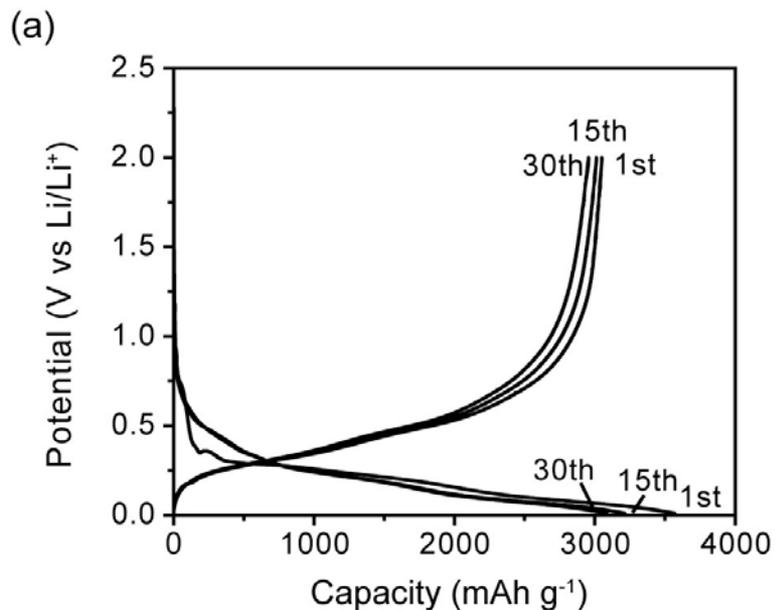


lithiated



de-lithiated

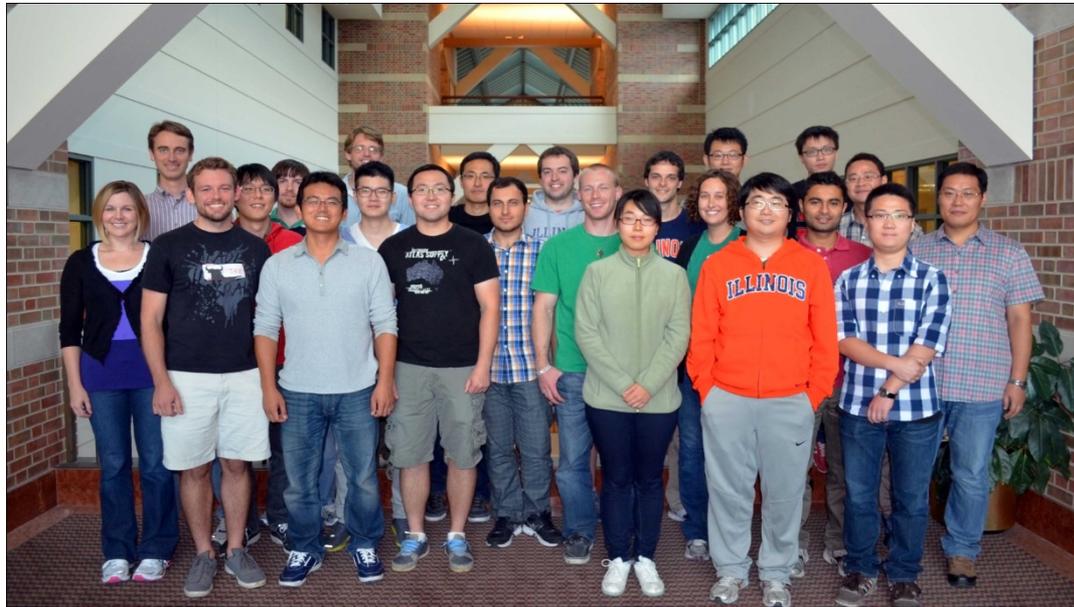
Bicontinuous Silicon Anodes



Concluding Thoughts

- **Nanotechnology can have “macroscopic” impact**
 - Providing new properties important
 - Important to consider what will provide high value added
 - Minimize necessity of top-down processing (\$\$)
- **Ask, what are the critical needs of industry?**
 - Mechanical
 - Thermal
 - Energy storage/harvesting
 - Optical
- **Long-term goal: make nanotechnology “invisible”**
 - Boring
 - Commonplace
 - Normal
- **Think beyond electronic materials and medicine**

Acknowledgements



Research Group and Collaborators

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Runyu Zhang
Qiye Zheng

Dr. Jinwoo Kim
Dr. Sung Kon Kim
Dr. Hyunh-Jun Koo
Dr. Jinyun Liu
Dr. Kris Waynant

Collaborators

Prof. David Cahill
Prof. Jim Coleman
Prof. Shanhui Fan
(Stanford)
Prof. William King
Prof. Jennifer Lewis
(Harvard)
Prof. Xiuling Li
Prof. Jeff Moore
Prof. John Rogers
Prof. Nancy Sottos
Dr. Richard Vaia
(AFRL)

Postdocs

Dr. Chaeryong Cho
Dr. Jiung Cho

Prof. Scott White

